

## UNIVERSIDADE FEDERAL DE SANTA CATARINA CENTRO DE CIÊNCIAS DA SAÚDE PROGRAMA DE PÓS-GRADUAÇÃO EM FARMÁCIA

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## ESTUDOS DE PERMEAÇÃO CUTÂNEA COM A TESTOSTERONA: UMA ATUALIZAÇÃO DAS CONDIÇÕES EXPERIMENTAIS EM DIFERENTES MODELOS DE PELE E INVESTIGAÇÃO DO EFEITO DE TERPENOS COMO PROMOTORES DE ABSORÇÃO

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O presente trabalho em nível de mestrado foi avaliado e aprovado por banca examinadora composta pelos seguintes membros:

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Certificamos que esta é a **versão original e final** do trabalho de conclusão que foi julgado adequado para obtenção do título de mestre em farmácia.

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Este trabalho é dedicado aos meus colegas e aos familiares queridos.

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"Existem muitas hipóteses em ciência que estão erradas. Isso é perfeitamente aceitável, elas são a abertura para achar as que estão certas". (Carl Sagan)

#### RESUMO

A primeira etapa deste estudo envolveu uma análise teórica do impacto de diferentes condições experimentais na taxa de permeação percutânea da testosterona (TST) para ensaios realizados em câmaras de difusão do tipo Franz. Os principais achados foram: (1) a presença de agentes solubilizantes (etanol/propilenoglicol) na fase doadora reduz a quantidade de TST permeada; (2) doses infinitas de TST na fase doadora podem reduzir a taxa de permeação caso eventos de supersaturação levem à precipitação de fármaco; (3) usar preferencialmente pele dermatomizada já que a derme funciona como uma barreira que dificulta a difusão do fármaco para o fluido receptor; (4) priorizar pele suína/humana à pele de rato e peles artificiais (superestimam a quantidade de TST absorvida). Considerando a dificuldade que a TST tem para permear a pele e o uso de promotores químicos de absorção irritantes em formulações do mercado, a segunda etapa deste trabalho envolveu ensaios de permeação cutânea ex vivo da TST em presença de três terpenos. O tratamento com carvacrol, α-bisabolol e mentol resultou em um raio de reforço da permeação de 7,61, 5,10 e 2,52, respectivamente. O carvacrol melhorou não só a permeação assim como a retenção da TST na pele. Análises espectroscópicas e térmicas revelaram que este agente causa maior desordem nos domínios lipídicos. Além da alta eficácia como promotor de absorção, este agente pode agir de forma sinérgica com a TST no hipogonadismo, no entanto, mais estudos ainda são necessários para comprovar as vantagens desta associação. Por fim, a terceira etapa envolveu uma análise comparativa de ensaios de transporte de fármacos realizados em membranas biológicas e não biológicas. Membranas sintéticas e modelos de pele reconstruída possuem como principais limitações: (1) maior taxa de permeação a fármacos que pele humana; (2) ausência de transporte folicular; (3) ausência de metabolismo no caso de membranas sintéticas; (4) diferenças em termos da organização lipídica; (5) são mais afetados por constituintes da formulação tais como promotores químicos. Assim, pele animal e humana ainda deve ser priorizada para ensaios de transporte de fármacos até que novos avanços com modelos de pele reconstruída sejam observados.

**Palavras-chaves:** permeação cutânea; promotores de permeação; testosterona; célula de Franz.

### ABSTRACT

The first step of this study involved a theoretical analysis of the impact of different experimental conditions on the percutaneous permeation rate of testosterone (TST) for assays performed in Franz-type diffusion chambers. The main findings were: (1) presence of solubilizing agents (ethanol/propylene glycol) in the donor chamber reduces the amount of TST permeated; (2) infinite doses of TST in the donor chamber can reduce the permeation rate if supersaturation events lead to drug precipitation; (3) dermatomed skin should be preferably used since the dermis works as a barrier that makes it difficult for the drug to diffuse into the recipient fluid; (4) prioritizing porcine/human skin over mouse and artificial skin (they overestimate the amount of TST absorbed). Considering the difficulty that TST has to permeate the skin and the use of irritating absorption chemical enhancers in market formulations, the second step of this work involved ex vivo skin permeation assays of TST in the presence of three terpenes. Treatment with carvacrol,  $\alpha$ -bisabolol and menthol resulted in a permeation enhancement ration of 7.61, 5.10 and 2.52, respectively. Carvacrol improved not only permeation but also the retention of TST in the skin. Spectroscopic and thermal analyzes revealed that this agent causes greater disorder in the lipid domains. In addition to its high efficacy as an absorption chemical enhancer, this agent can act synergistically with TST in hypogonadism; however, more studies are still needed to prove the advantages of this association. Finally, the third step involved a comparative analysis of drug transport assays performed on biological and non-biological membranes. The main limitations observed for synthetic membranes and reconstructed skin models were (1) higher drug permeation rate than the human skin; (2) absence of follicular transport; (3) lack of metabolism in the case of synthetic membranes; (4) differences in terms of lipid organization; (5) are more affected by formulation constituents such as chemical absorption enhancers. Thus, animal and human skin should still be prioritized for drug transport assays until further advances with reconstructed skin models may be observed.

**Keywords:** skin permeation; permeation chemical enhancers; testosterone; Franz-type diffusion cell.

### LISTA DE FIGURAS

### Capítulo I - Revisão Bibliográfica

Figura 1. Ritmo circadiano da TST no sangue de homens	20
Figura 2. Estrutura química da testosterona.	31
Figura 3. Estrutura da pele humana	33
Figura 4. Processos que ocorrem durante a liberação transdérmica de fármacos	36
Figura 5. Célula de difusão de Franz	38
Figura 6. Mecanismos de ação dos promotores de permeação	41
<b>Figura 7.</b> Estruturas químicas dos terpenos α-bisabolol, carvacrol e mentol	44

### **Capítulo III**

Figure 1. Chemical structures of terpenes and testosterone
Figure 2. Cutaneous permeation (A) and retention (B) of testosterone through the
human skin after 24 and 36 h (n=6) 134
<b>Figure 3.</b> Skin permeation profile of testosterone in the presence of carvacrol ( $\bullet$ ), $\alpha$ -
bisabolol ( $\blacktriangle$ ) and menthol ( $\triangledown$ ). TST in propylene glycol was used as control ( $\blacksquare$ ) 135
Figure 4. Retention of testosterone (TST) in the epidermis and dermis layers after
treatment with terpenes. Control: 1% TST (w/w) in propylene glycol (n=6) 136
Figure 5. FT-IR spectra of untreated stratum corneum (i) and tissue treated only with

 **Figure 6.** Thermograms of untreated stratum corneum (i), and stratum corneum treated with TST in propylene glycol (ii); TST in propylene glycol and carvacrol (iii); TST in propylene glycol and  $\alpha$ -bisabolol (iv) and TST in propylene glycol and menthol (v). 140

### Capítulo IV

Figure 1. Skin structure (left) and main layers of the epidermis (right)	153
Figure 2. Schematic representation of PAMPA model.	160
Figure 3. Reconstructed human skin models for drug transport studies	163
Figure 4. Main classes of lipids present in SkinEthic®, EpiDerm®, Epi	iSkin®

compared to native skin. Data were obtained from Ponec et al. (2002)...... 166

# LISTA DE QUADROS

Quadro 1. Sintomas do hipogonadismo.	23
Quadro 2. Propriedades físico-químicas e farmacocinéticas da testosterona	31
<b>Quadro 3.</b> Propriedades físico-químicas dos terpenos $\alpha$ -bisabolol, carvacrol e	
mentol	. 44

## LISTA DE TABELAS

# Capítulo II

Table 1.	Testosterone s	solubility in dif	ferent solver	nts and expen	rimental con	ditions	90
Table 2.	Experimental	conditions use	d in skin per	meation stud	lies with test	osterone1	02

# Capítulo III

Table 1. Permeation parameters of testosterone (TST) alone (control s	solution) and in
the presence of terpenes	135
Table 2. Classification and properties of terpenes	142

# Capítulo IV

<b>Table 1.</b> Advantages and limitations of different skin models. 152
<b>Table 2.</b> Comparison between human and animal skin structure. 157
Table 3. Permeation parameters (permeation coefficient and lag time) for caffeine and
testosterone obtained from reconstructed skin models and reference tissues 167

## LISTA DE ABREVIATURAS E SIGLAS

ACTH	Adrenocorticotrofina
ARs	Androgen receptors
CAEs	Chemical absorption enhancers
CF	Caffeine
CRH	Hormônio liberador de corticotrofina
DHT	Dihydrotestosterone
DSC	Calorimetria exploratória diferencial
ECM	Extracellular matrix
EMA	European Medicines Agency
EPA	Environmental Protection Agency
FDA	Food and Drug Administration
FT-IR	Espectroscopia de infravermelho com transformada de Fourier
GnRH	Hormônio liberador de gonadotrofina
GRAS	Generally recognized as safe
HPLC	High-performance liquid chromatography
ICSH	Hormônio estimulante de célula intersticial
IPCS	International Programme on Chemical Safety
Кр	Permeation coefficient
LH	Hormônio luteinizante
OECD	Organisation for Economic Co-operation and Development
PAMPA	Parallel Artificial Membrane Permeability Assay
PBS	Phosphate-buffered saline
PG	Propylene glycol
SC	Stratum corneum
SPE	Solid-phase extraction
TEER	Transepidermal electrical resistance
TEWL	Transepidermal water loss
TST	Testosterona
TWF	Transepidermal water flux
WHO	World Health Organization

# SUMÁRIO

1. INTR	RODUÇÃO	. 14
1.1. C	DBJETIVOS	. 18
1.1.1.	OBJETIVO GERAL	. 18
1.1.2.	OBJETIVOS ESPECÍFICOS	. 18
2. CAPÍ	TULO I - REVISÃO BIBLIOGRÁFICA	. 19
2.1. P	PROPRIEDADES FISIOLÓGICAS DA TESTOSTERONA	. 19
2.2. H	IIPOGONADISMO	. 22
2.3. F	ORMULAÇÕES PARA TRATAMENTO	. 25
2.4. P	PREPARAÇÕES TRANSDÉRMICAS	. 27
2.5. C INCORI	CARACTERÍSTICAS DESEJÁVEIS PARA FÁRMACOS PORADOS EM SISTEMAS TRANSDÉRMICOS	. 30
2.6. A QUÍMIO	ASPECTOS BIOFARMACÊUTICOS E CARACTERÍSTICAS FÍSICO- CAS DA TESTOSTERONA	. 31
2.7. A	SPECTOS ANATÔMICOS E FISIOLÓGICOS DA PELE	. 32
2.8. T	RANSPORTE DE MOLÉCULAS ATRAVÉS DA PELE	. 35
2.9. N FÁRMA	IODELOS PARA AVALIAÇÃO DA PERMEAÇÃO CUTÂNEA DE ACOS	. 36
2.10. P	PROMOTORES DE ABSORÇÃO	. 40
2.10.1	. Alfa-bisabolol	. 44
2.10.2	2. Carvacrol	. 45
2.10.3	3. Mentol	. 46
REFERÊ	NCIAS	. 47
3. CAPÍ CUTÂNE BIOFAPI	TULO II – COMO PLANEJAR ENSAIOS DE PERMEAÇÃO A PARA FÁRMACOS QUE APRESENTAM LIMITAÇÕES MACÊUTICAS COMO A TESTOSTEBONA?	84
3 1 A	ARTIGO CIENTÍFICO I	. 0 <del>1</del> 85
J.1. A		. 05
4. CAPÍ	TULO III – EFEITO DOS TERPENOS COMO PROMOTORES DE ÃO PARA A TESTOSTERONA	175
	AUTARA A TESTOSTERUNA	123
<b>T.I.</b> 1		120

5. CAPÍTULO IV – UMA ANÁLISE COMPARATIVA DE MODELOS DE PELES BIOLÓGICAS E SINTÉTICAS PARA ESTUDOS DE TRANSPORTE DE				
FÁ	RMACOS	148		
5	5.1. ARTIGO CIENTÍFICO III	149		
6.	DISCUSSÃO GERAL	180		
7.	CONSIDERAÇÕES FINAIS	187		

#### 1. INTRODUÇÃO

Estudos *in vivo* em humanos representam o padrão ouro para a análise da permeação cutânea de fármacos (LAI; MAIBACH, 2009), porém, são difíceis de serem realizados e interpretados (ABD *et al.*, 2016). Por este motivo, modelos *in vitro/ex vivo* ainda têm sido priorizados, pelo menos em uma etapa inicial de triagem de diferentes moléculas ou desenvolvimento de novas formulações. Idealmente, peles de tecido humano devem ser selecionadas para estes experimentos, as quais são frequentemente obtidas de cirurgias plásticas ou cadáveres (ESPOSTO BIONDO *et al.*, 2021).

Algumas técnicas utilizadas para a análise da penetração/permeação de fármacos incluem o *tape stripping*, a microdiálise e os testes de permeação *ex vivo* em células de difusão (RANEY *et al.*, 2015). O modelo bicompartimental das células de difusão de Franz ainda é o mais utilizado já que é capaz de fornecer vários tipos de respostas (taxa de permeação/retenção cutânea) e permite extrair informações a respeito do tipo de transporte do fármaco através da pele. Estas células podem ser encontradas nas configurações vertical ou horizontal, estática ou dinâmica (fluxo). Célula vertical e estática são as mais comuns (AULTON; TAYLOR, 2016).

Diferenças na taxa de absorção/permeação podem ser observadas dependendo da região do corpo selecionada (ABD et al., 2016; ROUGIER; LOTTE; MAIBACH, 1987), composição do tecido (quantidade de camadas de células, composição lipídica e proteica), quantidade de apêndices por unidade de área (pêlos e glândulas), hidratação, espessura (MARRAKCHI; MAIBACH, 2007; SANDBY-MØLLER; POULSEN; WULF, 2003) e presença de cicatrizes (MILLS; MAGNUSSON; CROSS, 2006). Assim, a extrapolação dos resultados de estudos deve ser realizada com cautela (WALTERS; ROBERTS, 2002).

Há um número expressivo de estudos de permeação cutânea com a testosterona (TST) na literatura. A TST é o hormônio é responsável pelo desenvolvimento dos órgãos sexuais masculinos durante a gestação. Na adolescência, é responsável pelo desenvolvimento das características físicas secundárias masculinas e, na vida adulta, realiza a manutenção destas (KIM; LEE; KIM, 2000; LEE; CHANG, 2003; NICOLAZZO et al., 2005). Ainda, é responsável pela formação da matriz óssea (BAIN, 2008) e tem ação anabolizante nos músculos (CIGARRÁN et al., 2013).

A produção deste hormônio varia durante o dia, com uma maior concentração entre 6 e 8 h da manhã. Ao longo do dia, observa-se uma queda progressiva dos níveis hormonais seguindo-se um padrão circadiano (BHASIN et al., 2010). No decorrer da vida e principalmente após os 40 anos de idade, a produção da TST começa a diminuir, com uma taxa anual de declínio de 0,4 a 2% (DOHLE et al., 2012), que pode levar ao hipogonadismo. Esta patologia ocorre quando se observam sintomas clínicos sugestivos da doença (ex.: impotência sexual, diminuição da libido, cansaço, depressão do humor, desmineralização óssea, perda muscular, deficiência imunológica e dificuldade no metabolismo de carboidratos), concentração plasmática de TST abaixo de 300 ng/dL e níveis de TST livre abaixo de 6,5 ng/dL (DOHLE *et al.*, 2012; MARTITS; COSTA, 2005).

Assim, várias formulações a base de TST têm sido desenvolvidas com o propósito de restaurar os níveis fisiológicos, incluindo testes em diferentes rotas de administração (MILLS; MAGNUSSON; CROSS, 2006). Embora a rota oral seja priorizada na maior parte dos casos devido a sua conveniência e alta adesão terapêutica que pode ser alcançada, a TST sofre alta metabolização hepática após administração oral (KAUFMAN, 2004). Modificações estruturais tem sido propostas para retardar a sua degradação hepática, porém, derivados sintéticos propostos até o momento tem mostrado uma série de efeitos colaterais indesejáveis (HOTLINE, 1998; KAUFMAN, 2004). Formulações injetáveis, por sua vez, provocam desconforto e dor na aplicação. Além disto, resultam em um rápido aumento da concentração plasmática a níveis superiores ao fisiológico, o que pode resultar em transtornos de humor (ARVER et al., 1997; JOCKENHÖVEL, 2004; LEICHTNAM et al., 2006; NIESCHLAG; BEHRE, 1998). Implantes subcutâneos foram desenvolvidos para evitar esta flutuação da concentração plasmática do hormônio com redução dos efeitos adversos (KAUFMAN, 2004), mas a aplicação requer pessoal treinado. Em vista deste cenário e as vantagens obtidas com o implante subcutâneo, sistemas transdérmicos começaram a ser desenvolvidos. Esta rota de administração não é invasiva, não requer profissional especializado para a aplicação, é indolor, e permite controlar de forma efetiva a liberação do fármaco. Como consequência, há uma maior adesão do paciente ao tratamento (BAERT et al., 2012).

No delineamento de formulações transdérmicas, deve-se analisar as propriedades físico-químicas do fármaco tais como solubilidade, pk<sub>a</sub> e log P a fim de definir a

composição da formulação. O fármaco deve ser capaz de ultrapassar regiões lipídicas do estrato córneo, a principal barreira da pele, bem como regiões hidrofílicas da epiderme até o alcance da circulação sanguínea. Candidatos ideais para esta rota de administração devem apresentar valores de log P intermediário (entre 1 e 4), reduzida massa molar (menor que 500 g/mol) e ponto de fusão menor que 250°C. É desejável também que o fármaco apresente um tempo de meia-vida intermediário que facilita a manutenção das doses do fármaco (AULTON; TAYLOR, 2016).

A TST é um hormônio relativamente lipofilico, com um coeficiente de partição óleo/água (log P) entre 3,32 e 3,42 e massa molar de 288,42 g/mol (HADGRAFT; LANE, 2015; KIM; LEE; KIM, 2000; MISRA et al., 1996). Sua solubilidade em água é de 0,039 mg/mL a 37°C (OKIMOTO; RAJEWSKI; STELLA, 1998). Ainda que a TST apresente baixa massa molar e log P intermediário, observa-se baixa permeação cutânea devido a alta afinidade pela pele e presença de receptores específicos no tecido cutâneo que metabolizam esse hormônio. Por outro lado, baixas concentrações plasmáticas são requeridas para um efeito terapêutico, o que torna a rota transdérmica atrativa. A TST é ainda capaz de chegar até a circulação sistêmica ou se depositar nas células de gordura, formando uma espécie de reservatório do fármaco, gerando uma liberação lenta e gradual na circulação sanguínea (ČEPONIS et al., 2017; WANG et al., 2000).

Considerando a natureza lipofílica e os vários estudos de absorção com essa molécula, esta dissertação apresenta dois estudos de revisão que trazem análises das condições experimentais e o uso de diferentes membranas de pele em ensaios de permeação cutânea (capítulos II e IV). Além disso, tendo em vista as propriedades de barreira do estrato córneo e a necessidade de inclusão de promotores de permeação/retenção da TST com o uso de terpenos foi avaliada (capítulo III). Estes agentes promotores podem agir rompendo a estrutura lipídica intercelular entre os corneócitos, fluidizando o estrato córneo; interagir com as proteínas do meio intercelular, melhorando a partição da molécula; agir nas conexões dos desmossomos entre os corneócitos ou alterar a atividade metabólica do tecido. Todos estes mecanismos podem modificar o arranjo estrutural do tecido e, consequentemente, diminuir as propriedades de barreira do estrato córneo (SILVA *et al.*, 2010; HERMAN; HERMAN, 2015; WILLIAMS; BARRY, 1991), aumentando a difusão da molécula-alvo. O mecanismo preferencial

dependerá das propriedades físico-químicas do fármaco e do promotor (GHAFOURIAN *et al.*, 2004).

Os terpenos foram selecionados como agentes de permeação dada a origem natural que reduz chances de irritação ou toxicidade. Os terpenos são reconhecidamente seguros pela FDA (GRAS, *generally recognized as safe*), fato extremamente relevante já que muitos destes agentes podem penetrar a pele e alcançar a corrente sanguínea (SAPRA; JAIN; TIWARY, 2008; SONGKRO; RADES; BECKET, 2009a). Além de serem rapidamente metabolizados e excretados (HERMAN; HERMAN, 2015), terpenos apresentam atividades farmacológicas específicas (WILLIAMS; BARRY, 2012), o que traz vantagens na sua utilização. O principal mecanismo de ação dos terpenos é a alteração da fluidez do estrato córneo através da interação com os lipídeos intercelulares (HERMAN; HERMAN, 2015; JIANG et al., 2017). Os terpenos selecionados para este estudo foram o alfa-bisabolol, o carvacrol e o mentol, os quais foram solubilizados em propilenoglicol, um cossolvente que age sinergicamente com os terpenos favorecendo a permeação de moléculas (ALAN ANDERSEN, 1999; BARRY; WILLIAMS, 1989; CHEN *et al.*, 2016; SONGKRO; RADES; BECKET, 2009a).

#### **1.1. OBJETIVOS**

### 1.1.1. OBJETIVO GERAL

Analisar como diferentes variáveis do ensaio de permeação afetam a taxa de transporte da TST através da pele humana e animal bem como em membranas não biológicas, além da avaliação do efeito de terpenos como promotores químicos de absorção.

### 1.1.2. OBJETIVOS ESPECÍFICOS

- ✓ Identificar as principais variáveis do ensaio de permeação cutânea que afetam a taxa de transporte de fármacos com limitações biofarmacêuticas (baixa solubilidade ou permeabilidade) a partir da análise de estudos com a TST;
- Realizar ensaios de permeação da TST em diferentes tempos utilizando células de difusão de Franz e pele humana;
- ✓ Selecionar, a partir da literatura, terpenos com reconhecido efeito promotor na absorção de compostos lipofílicos;
- Analisar o impacto dos terpenos nos parâmetros fluxo e coeficiente de permeação da TST durante seu transporte na pele;
- Monitorar o impacto do tratamento com os terpenos na distribuição da TST na derme e epiderme (estudos de retenção cutânea);
- ✓ Investigar as interações entre os terpenos e estruturas da pele por espectroscopia de infravermelho com transformada de Fourier (FT-IR) e calorimetria exploratória diferencial (DSC);
- ✓ A partir do fluxo de permeação encontrado, realizar o cálculo teórico da quantidade de TST que alcança a circulação sistêmica.
- ✓ Comparar o desempenho de modelos biológicos e não biológicos de pele em ensaios de transporte de fármacos selecionando a cafeína e TST como modelo de fármaco hidrofílico e lipofílico, respectivamente.

### 2. CAPÍTULO I - REVISÃO BIBLIOGRÁFICA

### 2.1. PROPRIEDADES FISIOLÓGICAS DA TESTOSTERONA

A testosterona (TST) é o principal hormônio andrógeno presente na circulação sanguínea de homens. Atua nas células-alvo através de seu metabólito ativo, a dihidrotestosterona, gerado pela enzima  $5\alpha$ -redutase (RANG et al., 2016). É produzida a partir do colesterol nas suprarrenais e principalmente pelas células de Leydig dos testículos (NIESCHLAG; BEHRE, 1998). As mais de 500 milhões de células de Leydig são responsáveis pela produção de 6 a 7 mg de TST a cada dia, o equivalente a 95% do total de TST produzido (COFFEY, 1988). Na literatura, há relatos de que a produção não se limita a glândulas, podendo também ser produzida em pequenas quantidades por células cerebrais (BAULIEU, 1997). Embora esta produção seja insignificante perto da quantidade total disponível na circulação sanguínea, acredita-se que seja responsável por algum efeito fisiológico local (NIESCHLAG; BEHRE, 1998).

Nas suprarrenais, o estímulo é derivado dos hormônios CRH (hormônio liberador de corticotrofina) e a ACTH (adrenocorticotrofina) originários do eixo hipotálamo-hipófise-adrenal. Já a produção da TST nas células de Leydig é estimulada pelo ICSH (hormônio estimulante de célula intersticial), também chamado de LH (hormônio luteinizante). A liberação do LH ocorre via adeno-hipófise que, por sua vez, é estimulada pelo GnRH (hormônio liberador de gonadotrofina) produzido no hipotálamo (DOHLE *et al.*, 2012; RANG *et al.*, 2016). Esses hormônios são capazes de regular a quantidade de TST disponível no sangue, bem como a proliferação e diferenciação das células de Leydig (CHEMES, 1996). Na corrente sanguínea, quase toda a TST está ligada a proteínas plasmáticas, especialmente a globulina ligante de esteroides sexuais (DOHLE *et al.*, 2012; RANG *et al.*, 2016).

A produção do hormônio varia durante o dia, iniciando por volta da meia noite com uma maior concentração entre 6 e 8 h da manhã. A concentração cai progressivamente ao longo do dia em um padrão circadiano (BHASIN *et al.*, 2010). A concentração de TST varia durante o dia devido à liberação pulsátil de LH na corrente sanguínea (BRIDGES *et al.*, 1993). Na figura 1, é possível acompanhar a variação da concentração sanguínea da TST ao longo do dia em homens jovens e adultos.



Figura 1. Ritmo circadiano da TST no sangue de homens.

Fonte: traduzido de Bremner & Vitiello & Prinz (1983)

Sua concentração plasmática fisiológica deve estar entre 300 e 800 ng/dL (SNYDER, 2018). Para manter a concentração nos níveis adequados, é necessário haver um equilíbrio entre a produção, o metabolismo e a excreção desse hormônio (NIESCHLAG; BEHRE, 1998).

A TST é responsável pelo desenvolvimento dos órgãos sexuais masculinos durante a gestação. Na vida adulta, mais precisamente na adolescência, é responsável pelo desenvolvimento das características físicas secundárias masculinas, bem como manutenção destas características ao longo da vida (KIM; LEE; KIM, 2000; LEE; CHANG, 2003; NICOLAZZO *et al.*, 2005). Nos ossos, a TST é responsável por duas funções complementares. Após ser convertido a estradiol pela enzima aromatase, inibe a atividade dos osteoclastos, responsáveis pela remodelação óssea, e, portanto, inibe a reabsorção óssea. Também atua via enzima 5 $\alpha$ -redutase, que a transforma em dihidrotestosterona, estimulando osteoblastos - células ósseas responsáveis pela formação da matriz óssea (BAIN, 2008). Ainda nos ossos, possui ação estimulante da hematopoiese, aumentando o hematócrito. A TST atua se ligando em receptores citoplasmáticos de células renais, sendo carreada para o núcleo onde incentiva a ação da RNA polimerase resultando na produção de eritropoietina e sua consequente secreção

(SHAHANI *et al.*, 2009). A eritropoietina é o hormônio regulador da hematopoiese, ou seja, a formação de glóbulos vermelhos. É devido a esta ação que o hematócrito deve ser acompanhado em pacientes que fazem uso da TST (BAIN, 2008).

Nos músculos, a testosterona atua como um hormônio anabolizante que promove hipertrofía (CIGARRÁN *et al.*, 2013). Auxilia na absorção de nitrogênio, que contribui para a formação e manutenção dos tecidos musculares (BAIN, 2008). Os músculos são originados a partir de células multinucleadas presentes em uma matriz de colágeno e nervos motores. Estas células satélites pluripotentes são responsáveis por se diferenciarem em mioblastos e, consequentemente, nas fibras musculares. A TST atua nestas múltiplas etapas do desenvolvimento das fibras musculares (HERBST; BHASIN, 2004). A falta desse hormônio é responsável pela diminuição da massa e força muscular. Além disso, contribui para o aumento da gordura corporal, aumentando o risco de doenças coronarianas e diabetes (BAIN, 2008). Há evidências de que o sedentarismo associado à obesidade esteja relacionado com a baixa concentração de TST em homens, indicando que o hipogonadismo esteja crescente na sociedade (BROWN, 2008).

Existem ainda algumas evidências da ação hormonal da TST no cérebro. Uma delas é a presença de receptores de di-hidrotestosterona e estradiol no cérebro, os quais são metabolitos da  $5\alpha$ - redutase e aromatase, respectivamente. Também há evidências da promoção de crescimento de células neuronais devido a esse hormônio, além de ser responsável pelo aumento da perfusão cerebral em regiões ligadas ao planejamento estratégico, funções motoras, cognição, emoções, concentração e memória (AZAD *et al.*, 2003; BAIN, 2008). Com relação ao humor e à depressão, estudos demonstram uma relação direta da baixa concentração de TST em homens de idade mais avançada com a depressão, embora estudos adicionais ainda se mostram necessários (BAIN, 2008).

Alguns estudos também relatam que a alta incidência de reumatismo em homens de idade mais avançada é decorrente da queda da biodisponibilidade da TST neste grupo. Isto é justificado pelo fato de os hormônios andrógenos serem capazes de abafar as respostas imunológicas hormonais e celulares atuando como uma espécie de antiinflamatório natural (BAIN, 2008; CUTOLO, 2000).

Acredita-se também que a TST é um dos hormônios responsável pelas características distintas entre homens e mulheres com relação aos comportamentos e

habilidades já que pode agir no Sistema Nervoso Central (HUTCHISON, 1991). No período neonatal, a TST contribui para o desenvolvimento cerebral das células indiferenciadas, atuando na região pré-orbital e no sistema límbico (amigdala) (BETTINI *et al.*, 1992; BRAIN; HAUG, 1992). Uma vez que estas regiões são ativadas, observam-se alterações comportamentais, particularmente durante a puberdade, a qual é caracterizada por um aumento da concentração circulante deste hormônio (BECKER; BREEDLOVE; CREWS, 1992). Embora haja restrições éticas na manipulação dos níveis hormonais humanos para estudos comportamentais e da função cerebral, sabe-se que esta influência é muito maior em animais que em humanos (NIESCHLAG; BEHRE, 1998).

A forma cristalina da TST foi primeiramente isolada de testículos de búfalos em 1935. E, neste mesmo ano, já foi sintetizada quimicamente. A partir daí, começou a ser utilizada clinicamente no tratamento do hipogonadismo sem qualquer comprovação científica (BUTENANDT; HANISCH, 1935; DAVID *et al.*, 1935; NIESCHLAG; BEHRE, 1998). Atualmente, a utilização de formulações a base de TST para o tratamento do hipogonadismo reúne evidências científicas suficientes, com uma diversidade de protocolos terapêuticos.

### 2.2. HIPOGONADISMO

O hipogonadismo é caracterizado pela deficiência na produção da TST, podendo ser resultante da produção inadequada deste hormônio pelos testículos ou até mesmo uma deficiência na produção dos precursores hormonais no hipotálamo ou pituitária (DOHLE et al., 2012; KIM; LEE; KIM, 2000).

O *hipogonadismo primário* representa a maior parte dos casos e é resultante da baixa produção testicular de TST com consequência na diminuição da espermatogênese e infertilidade. Pode ser resultante da Síndrome de Klinefelter ou tumores testiculares, porém, outras causas também têm sido atribuídas (má formação dos testículos, orquite, doenças virais como caxumba, medicações, toxinas, varicocele, entre outras). O *hipogonadismo secundário*, por sua vez, é derivado de falhas no sistema nervoso central, especificamente no hipotálamo ou na pituitária, sendo uma das principais causas de desenvolvimento tardio na puberdade. O *hipogonadismo adulto*, decorrente da disfunção do eixo hipotalâmico-hipófise, apresenta sintomas das duas categorias

anteriormente mencionadas. Por fim, há o *hipogonadismo por falhas do receptor de andrógeno*, que tem uma frequência mais rara (DOHLE et al., 2012).

Ao longo da vida e principalmente após os 40 anos de idade, a produção da TST começa a diminuir, com uma taxa anual de declínio de 0,4 a 2%. Nos homens de meia idade entre 40 e 79 anos, a incidência do hipogonadismo em análises bioquímicas varia entre 2,1 e 12,8% enquanto que, a incidência de sintomas, varia entre 2,1 a 5,7%. A obesidade, a presença de doenças crônicas e a baixa qualidade de vida representam fatores de risco para o desenvolvimento da doença (DOHLE et al., 2012; MOORADIAN; MORLEY; KORENMAN, 1987).

Os sintomas que podem acometer o homem no hipogonadismo adulto estão descritos no Quadro 1. Dentre estes sintomas, destaca-se a impotência sexual, diminuição da libido, cansaço, depressão do humor, desmineralização óssea, perda muscular, deficiência imunológica e dificuldade no metabolismo de carboidratos (BIZZARRO et al., 1987; DOHLE et al., 2012; LU et al., 2013; MOORADIAN; MORLEY; KORENMAN, 1987).

Quadro 1. Sintomas do hipogonadismo.

Sintomas clínicos sugestivos da deficiência de testosterona			
Tamanho diminuído dos testículos			
Infertilidade masculina			
Diminuição dos pelos do corpo			
Ginecomastia			
Diminuição da massa corporal e força muscular			
Obesidade visceral			
Síndrome metabólica			
Diabetes tipo II e resistência à insulina			
Diminuição da densidade óssea			
Anemia branda			
Sintomas sexuais			
Redução da libido e atividade sexual			
Disfunções de ereção			
Poucos ou nenhumas ereções noturnas			
Sintomas cognitivos e psicológicos			
Fogachos			
Mudanças de humor, cansaço e irritação			
Distúrbios do sono			
Depressão			
Diminuição das funções cognitivas			

Fonte: Dohle et al. (2012).

Quando se observam sintomas clínicos sugestivos da doença, concentração plasmática abaixo de 300 ng/dL e níveis de TST livre abaixo de 6,5 ng/dL (DOHLE et al., 2012; MARTITS; COSTA, 2005), é recomendada a reposição hormonal. Para tal, podem-se utilizar formulações nas mais diversas formas de apresentação como a oral, intravenosa, subcutânea, sublingual, nasal, anal ou transdérmica, como será apresentado mais adiante (DOHLE *et al.*, 2012; KIM; LEE; KIM, 2000; RANG *et al.*, 2016).

Muitos pacientes com fraturas vertebrais e osteoporose também apresentam hipogonadismo. Receptores de androgênio são encontrados nos osteoblastos em quantidades significativas para promover a mineralização óssea. Estudos apontam que a reposição hormonal é necessária para a manutenção da densidade óssea (NATIONAL RESOURCE CENTER, 2018; ORWOLL; KLEIN, 1995).

Da mesma forma, o hipogonadismo é também associado a pacientes masculinos infectados com HIV, sendo que 75% deles apresentam os sintomas secundários. As causas aparentam ser multifatoriais e relacionadas a comorbidades. Além disso, alguns medicamentos antirretrovirais apresentam como efeitos adversos a desmineralização óssea. A reposição hormonal é indicada para esses pacientes com a intenção de recuperar a massa muscular e diminuir a osteoporose (BEDIMO *et al.*, 2012; WONG; LEVY; STEPHENSON, 2017).

Segundo um estudo de meta-análise de 2005, os efeitos adversos mais relevantes do uso terapêutico da TST incluem o aumento do hematócrito e eventos prostáticos. Associação com câncer de próstata, problemas cardiovasculares e apneia do sono são mencionados, contudo, ressalta-se a necessidade de comprovação científica (CALOF et al., 2005). Em 2007, uma meta-análise não comprovou o aumento da pressão sanguínea, hiperglicemia e dislipidemia devido ao uso de TST, no entanto, menciona maior ocorrência de riscos cardiovasculares (HADDAD et al., 2007). Na meta-análise de 2010, não foram encontradas diferenças significativas nos eventos de morte, infarto do miocárdio, arritmias cardíacas, diabetes, colesterol LDL, triglicerídeos, pressão arterial e eventos prostáticos (câncer e aumento do índice de PSA) entre usuários e não usuários de TST. Por outro lado, esta análise mais recente relatou queda acentuada nos níveis de colesterol HDL, o que poderia estar associado com doenças cardíacas. Além disto, é relatado o aumento significativo da quantidade de hemoglobina e, consequentemente,

do hematócrito, com aumento do risco de eritrocitose (BHASIN et al., 2003; COVIELLO et al., 2008; FERNÁNDEZ-BALSELLS et al., 2010), que se manifesta com sensação de fraqueza, falta de ar, dores de cabeça, sonolência, zumbido, entre outros (AMARU et al., 2019). Uma meta-análise foi realizada por BORST *et al.* (2014) para verificar a influência da rota de administração nos efeitos adversos. A administração oral mostrou maiores chances de causar problemas cardiovasculares comparativamente a administração transdérmica ou injetáveis.

De acordo com o Guia Prático da Sociedade de Endocrinologia Clínica, homens sob tratamento hormonal devem ser monitorados quanto ao bem-estar, libido, atividade sexual, concentrações séricas de TST, hemoglobina e hematócrito, PSA e exame de toque retal. No primeiro ano, sugere-se um monitoramento após 3, 6 e 12 meses do início da terapia. Nos anos subsequentes, é suficiente uma avaliação anual (DOHLE *et al.*, 2018). É aconselhável o acompanhamento a cada dois anos da densidade mineral óssea (NIESCHLAG, 2015). Como os efeitos cardíacos são questionáveis, a agência norte-americana *Food and Drug Administration* (FDA) menciona que as indústrias farmacêuticas não são obrigadas a incluir nas embalagens/bulas de produtos à base de TST informações relativas a este risco (FDA, 2014).

### 2.3. FORMULAÇÕES PARA TRATAMENTO

Desde o isolamento e síntese da TST, várias formulações têm sido desenvolvidas com o propósito de repor os níveis fisiológicos da TST em pacientes com hipogonadismo. Uma das principais limitações do tratamento, principalmente nos pacientes que necessitam de administração prolongada de TST, é a via de administração (MILLS; MAGNUSSON; CROSS, 2006).

As formulações orais de TST sofrem um efeito de primeira passagem expressivo, com um curto tempo de meia vida de aproximadamente 10 min. A alquilação da TST permite atrasar a degradação, porém, o derivado sintético é menos potente (KAUFMAN, 2004). Formulações orais de metiltestosterona e oxandrolona também aparecem como alternativas a TST para contornar a baixa biodisponibilidade, mas alta hepatotoxicidade é observada com a administração destes derivados (HOTLINE, 1998). Apesar destas modificações estruturais, muitas destes derivados ainda são altamente metabolizados, alcançando concentrações sistêmicas ainda insuficientes (KAUFMAN, 2004; MILLS; MAGNUSSON; CROSS, 2006).

A rota sublingual tem sido utilizada para a administração de  $17\alpha$ metiltestosterona associada a ciclodextrinas, pois sua forma isolada também é caracterizada por alta hepatotoxicidade (NIESCHLAG; BEHRE, 1998). Apesar do retardo na metabolização, o composto ainda foi caracterizado por curta meia-vida (1-6 h), o que requer várias doses diárias. Outra desvantagem é a necessidade de se melhorar a palatabilidade destes sistemas (KAUFMAN, 2004; NIESCHLAG; BEHRE, 1998).

Considerando estas limitações, iniciou-se o desenvolvimento de formulações injetáveis, as quais eram administradas a cada 10 ou 20 dias. Além do desconforto e dor, estas formulações proporcionam um rápido aumento da concentração plasmática a níveis superiores ao fisiológico (ARVER *et al.*, 1997; JOCKENHÖVEL, 2004; LEICHTNAM *et al.*, 2006a). Esta flutuação dos níveis hormonais pode causar transtornos de humor no paciente, além de afetar a libido, atividade sexual e vigor físico (NIESCHLAG; BEHRE, 1998). Devido aos picos de concentração apresentados e efeitos adversos relatados, outros tipos de veículos/rotas de administração continuaram a serem explorados nos anos 80 (KAUFMAN, 2004).

Implantes subcutâneos foram desenvolvidos com o objetivo de manter as concentrações plasmáticas hormonais mais estáveis. Estes dispositivos são introduzidos sob a pele, logo abaixo da epiderme, liberando o hormônio por entre 4 e 5 meses. Ao contrário das injeções intramusculares, esta nova forma farmacêutica evitava flutuações plasmáticas, reduzindo ou até evitando efeitos adversos (KAUFMAN, 2004). Por outro lado, ainda requer profissionais treinados para aplicação da formulação, além do desconforto que se observa após a aplicação.

As vias retal e nasal foram testadas, mas abandonadas devido ao desconforto para o paciente que necessitava de várias aplicações diárias (NIESCHLAG; BEHRE, 1998).

Com o sucesso dos implantes subcutâneos comparativamente a outras vias, várias pesquisas começaram a ser direcionadas para a via transdérmica (NICOLAZZO et al., 2005). Esta rota de administração não é invasiva, nem dolorida, não requer profissional especializado para a aplicação e permite desenvolver sistemas com controle

de liberação. Como consequência, há uma maior adesão do paciente ao tratamento (BAERT et al., 2012). A possibilidade de reduzir a frequência de administração e dose, bem como a necessidade de se aumentar a permeação da TST em curto intervalo temporal representaram os pilares deste novo tipo de formulação (PABLA; ZIA, 2007).

### 2.4. PREPARAÇÕES TRANSDÉRMICAS

Com o desenvolvimento das formulações transdérmicas, as desvantagens anteriormente citadas foram superadas. Um dos benefícios clínicos descritos é a liberação do fármaco de forma sustentada por longos períodos, já que a administração transdérmica possui cinética de ordem zero, o que mantém a concentração plasmática estável por maior tempo, sem a flutuação apresentada pelas outras formas farmacêuticas (KIM; LEE; KIM, 2000). Além do controle da formulação em si, a própria pele atua como uma barreira que limita a quantidade de fármaco que alcança a circulação sanguínea.

Outra vantagem, como já mencionado, é o fato destas formulações evitarem o metabolismo de primeira passagem. Como consequência, não é mais necessário utilizar altas concentrações de TST devido a sua inativação, o que pode reduzir a ocorrência de efeitos colaterais (BAERT *et al.*, 2012; GHAFOURIAN *et al.*, 2004; LEICHTNAM *et al.*, 2006a). É um tipo de forma farmacêutica que proporciona alta adesão terapêutica devido à facilidade da administração, com a possibilidade de remoção imediata caso se observem efeitos indesejáveis (BAERT et al., 2012; THOMAS; FINNIN, 2004; WILLIAMS; BARRY, 1992). São indolores durante a aplicação e não causam o desconforto de preparações nasais e/ou retais (KANG *et al.*, 2013; KOVÁČIK; KOPEČNÁ; VÁVROVÁ, 2020). Todas estas vantagens justificam o grande número de formulações comerciais atualmente disponíveis para a TST.

As formulações transdérmicas incluem diversos tipos de formas farmacêuticas/veículos. Podem variar entre loções, cremes, unguentos, géis, adesivos, aerossóis e espumas. Em comum, todas estas preparações devem ser capazes de fazer com o que o ativo penetre o estrato córneo, permeando os vários tecidos até alcançar a corrente sanguínea (AULTON; TAYLOR, 2016).

Os cremes, géis e adesivos são os veículos mais utilizados para a TST. Os *cremes* são formulações semissólidas bifásicas constituídas de uma fase oleosa e uma fase aquosa, as quais são estabilizadas por uma agente emulgente. Os *géis*, por sua vez, são sistemas semissólidos em que uma fase líquida está confinada dentro de uma matriz polimérica. Por fim, e não menos importante, há os *adesivos transdérmicos*, que são formas farmacêuticas sólidas que contêm polímeros associados a um sistema aquoso ou um solvente volátil que forma um filme para a aplicação do fármaco (AULTON; TAYLOR, 2016). Os adesivos foram as primeiras formulações transdérmicas desenvolvidas para a TST (KAUFMAN, 2004). Além das vantagens já citadas, adesivos transdérmicos são capazes de liberar uma dose constante e controlada da TST, algo de extrema importância para os casos em que efeitos adversos são sensíveis a variações na concentração plasmática do fármaco (KANG *et al.*, 2013). Por outro lado, podem causar irritação local devido à oclusão do tecido e/ou alta concentração de TST usada. Quando há movimentação vigorosa e/ou excesso de suor (ex.: durante os exercícios), pode ocorrer desprendimento da formulação (JORDAN, 1997; THOMAS; FINNIN, 2004).

Adesivos foram inicialmente desenvolvidos para serem utilizados na bolsa testicular, que possui um estrato córneo mais fino e é um tecido muito vascularizado, diferentemente de outras áreas do corpo. Anteriormente a aplicação, era necessário que os pelos fossem removidos para facilitar a fixação do adesivo, o que gerava desconforto aos pacientes. Além disso, em alguns casos, o tamanho do adesivo era superior ao tamanho da bolsa escrotal, já que muitos pacientes apresentavam hipogonadismo, o que também acabava por reduzir a adesão ao tratamento (CUNNINGHAM, 1989). O Testoderm<sup>®</sup> foi o primeiro adesivo transdérmico lançado no mercado norte-americano, em 1994, o qual promovia uma liberação diária entre 10 a 15 mg do hormônio (CUNNINGHAM, 1989; KAUFMAN, 2004; NIESCHLAG; BEHRE, 1998). A concentração plasmática máxima é alcançada 2 a 3 h após a aplicação e, quando aplicado pela manhã, pode simular a evolução circadiana natural do organismo (KAUFMAN, 2004). Devido ao local de aplicação e a necessidade de remoção frequente dos pelos, esse tratamento não foi bem aceito pelos pacientes.

Apesar do insucesso dos adesivos escrotais, esta rota de administração continuou a ser explorada pela série de vantagens já citadas. Adesivos transdérmicos aplicados em outras partes do corpo passaram a ser desenvolvidos/estudados. O Testoderm<sup>®</sup> começou a ser aplicado na região dos braços, costas e nádegas (KAUFMAN, 2004). Estudos farmacocinéticos foram realizados por Meikle *et.al.* (1996) para entender o impacto da região de aplicação na concentração plasmática da TST. A biodisponibilidade do hormônio foi maior para os adesivos posicionados nas costas, seguido da coxa, braço superior, abdômen, peito e canela. O estudo ressaltou a importância de se utilizar os adesivos nas quatro primeiras regiões a fim de otimizar a absorção do hormônio. Como alguns usuários destes adesivos relataram dermatite e descolamento do adesivo (KAUFMAN, 2004), a indústria continuou a busca por novas formas farmacêuticas que pudessem melhorar a adesão ao tratamento. Frente a isto, géis transdérmicos foram propostos.

Após a administração de gel de TST a 1%, Wang et al. (2000) observaram um rápido alcance da concentração plasmática máxima (30 min após a aplicação), com redução ao longo de 24 h. Após remoção da formulação, os níveis séricos de TST caíram entre 40 e 50% dentro de 48 h. A formulação desenvolvida não causou irritação cutânea e a rápida secagem após a aplicação reduziria a contaminação acidental entre indivíduos.

O Androgel<sup>®</sup> (Unimed Pharmaceuticals, GA, USA) foi o primeiro gel transdérmico de TST aprovado nos Estados Unidos no ano 2000. Um ano após a sua comercialização, esta forma farmacêutica já passou a ter a preferência dos consumidores em relação aos adesivos (HADGRAFT; LANE, 2015). Disponibilizado comercialmente nas dosagens de 5, 7,5 e 10 g, foi desenvolvido para ser aplicado nos ombros, parte superior dos braços e/ou abdômen, atingindo níveis terapêuticos entre 30 min e 4 h após a aplicação. A estabilização terapêutica ocorre entre 1 e 3 dias da primeira administração (SWERDLOFF et al., 2000). Nesta primeira formulação desenvolvida, apenas 10% da dose total de TST encontra-se biodisponível para a absorção (WANG et al., 2000), abrindo um campo para o aprimoramento de formulações desta categoria.

Testim<sup>®</sup> Três anos depois, lancou-se o gel transdérmico (Ferring Pharmaceuticals, UK). Em comparação com o Androgel<sup>®</sup>, possui maior absorção e, desta forma, não podem ser considerados bioequivalentes. Isto se deve a presença do promotor de permeação pentadecalactona (HADGRAFT: LANE. 2015; MCNICHOLAS et al., 2003). Em 2010 e 2014, lançaram-se no mercado os géis Fortesta<sup>®</sup> e Volexo<sup>®</sup>, respectivamente (DOBS et al., 2012; HADGRAFT; LANE, 2015).

Como as formas farmacêuticas em gel ficam expostas, diferentemente das formulações adesivas, podem contaminar parceiras e crianças do usuário. Esta contaminação acidental pode levar a efeitos indesejados como amadurecimento da idade óssea com comprometimento do crescimento da criança, aumento dos órgãos sexuais, desenvolvimento de puberdade precoce, aumento de libido e comportamentos agressivos (HADGRAFT; LANE, 2015; PABLA; ZIA, 2007; THOMAS; FINNIN, 2004).

# 2.5. CARACTERÍSTICAS DESEJÁVEIS PARA FÁRMACOS INCORPORADOS EM SISTEMAS TRANSDÉRMICOS

No delineamento de formulações transdérmicas, deve-se analisar as propriedades físico-químicas do fármaco tais como solubilidade e o pka a fim de definir a composição da formulação. Caso este fármaco apresente baixa solubilidade aquosa, por exemplo, agentes solubilizantes podem ser incluídos a fim de evitar fenômenos de separação de fase, precipitação ou problemas de uniformidade de conteúdo. O pka, por sua vez, interfere na relação entre espécies ionizadas e não ionizadas, trazendo impactos tanto na solubilidade quanto absorção. A formulação deve se manter estável até o momento da sua utilização, seus componentes devem ser compatíveis entre si e uma quantidade suficiente de fármaco deve ser liberada do sistema. O fármaco deve ultrapassar regiões lipídicas do estrato córneo, a principal barreira da pele, bem como regiões hidrofílicas da epiderme até o alcance da circulação sanguínea. Candidatos ideais para esta rota de administração devem apresentar valores de log P intermediário (entre 1 e 4), baixos valores de massa molar (menor que 500 g/mol) e ponto de fusão menor que 250°C. Para adesivos transdérmicos, a dose terapêutica deve ser baixa (menor que 10 mg por dia) ou os fármacos devem ser potentes já que a formulação apresenta dimensões menores (diferente das preparações transdérmicas semissólidas que podem ser aplicadas em uma área maior). É desejável também que o fármaco apresente um tempo de meia-vida intermediário, pois uma metabolização muito alta exigiria altas doses de fármaco na formulação (AULTON; TAYLOR, 2016).

# 2.6. ASPECTOS BIOFARMACÊUTICOS E CARACTERÍSTICAS FÍSICO-QUÍMICAS DA TESTOSTERONA

A TST apresenta baixa permeação cutânea, o que pode ser associado à sua alta afinidade pela pele e aos receptores específicos existentes no tecido cutâneo. A existência destes receptores se deve à variedade de funções que a TST exerce no organismo, como já mencionado. Os receptores localizam-se nos queratinócitos da epiderme e fibroblastos da derme. A TST é metabolizada a di-hidrotestosterona pela enzima 5 $\alpha$ -redutase e a estradiol pela aromatase presente no tecido adiposo e na pele (CHEN; THIBOUTOT; ZOUBOULIS, 2002).

A TST é um hormônio relativamente lipofilico, com um coeficiente de partição óleo/água (log de P) entre 3,32 e 3,42 (HADGRAFT; LANE, 2015; KIM; LEE; KIM, 2000; MISRA *et al.*, 1996). Sua solubilidade em água é de 0,039 mg/mL a 37°C (OKIMOTO; RAJEWSKI; STELLA, 1998). O Quadro 2 e a Figura 2 apresentam informações físico-químicas/farmacocinéticas e a estrutura química da TST, respectivamente.

Quadro 2. Propriedades físico-químicas e farmacocinéticas da testosterona.

Massa molar	288,4 g/mol
Intervalo de fusão	152 a 157°C
Log P	3,32 a 3,42
Solubilidade	0,039 mg/mL a 37°C
Tempo de meia-vida $(t_{1/2})$	10 a 100 min
Ligação a proteínas plasmáticas	97 a 98%
Clearence renal	1272 ± 168 L/dia





Fonte: Hadgraft & Lane (2015)

Seu tempo de meia vida varia entre 10 a 100 min no plasma (HELLMAN; ROSENFELD, 1974). Seu grupo 17-OH sofre oxidação hepática, onde é reduzida a dihidrotestosterona e estradiol. Os metabólitos resultantes ainda sofrem conjugação, glicuronidação e sulfonação para serem secretados pela urina. Apenas 6% do hormônio é excretado na sua forma original (HADGRAFT; LANE, 2015).

Na circulação sistêmica, a TST encontra-se ligada fortemente a globulinas específicas de ligação de hormônios sexuais e fracamente ligada a albuminas plasmáticas. Apenas 2% é encontrada na forma livre (PARDRIDGE, 1986).

Ainda que a TST apresente baixa massa molar e log P intermediário, observa-se baixa permeação cutânea pelas razões já expostas. Por outro lado, concentrações plasmáticas baixas são requeridas para um efeito terapêutico, o que torna a rota transdérmica atrativa. Ao mesmo tempo em que a alta afinidade do composto dificulta sua passagem pela derme, ao alcançar esta região, a TST é capaz de chegar até a circulação sistêmica ou ainda se depositar nos tecidos subjacentes. Os tecidos subcutâneos apresentam alta quantidade de células de gordura, formando uma espécie de reservatório do fármaco. Com isto, ocorre uma liberação lenta e gradual da TST na circulação sanguínea (ČEPONIS et al., 2017; WANG et al., 2000).

### 2.7. ASPECTOS ANATÔMICOS E FISIOLÓGICOS DA PELE

A pele é considerada o maior órgão do organismo, com uma área de 1,8 m<sup>2</sup>. É constituída por várias camadas de células (figura 3). Além de exercer propriedades de barreira a xenobióticos e microrganismos (BAERT et al., 2012), contribui para a manutenção da hidratação corpórea e regulação da temperatura corporal (FOX *et al.*, 2011). Sua espessura varia de acordo com o local e a idade do indivíduo (ABD *et al.*, 2016)



Figura 3. Estrutura da pele humana.

Fonte: traduzido e adaptado de Britannica (2018).

A *epiderme* está localizada acima da derme. Na sua composição, aparecem diferentes tipos celulares tais como queratinócitos, melanócitos e células de Langherans (PALMER; DELOUISE, 2016). Os queratinócitos se originam no <u>estrato basal</u> e se diferenciam conforme migram para as camadas mais externas, formando o estrato espinhoso, o estrato granuloso e, por último, o estrato córneo (ABD *et al.*, 2016; AULTON; TAYLOR, 2016). Estima-se que esta diferenciação celular dure aproximadamente 14 dias. No estrato basal, as células apresentam organelas típicas, porém, ao alcançar o estrato córneo, mostram-se anucleadas, achatadas e desvitalizadas (AULTON; TAYLOR, 2016). Por este motivo, a epiderme é subdividida em viável e estrato córneo (ALJUFFALI; LIN; FANG, 2014).

O *estrato córneo* possui de 10 a 15 camadas de células, chamadas de corneócitos, com uma espessura entre 10 e 20 µm (BROWN *et al.*, 2006; ZHANG; FANG; FANG; FANG, 2011), que varia dependendo da região do corpo e hidratação da pele. Na sola dos pés e palma das mãos, como são regiões de atrito, há presença de uma camada de células adicional, o estrato lúcido. Ao contrário de outras barreiras lipídicas do corpo, as células do estrato córneo são desvitalizadas e constituídas basicamente por
lipídeos tais como ceramidas (40 a 50%), ácidos graxos (15 a 25%), triglicerídeos (10%), colesterol (25%) e queratina (BAKONYI *et al.*, 2018; CHEN *et al.*, 2016). As ceramidas interconectam os corneócitos resultando em uma estrutura semelhante a uma parede de tijolos - corneócitos preenchidos de queratina seriam os "tijolos" e o "cimento" seriam os lipídeos. As células do estrato córneo permanecem na superfície da pele por aproximadamente 2 semanas até serem substituídas/descamadas (AULTON; TAYLOR, 2016).

A *derme* encontra-se acima do tecido subcutâneo e abaixo da epiderme. Possui uma espessura maior que a epiderme, entre 3 e 5 mm. É constituída de colágeno e elastina banhados por um gel de mucopolissacarídeos formando uma estrutura semelhante a um hidrogel. É nela onde se encontram os anexos da pele (folículos pilosos, glândulas sebáceas e sudoríparas). É a porção metabolicamente ativa da pele por ser altamente irrigada pelos vasos sanguíneos que chegam até 0,2 mm abaixo da superfície, no limite entre a derme e a epiderme. Por esse motivo, é responsável por realizar o controle da temperatura corporal, reparar danos teciduais, bem como realizar a reposição de oxigênio e nutrientes na pele (AULTON; TAYLOR, 2016). Por ser uma camada essencialmente hidrofílica, não interrompe o fluxo da maioria das moléculas, com exceção das excessivamente lipofílicas (KOVÁČIK; KOPEČNÁ; VÁVROVÁ, 2020).

O *tecido subcutâneo* é a camada mais profunda da pele (ABD *et al.*, 2016). É formado por lóbulos de gordura, com uma espessura variável dependendo da localização na superfície corpórea, sendo mais espesso no abdômen e mais fino na bolsa escrotal (IYER *et al.*, 2017). Tem como função a proteção mecânica contra impactos, auxilia na manutenção da temperatura corpórea e estoca energia para as atividades metabólicas (AULTON; TAYLOR, 2016).

Por fim, tem-se os *anexos da pele*, que podem funcionar como "atalhos" para o transporte de moléculas até a corrente sanguínea. Os folículos pilosos estão associados as glândulas sebáceas. Estão presentes entre 50 e 100 folículos pilosos por cm<sup>2</sup> de pele. As glândulas sebáceas são responsáveis pela produção do sebo, o qual é composto por ácidos graxos, ceras e triglicerídeos, que atua lubrificando e auxiliando na manutenção do pH da pele, que é levemente ácido (AULTON; TAYLOR, 2016).

### 2.8. TRANSPORTE DE MOLÉCULAS ATRAVÉS DA PELE

Na pele, as moléculas podem seguir três caminhos distintos: rotas intracelulares, intercelulares e apêndices – folículo ou glândulas sudoríparas (Figura 4) (ALJUFFALI; LIN; FANG, 2014; HERMAN; HERMAN, 2015). As rotas intra- e intercelulares são ambos processos de difusão passiva, ou seja, o fármaco é transportado a favor de um gradiente de concentração. Na rota intracelular ou transcelular, o fármaco atravessa os corneócitos que compõe o estrato córneo, passando pelo seu interior repleto de queratina. Já na rota intercelular, a difusão ocorre de forma labiríntica, no meio intercelular lipídico entre os corneócitos. As rotas de desvio seguem pelos anexos da pele, difundindo-se pelo suor ou pelo sebo contra o fluxo das glândulas até atingirem o vaso sanguíneo que irriga estas estruturas (KNORR *et al.*, 2009a; PATZELT; LADEMANN, 2013). Compostos hidrofílicos tendem a utilizar as rotas de desvio como a intracelular ou transcelular (MITRAGOTRI, 2003; TODO; MOHD, 2017).



Figura 4. Processos que ocorrem durante a liberação transdérmica de fármacos

Fonte: traduzido e adaptado de Palmer & Delouise (2016).

# 2.9. MODELOS PARA AVALIAÇÃO DA PERMEAÇÃO CUTÂNEA DE FÁRMACOS

Estudos *in vivo* em humanos representam o padrão ouro para a análise da permeação cutânea de fármacos (LAI; MAIBACH, 2009), porém, são difíceis de serem realizados já que exigem autorização de comitês de ética bem como a participação de voluntários. Além disso, as respostas destes ensaios são mais difíceis de serem calculadas e interpretadas por apresentarem grande variação entre os indivíduos (ABD *et al.*, 2016). Por este motivo, modelos *in vitro/ex vivo* ainda têm sido priorizados, pelo menos em uma etapa inicial de triagem de diferentes moléculas ou desenvolvimento de novas formulações.

Idealmente, peles de tecido humano devem ser utilizadas nestes experimentos, as quais são frequentemente obtidas de cirurgias plásticas e de cadáveres, com a devida aprovação ética de uso. Deve-se ainda levar em consideração que as diferenças de absorção das diferentes partes do corpo (ABD *et al.*, 2016; ROUGIER; LOTTE; MAIBACH, 1987) decorrentes da composição do tecido, hidratação e espessura influenciam nos resultados (MARRAKCHI; MAIBACH, 2007; SANDBY-MØLLER; POULSEN; WULF, 2003).

Algumas técnicas utilizadas para a análise da penetração/permeação de fármacos incluem o *tape stripping*, a microdiálise ou microperfusão *in vivo* e os testes de permeação *ex vivo* utilizando células de difusão (RANEY *et al.*, 2015).

O ensaio do *tape stripping* considera apenas o fármaco presente no estrato córneo, não monitorando a quantidade do mesmo presente nas camadas subjacentes ou corrente sanguínea. Por esta razão, é recomendado para comparar a *performance* de formulações tópicas (ESCOBAR-CHAVEZ *et al.*, 2008; SURBER; SCHWARB; SMITH, 2001). Neste ensaio, fitas adesivas sucessivas são aplicadas no estrato córneo para a remoção dos corneócitos contendo o fármaco. Na sequência, o fármaco é extraído destas células com um solvente específico e dosado por técnicas analíticas que sejam sensíveis o suficiente para detectar a quantidade de fármaco ali presente (SHAH; MAIBACH; JENNER, 2014). O operador deve remover as fitas aplicando sempre a mesma força de modo a garantir maior reprodutibilidade dos resultados. (ABD *et al.*, 2016).

Na microdiálise, uma cânula semipermeável é inserida sob a pele, alcançando a derme. No interior desta cânula, há uma solução de perfusato que mimetiza a circulação sistêmica. Esta disposição deve permitir a troca de substâncias entre o fluido extracelular e a sonda, e assim determinar a concentração local do fármaco para que se possa traçar um perfil de concentração de fármaco ao longo do tempo (ABD *et al.*, 2016). Como desvantagem desta técnica, pode-se destacar a possibilidade do desenvolvimento de inflamações locais que interferem no fluxo dos xenobióticos e moléculas endógenas (STENKEN *et al.*, 2010) e a grande variação pessoal nos resultados interindividuais (BENFELDT *et al.*, 2007).

Como ambos os ensaios mencionados acima apresentam muita variação, o FDA recomenda a utilização dos ensaios *in vitro/ex vivo* devido a maior correlação com os resultados *in vivo*, além da alta reprodutibilidade e precisão. Ensaios em células de difusão utilizando pele humana representam modelos promissores neste sentido já que

conseguem predizer com certa confiabilidade resultados de biodisponibilidade (RANEY *et al.*, 2015). Além disto, este modelo tem sido utilizado para análise de bioequivalência de doses (ABD *et al.*, 2016).

O modelo bicompartimental das células de difusão de Franz (figura 5) é o mais tradicionalmente utilizado nos ensaios de permeação cutânea, o qual pode apresentar diferentes configurações (sistemas horizontal e vertical). No compartimento superior ou doador, é inserida a solução do fármaco ou formulação-teste. Quantidade finitas ou infinitas de fármaco podem ser consideradas. Em geral, doses infinitas são utilizadas para a composição de sistemas transdérmicos. No compartimento inferior, encontra-se o fluido receptor, o qual deve mimetizar o comportamento fisiológico do sangue, manter a integridade do tecido e uma completa solubilização do fármaco. A presença de um agitador magnético neste compartimento visa garantir uma distribuição uniforme do fármaco na solução, impedindo o acúmulo do fármaco na região abaixo da membrana. Além disto, há a presença de uma jaqueta de água externa para garantir um controle da temperatura do experimento, que deve ser de 37 ou 32°C (a depender da literatura consultada). As amostras para análise são retiradas deste último compartimento em intervalos pré-definidos. Por fim, o fármaco é quantificado a fim de se conhecer a quantidade permeada (AULTON; TAYLOR, 2016).

Figura 5. Célula de difusão de Franz.



Fonte: elaborado pelo autor.

Embora a pele humana represente o tecido ideal para este tipo de ensaio, limitações relativas a disponibilidade de material e questões éticas tem contribuído para uso de tecidos de animais como suínos, ratos, camundongos, porquinho da índia e cobras (ABD et al., 2016; HERMAN; HERMAN, 2015). Por outro lado, a composição da pele (quantidade de camadas de células, composição lipídica e proteica), espessura das camadas que a compõe, porosidade e irrigação sanguínea varia entre as espécies. Além disto, observam-se diferenças relativas à quantidade de apêndices por unidade de área (pêlos e glândulas). Alguns animais possuem mais pêlos do que o tecido cutâneo humano (GODIN; TOUITOU, 2007; HUEBER; SCHAEFER; WEPIERRE, 1994; LIN et al., 1992; MONTEIRO-RIVIERE, 1991). Cicatrizes como estrias também afetam a permeação cutânea devido a diferenças na composição da pele (MILLS; MAGNUSSON; CROSS, 2006). A extrapolação dos resultados de estudos nos diversos tipos de pele apresenta limitações e deve ser analisada com cuidado (WALTERS; ROBERTS, 2002). A aplicação de uma mesma formulação em diferentes partes do corpo humano também pode resultar em diferenças na quantidade absorvida (ORIBA; BUCKS; MAIBACH, 1996).

Como alternativa às membranas biológicas, há as membranas sintéticas e os modelos de pele reconstruída. As membranas sintéticas são compostas por acetato de celulose, dimetilpolissiloxano, hidrocarbonetos, álcoois de cadeia longa, miristato de isopropila, lipídios cutâneos (OTTAVIANI; MARTEL; CARRUPT, 2006), colesterol, ácidos graxos livres e análogos de ceramidas (ceramidas sintéticas, diamidas de ácido tartárico de cadeia longa) (SINKÓ et al., 2012). Essas membranas podem ser colocadas na interface das câmaras de difusão ou preparadas diretamente no aparato PAMPA (Parallel Artificial Membrane Permeability Assay). O ensaio PAMPA é geralmente utilizado para uma triagem rápida de fármacos transportados passivamente através da pele. Nele, uma membrana lipídica artificial separa os compartimentos doador e receptor em placas de 96 cavidades (NEUPANE et al., 2020). Já nos modelos de pele reconstruída, as camadas de células humanas são geralmente cultivadas em uma matriz polimérica. Diferentes tipos de células podem ser considerados para alcançar uma estrutura com a composição e complexidade desejadas. Modelos reconstruídos podem ser projetados para simular tanto a epiderme quanto a pele de espessura total (FLATEN et al., 2015; KÜCHLER; STRÜVER; FRIESS, 2013).

#### 2.10. PROMOTORES DE ABSORÇÃO

Tendo em vista as propriedades de barreira do estrato córneo, promotores ou facilitadores de penetração/permeação podem ser incluídos na formulação. Estes compostos alteram o arranjo celular do tecido, facilitando a difusão do fármaco (PABLA; ZIA, 2007). Não devem possuir atividade farmacológica, ou seja, se ligar a receptores desencadeando respostas. Além disto, devem modificar a barreira da pele de forma "reversível". Uma vez que o promotor é removido, as ligações entre os lipídeos começam a se refazer, reconstituindo a estrutura original (KANG *et al.*, 2013). Ainda, devem ser atóxicos, não alergênicos, não irritantes e compatíveis com os componentes da formulação, inclusive o fármaco (HERMAN; HERMAN, 2015).

Os promotores de permeação podem agir rompendo a estrutura lipídica intercelular entre os corneócitos, fluidizando o estrato córneo; interagir com as proteínas do meio intercelular, melhorando a partição da molécula; agir nas conexões dos desmossomos entre os corneócitos ou alterar a atividade metabólica do tecido (Figura 6). Todos estes mecanismos podem modificar o arranjo estrutural do tecido e, consequentemente, diminuir as propriedades de barreira do estrato córneo (SILVA *et al.*, 2010; HERMAN; HERMAN, 2015; WILLIAMS; BARRY, 1991). O mecanismo preferencial dependerá das propriedades físico-químicas do fármaco e do promotor (GHAFOURIAN *et al.*, 2004).

A extração de lipídeos é realizada pelas formulações que contem álcoois como o etanol em concentrações de 40 a 80%. Esta extração promove a permeação de fármacos polares e apolares através de "poros" que são formados no tecido (DRAGICEVIC-CURIC; MAIBACH, 2015; LEVANG; ZHAO; SINGH, 1999). O etanol também é responsável pela fluidização dos lipídeos do estrato córneo em concentrações entre 20 a 60%, o que melhora a permeação de fármacos lipofílicos principalmente pela rota intercelular. A desorganização da camada lipídica também pode ocorrer devido a diferenças na hidratação do tecido (DRAGICEVIC-CURIC; MAIBACH, 2015).



Figura 6. Mecanismos de ação dos promotores de permeação.

Fonte: elaborado pelo autor.

Existem diversos tipos de promotores de permeação, desde sintéticos a naturais. Alguns representantes incluem sulfóxidos como o DMSO, Azone<sup>®</sup>, (primeiro composto desenvolvido para esta finalidade), pirrolidona, ácidos graxos, álcoois e glicol, tensoativos, ureia, óleos essenciais como os terpenos e fosfolipídios, dentre outros (KANG *et al.*, 2013; WILLIAMS; BARRY, 2012).

Neste trabalho, terpenos foram selecionados como agentes de permeação dada a origem natural que reduz chances de irritação ou problemas de biocompatibilidade (comuns para os promotores sintéticos). Estes compostos são metabólitos secundários e lipofílicos derivados do mevalonato e pirofosfato de isopentenilo. Aparecem na composição de óleos essenciais de vários produtos naturais (KAMATOU; VILJOEN, 2010).

De acordo com a quantidade de unidades de isoprenos que possuem, são classificados em hemiterpenos (C<sub>5</sub>) monoterpenos (C<sub>10</sub>), sesquiterpenos (C<sub>15</sub>), diterpenos (C<sub>20</sub>), sesterterpenos (C<sub>25</sub>), triterpenos (C<sub>30</sub>) e tetraterpenos (C<sub>40</sub>), os quais

apresentam 1, 2, 3, 4, 5, 6 e 8 unidades de isopreno, respectivamente (KANG *et al.*, 2013; YAZAKI; ARIMURA; OHNISHI, 2017). Existe também uma outra classificação que se baseia na quantidade de anéis aromáticos (KANG *et al.*, 2013).

Para estudar o mecanismo de ação dos promotores de ação sobre o tecido, técnicas termoanalíticas (ex.: calorimetria exploratória diferencial – DSC) e espectroscópicas (espectroscopia no infravermelho por transformada de Fourier – FTIR) podem ser consideradas (SAPRA; JAIN; TIWARY, 2008). Comparando-se as endotermas e as exotermas da temperatura de transição ( $T_m$ ), juntamente com as variações de entalpia ( $\Delta$ H), é possível extrair informações a respeito do grau de desordem da camada lipídica e da desnaturação de proteínas (DRAGICEVIC-CURIC; MAIBACH, 2015). Termogramas de DSC de estrato córneo humano são caracterizados por três endotermas principais: 65°C, 75°C e 95°C. A primeira T<sub>m</sub> indica fusão lipídica. A segunda endoterma tem relação com a quebra do complexo entre lipídeos e proteínas e a terceira podem ser associada à desnaturação das proteínas (LIM; LIU; CHAN, 2009).

As análises de FTIR apontam para mudanças conformacionais e moleculares nos lipídeos e proteínas do estrato córneo. Variações de alargamento/estreitamento bem como intensidade das bandas de FITR em relação a um controle (tecido não tratado) podem sugerir modificações químicas no estrato córneo. Na região entre 2.920 e 2.850 cm<sup>-1</sup>, observam-se vibrações de C-H assimétricos e simétricos, respectivamente, as quais estão relacionadas à fluidização de lipídeos (LIM; LIU; CHAN, 2009). A quantidade de lipídeos está relacionada a intensidade e largura da banda. Alterações na intensidade e largura da banda sinalizam extração de lipídeos da pele (CHEN *et al.*, 2016). As vibrações C=O na região de 1.740 cm<sup>-1</sup> referem-se as cabeças polares dos lipídeos. Variações na região entre 3.000 e 3.600 cm<sup>-1</sup> sinalizam alterações no grau de hidratação. As bandas de 1.650 cm<sup>-1</sup> e 1.550 cm<sup>-1</sup> representam as amidas I e II das proteínas do estrato córneo (CHEN *et al.*, 2016; SAPRA; JAIN; TIWARY, 2008).

O principal mecanismo de ação dos terpenos é a alteração da fluidez do estrato córneo através da interação com os lipídeos intercelulares (HERMAN; HERMAN, 2015; JIANG *et al.*, 2017). O grau de desordem proporcionado por estes agentes parece estar relacionado com o tamanho da cadeia alquila, já que terpenos menores tem se mostrado mais efetivos que os sesquiterpenos (MOGHADAM *et al.*, 2013;

WILLIAMS; BARRY, 2012). A presença de anéis aromáticos mostrou um efeito negativo sobre a absorção de fármacos comparativamente aqueles terpenos com cadeias alquílicas longas (MOGHADAM *et al.*, 2013). Além disso, terpenos com baixo ponto de fusão mostraram maior ação reforçadora da permeação (JIANG *et al.*, 2017).

Em geral, terpenos com grupamentos polares ou oxigênio na sua estrutura tem se mostrado mais efetivos como promotores de permeação de moléculas hidrofílicas. Por outro lado, terpenos com hidrocarbonetos tem garantido um maior incremento da permeação de moléculas mais hidrofóbicas (GODWIN; MICHNIAK, 1999; KANG *et al.*, 2013; WILLIAMS; BARRY, 2012).

Kang *et al.* (2013) estabeleceram uma relação entre a presença de grupamentos químicos específicos no terpeno e a capacidade de promoção da permeação. Além de mencionar a importância do tamanho molecular, como já descrito, sugeriu a seguinte escala de permeação: éster > aldeído > óxido > hidrocarboneto > álcool > cetona > fenol > ácido. Assim, terpenos com grupamentos ésteres seriam os mais efetivos e os com grupamentos ácidos os menos efetivos.

Os terpenos são reconhecidamente seguros pela FDA (GRAS, *generally recognized as safe*), fato extremamente relevante já que muitos destes agentes podem penetrar a pele e alcançar a corrente sanguínea (SAPRA; JAIN; TIWARY, 2008; SONGKRO; RADES; BECKET, 2009a). Além de serem rapidamente metabolizados e excretados, (HERMAN; HERMAN, 2015), terpenos apresentam atividades farmacológicas específicas (WILLIAMS; BARRY, 2012), o que traz vantagens na utilização.

Nas formulações transdérmicas, geralmente emprega-se 5% (v/v) destes agentes, que é suficiente para a promoção da permeação de moléculas. O principal desafio é encontrar a concentração que promove alta permeação e, ao mesmo tempo, não resulte em toxicidade ou reações alérgicas/irritantes (KANG *et al.*, 2013).

Os terpenos selecionados para este trabalho foram o alfa-bisabolol, o carvacrol e o mentol (figura 7; tabela 1), os quais foram solubilizados em propilenoglicol, um cossolvente que age sinergicamente com os terpenos favorecendo a permeação de moléculas como a TST (ALAN ANDERSEN, 1999; BARRY; WILLIAMS, 1989; CHEN *et al.*, 2016; SONGKRO; RADES; BECKET, 2009a).





Fonte: Lim & Liu & Chan (2009).

C	)uadro 3.	Propr	iedades	físico-c	uímicas	dos ter	penos o	x-bisabo	olol,	carvacrol	e mentol.

Terpeno	Classe	Massa molar (g/mol)	Ponto de fusão (°C)	Ponto de ebulição (°C)	Log P
α-bisabolol	Sesquiterpeno	222,37	Líquido	314,5	5,01 a
					5,07
Carvacrol	Monoterpeno	150,22	3,5	237,7	3,28
Mentol	Monoterpeno	156,27	43	215,4	3,20

Fonte: Chen et al. (2016) e Kang et al. (2013).

Os terpenos com valores de log P mais elevados tendem a favorecer a permeação de moléculas mais lipofilicas como a TST, pois misturam-se com os lipídeos intercelulares e, desta forma, promovem maior fluidização e perturbação das propriedades de barreira do estrato córneo. O ponto de ebulição, por sua vez, relaciona-se de forma inversa com a taxa de permeação (CHEN *et al.*, 2016).

Abaixo estão descritas as características individuais dos terpenos utilizados nas formulações desenvolvidas neste trabalho.

#### 2.10.1. Alfa-bisabolol

O  $\alpha$ -bisabolol é um sesquiterpeno monocíclico insaturado. É extraído principalmente da *Matricaria chamomilla*, popularmente conhecida como camomila, mas também pode ser obtido sinteticamente. Foi isolado pela primeira vez em 1951, sendo mais estável na forma  $\alpha$ -(-)-Bisabolol. Possui massa molar de 222,37 g/mol, densidade de 0,929 g/ml, ponto de ebulição de 315°C, sendo praticamente insolúvel em

água, porém, solúvel em álcoois como o propilenoglicol (BRUNKE; HAMMERSCHMIDT, 1985; KAMATOU; VILJOEN, 2010; KANG *et al.*, 2013; PERBELLINI *et al.*, 2004).

É caracterizado por um efeito anti-inflamatório e capacidade de diminuir o tempo de cicatrização (KADIR; BARRY, 1991; VILLEGAS *et al.*, 2001). Possui atividade antimicrobiana frente a patógenos como a *Candida albicans*, porém, menor que a apresentada pelo carvacrol (KAMATOU; VILJOEN, 2010). A literatura informa ainda que não há evidências de potencial teratogênico para o composto (ALAN ANDERSEN, 1999). Estudos apontam também a atividade inseticida e larvicida do  $\alpha$ -bisabolol, atuando como repelente (FURTADO *et al.*, 2005; KAMATOU; VILJOEN, 2010).

Kim *et.al.* (2008) observaram que o  $\alpha$ -bisabolol apresenta ação frente a hiperpigmentação cutânea, sendo útil para tratar patologias em que há aumento da produção de melanina como inflamações da pele, alergias de contato e dermatites ou ainda no melasma induzido por certas formulações (HASSUN; BAGATIN; VENTURA, 2008).

Com relação ao mecanismo de promoção da absorção, acredita-se que o  $\alpha$ bisabolol atue na fluidização dos lipídeos do estrato córneo, facilitando o particionamento de compostos na pele (CUI *et al.*, 2011; KADIR; BARRY, 1991). Cui *et.al.* (2011) destacam ainda que o  $\alpha$ -bisabolol proporcionou a maior taxa de permeação para o cloridrato de propranolol dentre os promotores investigados (bornel e cânfora), além de prevenir a desidratação do estrato córneo e reduzir a irritação cutânea.

#### 2.10.2. Carvacrol

O carvacrol é um monoterpeno fenólico, biossintetizado a partir do  $\gamma$ -terpineno passando pelo *p*-cimeno (CAN BASER, 2008). É líquido a temperatura ambiente, com densidade de 0,976 g/mL a 25°C, sendo insolúvel em água e muito solúvel em etanol, acetona e éter (YADAV; KAMBLE, 2009). Possui ação antimicrobiana, anticancerígena, antioxidante e analgésica (CAN BASER, 2008; SHARIFI-RAD *et al.*, 2018; SILVA *et al.*, 2010). É também indicado para problemas gastrointestinais,

redução dos níveis plasmáticos de colesterol e glicose, tratar dor no reumatismo e cicatrizante (BASER, 2002).

No estudo de permeação realizado por Songkro & Rades & Becket (2009), o carvacrol foi mais efetivo que o mentol no reforço da permeação do hormônio luteinizante. Este resultado foi atribuído as diferenças estruturais dos promotores de absorção (presença do fenol no carvacrol) (JEEVAN; VENKAT; KHAN, 1997; SONGKRO; BECKET; RADES, 2004). Quanto ao mecanismo de ação de promoção da absorção, acredita-se que o carvacrol haja modificando a barreira lipídica intercelular do tecido, com ganhos na absorção mais tardios (após um tempo de latência de 4 h) (LAOTHAWEERUNGSAWAT *et al.*, 2020).

#### 2.10.3. Mentol

O mentol é um monoterpeno monocíclico, apresentando-se como um sólido cristalino à temperatura ambiente. Apresenta densidade de 0,890 Kg/dm<sup>3</sup> a 25°C, ponto de fusão entre 41 e 44°C, é insolúvel em água e solúvel em formulações alcoólicas e propilenoglicol. É considerado seguro pelo FDA, com concentração aprovada para uso externo de até 16% (KAMATOU *et al.*, 2013).

É o óleo essencial das plantas *Mentha canadenses, Mentha x piperita* e *Mentha arvensis*, espécies semelhantes ao hortelã (GALEOTTI *et al.*, 2002; KAMATOU *et al.*, 2013). Pode ainda ser extraído e sintetizado do óleo de citronela, eucalipto e terebintina (GALEOTTI *et al.*, 2002). Produz ação refrescante, sendo cultivado no Japão há mais de 2000 anos atrás com finalidades medicinais (KAMATOU *et al.*, 2013). Foi isolado pela primeira vez em 1771 pelo botânico alemão Gambius (KAMATOU *et al.*, 2013; PATEL; ISHIUJI; YOSIPOVITCH, 2007).

Apresenta atividade antimicrobiana, anticancerígena, e anti-inflamatória, além de funcionar como repelente de insetos (KAMATOU *et al.*, 2013). Além disto, é utilizado para problemas estomacais, resfriado comum e demais complicações respiratórias leves bem como para dores musculares. Confere refrescância ao estimular os receptores de frio, inibindo o influxo de  $Ca^{2+}$  das membranas neuronais (ECCLES, 2011).

Kaplun-frischoff e Touitou (1997) relataram que o mentol foi capaz de aumentar a permeação da TST em oito vezes em relação a solução controle (TST livre) devido à formação de uma mistura eutética que diminuiu seu ponto de fusão de 153,7°C para 39,9°C. Além de aumentar a solubilidade da TST, o mentol parece afetar os espaços intercelulares (JEEVAN; VENKAT; KHAN, 1997).

Ainda, estudos sugerem que a ação do mentol envolve a extração de lipídeos do estrato córneo devido à interação com as cadeias apolares, o que foi confirmado pela redução da intensidade das bandas de FTIR relativas as vibrações de C-H. Estudos de FTIR também indicaram alterações nas vibrações de amidas, sugerindo interação com a queratina dos corneócitos (CHEN *et al.*, 2016).

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# 3. CAPÍTULO II – COMO PLANEJAR ENSAIOS DE PERMEAÇÃO CUTÂNEA PARA FÁRMACOS QUE APRESENTAM LIMITAÇÕES BIOFARMACÊUTICAS COMO A TESTOSTERONA?

Neste capítulo, estudaram-se como condições experimentais tais como espessura de pele, modelo animal, composição das fases doadora e receptora, concentração do fármaco na formulação, índice de supersaturação, tempo do ensaio, quantificação das amostras e retenção cutânea do fármaco afetam a permeação cutânea da testosterona em ensaios realizados em células de difusão do tipo Franz. Este fármaco foi selecionado por apresentar limitações biofarmacêuticas (baixa solubilidade e permeação percutânea) e devido a sua ampla utilização em formulações transdérmicas. O produto deste capítulo é um artigo científico de revisão, que foi publicado na revista *International Journal of Pharmaceutics*.

# HOW TO DEFINE THE EXPERIMENTAL CONDITIONS OF SKIN PERMEATION ASSAYS FOR DRUGS PRESENTING BIOPHARMACEUTICAL LIMITATIONS? THE EXPERIENCE WITH TESTOSTERONE

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#### ABSTRACT

Cutaneous permeation assays are crucial to attest the performance or bioequivalence of topical or transdermal products. Although the official guidelines (e.g., FDA/EMA) play a key role in harmonizing the experimental design, alternative methods are often proposed by the scientific community, which makes it difficult to compare results from different studies. In this review, permeation assays with testosterone (TST) were selected to show this high variability in drug transport rate. The main sources of variation discussed were tissue thickness, animal model, donor and receptor fluid constitution, type of solubilizing agent used in aqueous fluids, drug concentration, degree of supersaturation, skin lipid content, number of experimental times and the physical-chemical stability of the molecule in test fluids. This variation becomes even more critical for molecules that present biopharmaceutical limitations such as TST. In addition, the skin presents specific receptors for this hormone due to its physiological action in this region of the body, which makes the evaluation of the TST transport rate in this tissue even more challenging. The impact of each experimental parameter mentioned above on the flux or permeation coefficient of TST is discussed in detail in the review. Assays used to evaluate tissue integrity are also presented.

**Keywords**: permeation studies; Franz-type diffusion chamber; skin integrity evaluation; testosterone.

## **1. INTRODUCTION**

Many topically administrated therapeutic agents present a limited or suboptimal efficacy due to low penetration into the skin. Solubility and partition coefficient are key aspects related to drug transport through the skin. Overall, high solubility results in high drug concentration in the donor phase (or region of application of the formulation), improving the permeation flux. The drug partitioning from the donor phase into skin layers, in turn, represents the rate-limiting step for drug flux when these molecules are characterized by a poor aqueous solubility taking into account the lipid constitution of biological membranes (Ceschel et al., 2005).

Steroidal hormones, for example, are characterized by a low aqueous solubility (Norman e Litwack, 1997), requiring the investigation of technological/formulation approaches aiming to increase the free drug concentration in the donor phase. Drug complexation (e.g. association with cyclodextrins), incorporation in vesicular or particulate systems (e.g. micro- and nanoparticles) and supersaturated systems have been often considered for this purpose (Ceschel et al., 2005).

On the other hand, the low aqueous solubility of hormones allows a drug partitioning from the donor phase into skin layers. Due to the high value of log P, these compounds bind so strongly to the tissues, particularly the more superficial layers (SC and viable epidermis), resulting in a reservoir effect and low transport rate to the dermis (Magnusson et al., 2006). This event is particularly noticed for TST after application of low concentrations of this agent on the skin (Schlupp et al., 2014).

In addition to the slow tissue diffusion because the high affinity by lipids and high molar mass, the skin presents receptors for these compounds due to the physiological action they play in that region of the body. The TST interacts directly with androgen receptors (ARs), which have been localized in most epidermis keratinocytes. In the dermis, ARs were found in approximately 10% of fibroblasts. AR expression was also found in both basal cells and sebocytes of sebaceous glands whereas it was restricted to dermal papilla cell in hair follicles (Pelletier e Ren, 2004). A modest reduction in epidermal thickness is also observed in TST replacement therapies (Kao et al., 2001).

In view of this high affinity of sex hormones by skin tissues, the inclusion of chemical absorption enhancer combinations able of providing membrane fluidification or lipid extraction is often recommended during the development of novel formulations. These compounds may interact with both lipid and polar domains (Magnusson et al., 2006) and thus chemical absorption enhancers of both polarities should be considered.

The evaluation of percutaneous permeation of molecules is a key step when new dermal or transdermal delivery systems are purposed and different *in silico*, *ex vivo* and *in vivo* models may be considered. Although *in vivo* human assays represent the gold standard, ethical, practical, or economic reasons have limited its use. Alternatively, tissues from different animal models, particularly pig ear skin, have been selected in many permeation studies. These tissues are easier to obtain compared to human tissues, however, high experimental variability may be found. Barrier properties may also vary from human skin depending the selected animal model (Abd et al., 2016).

In addition to the inherent variability of the special animal considered, the tissue thickness and preparation method, composition of the receptor and donor fluid can also represent sources of experimental variation. All these aspects will be discussed detailly in this review article, which considered the TST as a drug model. This drug was selected based on its biopharmaceutical limitations (low solubility and reservoir effect), the various permeation studies already carried out with this compound and its wide use in hormone replacement therapies. Although TST therapy has effectively treated hypogonadism for decades, therapies simpler and more convenient to use, safer and able to mimic physiological levels are still needed (Kaminetsky e Wynia, 2015).

The main objective of this study is to provide tools so that future permeation assays can be planned more rationally considering a series of problems that may be found in each step assay. Once free drug solutions are frequently used as the control during the evaluation of the performance of new topical formulations in skin permeation studies, this review focused on the analysis of studies with free TST solutions/suspensions. The solubility of this molecule in different solvents, the importance of this physicochemical parameter in the design of permeation studies as well as aspects related to the drug quantification step are described in detail.

#### 2. SOLUBILITY STUDIES

Prior to the cutaneous permeation assay, a study of the drug's solubility in different media should be performed. Drug should be soluble in both media used for donor and receptor phases of diffusion chambers and in solvents used to extract the drug from tissues or analytical/quantification step. If a drug is not soluble in the receptor

phase, for example, low drug partitioning into this phase with consequent low transdermal flux is obtained (permeation rate can be underestimated) and solubilizing agents or cosolvents need to be included in these situations. For this reason, many permeation guidelines provide information on the need to carry out these assays under "sink conditions" (OECD, 2004). In other words, the drug concentration should not reach values >10% of its saturation to assure that concentration in the receptor fluid does not limit the drug permeation across the tissue (Azarmi et al., 2007)

For ionizable drugs, solubility study at different pH values should be carefully performed. The ionized-non-ionized drug fraction changes with pH, which impacts not only on solubility but also on the affinity by the biological tissues.

In general, the thermodynamic solubility is evaluated via a long duration incubation (24-72 h) starting with solid materials, which is known as shake-flask method (Zhou et al., 2007). After adding an excess of solid material in given amount of solvent (above saturation), the flasks are placed on an orbital shaker with controlled temperature for 18-72 h (until the system reaches a thermodynamic equilibrium condition). Temperature close to that used in permeation assays should be preferentially selected. Finally, an aliquot of this medium is removed and then centrifuged or filtered through a 0.45 mm membrane for drug quantification.

Ethanol has been one of the most used solvents for the solubilization of drugs. It may also play a role as a skin penetration enhancer (Williams e Barry, 2012), but disruption in the SC may be observed depending on used ethanol concentration. Overall, the permeation assays reported for the TST use high concentrations of ethanol, or even the solvent pure in the donor receptor (Hewitt et al., 2020). Although permeation of lipophilic molecules is less affected by ethanol concentration than hydrophilic molecules (Horita et al., 2012), the establishment of a maximum amount of solvent is an important criterion for not having overestimated permeability (Thomas e Panchagnula, 2003). The solvent evaporation is another concern regarding the use of ethanol in the donor phase. This fact leads to an increase in the TST concentration with formation of precipitates, which impairs the entry of the molecules into the SC. Thus, the evaporation rates depend on ethanol concentration, duration of the assay and temperature used.

Although the solubility of TST is higher in pure ethanol, its combination with water or buffer solutions can be considered for skin permeation studies using Franz-type diffusion cells. Determination of experimental conditions as concentration of TST in the donor chamber and the temperature used in the assay are relevant for the selection of hydroethanolic solution. An increase in ethanol concentration from 20 to 50%, for example, increases the TST solubility in approximately 31-fold. When this parameter is changed from 50 to 70 %, in the same conditions of temperature (37 °C) and agitation time (48 h), the hormone solubility increases about 6 times (Table 1).

Propylene glycol has also been used in various percutaneous permeation studies with TST. The hormone solubility in PBS/propylene glycol (1:1) is approximately 24-fold higher than in pure PBS (Table 1). PG can also act as a chemical permeation enhancer. This agent permeates through human SC and can change the thermodynamic activity of drugs (Williams and Barry, 2012). Carrer et al., (2020) investigated its effect on the transport of molecules with different physicochemical properties. Interestingly, the permeation enhancing effect of PG was more pronounced for hydrophilic compounds. Protein denaturation (solvation the  $\alpha$ -keratin structures of the cells) is the most probable mechanism of action, which can contribute to the barrier disruption and fluidization of intercellular lipids or intracellular expansion (Haq e Michniak-Kohn, 2018).

Medium/solvent	pН	Temperature	Incubation	Solubility	Reference
	•	•	time	·	
HEPES (10 mM + NaCl 0,154 M)	7.4	25 °C	18 h	0.02 mg/mL	
HEPES + HSA (4% w/v)	7.4	25 °C	18 h	0.21 mg/L	
HEPES + β-CD (1,5% w/v)	7.4	25 °C	18 h	0.06 mg/mL	(Schwarz et al.,
HEPES + HP-β-CD (4% w/v)	7.4	25 °C	18 h	3.90 mg/mL	2017)
HEPES + SBE7-β-CD (4% w/v)	7.4	25 °C	18 h	3.23 mg/mL	
HEPES + HSES (1,5% w/v)	7.4	25 °C	18 h	5.72 mg/mL	
HEPES + HTMT (1,5% w/v)	7.4	25 °C	18 h	2.23 mg/mL	
HEPES + HTG (1,5% w/v)	7.4	25 °C	18 h	0.07 mg/mL	
Ethanol/PG/Water (4:1:1)	-	25 °C	12 h	26.99 mg/mL	(Leichtnam et al., 2006)
Isopropyl myristate	-	25 °C	Several	0.03 mg/mL	(Imai et al.,
			days		2016)
Water	-	$25 \pm 1 \ ^{o}C$	72 h	0.02 mg/mL	
PBS	7.4	$25 \pm 1 \ ^{o}C$	72 h	0.02 mg/mL	
PEG 200 water solution (40%)		$25 \pm 1 \ ^{o}C$	72 h	0.32 mg/mL	
Azone®	-	$25 \pm 1 \ ^{o}C$	72 h	92.38 mg/mL	(Lu et al., 2013)
<i>N</i> -methyl-2-pyrrolidone (NMP)	-	$25 \pm 1 \ ^{o}C$	72 h	518.89 mg/mL	

**Table 1.** Testosterone solubility in different solvents and experimental conditions.

Isopropyl myristate         - $25 \pm 1\%$ $72$ h $6.83$ mg/mL           Ethanolwater (70:30)         - $37$ °C $48$ h $68.31$ mg/mL           Ethanolwater (70:30)         - $37$ °C $48$ h $66.71$ mg/mL           Ethanolwater (70:30) + oleic acid         - $37$ °C $48$ h $66.71$ mg/mL         2001)           Ethanolwater (70:30) + lauric         - $37$ °C $48$ h $66.35$ mg/mL         2001)           Ethanolwater (70:30) + lauric         - $37$ °C $48$ h $66.35$ mg/mL         2001)           Ethanolwater (70:30) +         - $37$ °C $48$ h $69.05$ mg/mL         2006)           PBS         - $37$ °C $48$ h $69.05$ mg/mL         2006)           PBS         - $37$ °C $48$ h $69.05$ mg/mL         2006)           PBS         - $30$ °C $24$ h $1.85$ mg/mL         (Mills cal., 2007)           PBS /PG (50:50; w/w)         - $30$ °C $48$ h $1.82$ mg/mL         2006)           PBS /PG (50:50; w/w)         - $30$ °C $48.72$ h $1.30$ mg/mL	PG	-	$25 \pm 1^{\circ}C$	72 h	103.00 mg/mL		
Ethanol/water (70:30)       -       37 °C       48 h       68.31 mg/mL         Ethanol/water (70:30)       -       37 °C       48 h       66.71 mg/mL         Ethanol/water (70:30) + bleic acid       -       37 °C       48 h       66.71 mg/mL         Ethanol/water (70:30) + lauric       -       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) + lauric       -       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) + lauric       -       37 °C       48 h       69.05 mg/mL       2001)         Ethanol/water (70:30) +       -       37 °C       48 h       69.05 mg/mL       2000)         PBS       -       37 °C       24 h       0.06 mg/mL       2006)         PBS       -       37 °C       24 h       1.85 mg/mL       (Mills et al., 2007)         PBS (FG (50:50; w/w)       -       30 °C       24 h       1.48 mg/mL       2006)         PBS (FG (50:50; w/w)       -       30 °C       48 *72 h       0.03 mg/mL       48.72 h       1.48 mg/mL       48.72 h       1.48 mg/mL       49.33 mg/mL       49.33 mg/mL       48.72 h       0.30 mg/mL       1993)       1.abrasol <sup>m</sup> 30 °C       48.72 h       0.30 mg/mL       <	Isopropyl myristate	-	$25 \pm 1^{\circ}\mathrm{C}$	72 h	6.83 mg/mL		
Ethanol/water (70:30)       -       37 °C       48 h       66.71 mg/mL         *Dodecytamine 1% (w/v)       -       37 °C       48 h       66.71 mg/mL       (Kim et al., 2001)         1% (w/v)       -       37 °C       48 h       66.35 mg/mL       2001)         ethanol/water (70:30) + there 101       -       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) + there 101       -       37 °C       48 h       69.05 mg/mL       2001)         Ethanol/water (70:30) +       -       37 °C       48 h       69.05 mg/mL       2001)         PBS       -       37 °C       48 h       69.05 mg/mL       2006)         PBS/F06 (50:50; w/w)       -       37 °C       24 h       1.06 mg/mL       2006)         PBS/F06 (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills et al., 2007)         PBS/F06 (50:50; w/w)       -       30 °C       48 ~2 h       0.43 mg/mL       1993)         Liquid petrolatum       -       30 °C       48 ~2 h       0.43 mg/mL       1993)         Labra6ff       -       30 °C       48 ~2 h       10.40 mg/mL       1993)         Labra6ff       -       30 °C       48 ~2 h <t< th=""><th>Ethanol/water (70:30)</th><th>-</th><th>37 °C</th><th>48 h</th><th>68.31 mg/mL</th><th></th></t<>	Ethanol/water (70:30)	-	37 °C	48 h	68.31 mg/mL		
+Dodecylamine 1% (w/v)       37 °C       48 h       60. /1 mg/mL         Ethanol/water (70:30) + oleic acid       -       37 °C       48 h       68.71 mg/mL       2001)         Ethanol/water (70:30) + lauric       -       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) + lauric       -       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) +       -       37 °C       48 h       69.05 mg/mL       2001)         PBS       -       3°C       24 h       0.06 mg/mL       2006)         PBS/FG (50:50; w/w)       -       3°C       24 h       1.85 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       48.72 h       75.90 mg/mL       2006)         PBS/FG (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       1993)         Labraolf       -       30 °C       48.72 h       10.400 mg/mL       1993)         Labraolf       -       30 °C       48.72 h       10.400 mg/mL       1993)         Labraolf       -       30 °C       48.72 h       13.20 mg/mL	Ethanol/water (70:30)	-	27.00	40.1			
Ethanol/water (70:30) + oleic acid 1% (w/v)       -       37 °C       48 h       68.71 mg/mL (Kim et al., 2001)         Ethanol/water (70:30) + lauric acid 1% (w/v)       -       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) + HFE-101 1% (w/v)       -       37 °C       48 h       69.05 mg/mL       2001)         Ethanol/water (70:30) + transcutol <sup>2</sup> 1% (w/v)       -       37 °C       48 h       69.05 mg/mL       2006)         PBS       -       37 °C       24 h       0.06 mg/mL       2006)         PBS/Ethanol (50:50; w/w)       -       37 °C       24 h       1.85 mg/mL       (Mills et al., 2006)         PBS/PG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48-72 h       1.03 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48-72 h       10.40 mg/mL       (Bonina et al., Labratol <sup>m</sup> 1.993)         Labratol <sup>m</sup> -       30 °C       48-72 h       10.40 mg/mL       (Bonina et al., Labratol <sup>m</sup> 1.993)         Labratol <sup>m</sup> -       30 °	+Dodecylamine 1% (w/v)		37°C	48 h	66.71 mg/mL		
1% (w/v)       37 °C       48 h       68.71 mg/mL       (Kim et al., 2001)         Ethanol/water (70:30) + Huric acid 1% (w/v)       37 °C       48 h       66.35 mg/mL       2001)         Ethanol/water (70:30) + HPE-101       -       37 °C       48 h       69.05 mg/mL       2001)         Ethanol/water (70:30) +       -       37 °C       48 h       69.05 mg/mL       2001)         PBS       -       37 °C       48 h       69.05 mg/mL       2001)         PBS       -       37 °C       24 h       0.06 mg/mL       2006)         PBS/FG (50:50; w/w)       -       30 °C       24 h       1.80 mg/mL       2006)         PBS/FG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       Water         Transcutof <sup>®</sup> -       30 °C       48.72 h       0.43 mg/mL       Bonia et al., 1993)         Labrasit <sup>®</sup> -       30 °C       48.72 h       0.40 mg/mL       Igoral, 1993)         Labrasit <sup>®</sup> -       30 °C       48.72 h       1.3.20 mg/mL       Weter         Transcutof <sup>®</sup> -       30 °C       48.72 h       1.3.20 mg/mL </th <th>Ethanol/water (70:30) + oleic acid</th> <th>-</th> <th>27.00</th> <th>40.1</th> <th>(0<b>7</b>1 / I</th> <th></th>	Ethanol/water (70:30) + oleic acid	-	27.00	40.1	(0 <b>7</b> 1 / I		
Ethanol/water (70:30) + lauric       -       37 °C       48 h       66.35 mg/mL       2001)         acid 1% (w/v)       -       37 °C       48 h       64.11 mg/mL       -         Ethanol/water (70:30) + HPE-101       -       37 °C       48 h       69.05 mg/mL       -         PBS       -       3 °C       24 h       0.06 mg/mL       -       -         PBS       -       3 °C       24 h       0.06 mg/mL       2006)       -         PBS/Ethanol (50:50; w/w)       -       3 °C       24 h       1.85 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       24 h       1.88 mg/mL       (Mills, 2007)         PBS/PFG (50:50; w/w)       -       30 °C       48 -72 h       0.43 mg/mL       Mills, 2007)         PBS/PFG (50:50; w/w)       -       30 °C       48 -72 h       0.43 mg/mL       Mills, 2007)         PBS/PFG (50:50; w/w)       -       30 °C       48 -72 h       0.43 mg/mL       Holdon mg/mL       1993)         Liquid petrolatum       -       30 °C       48 -72 h       0.40 mg/mL       Honia et al.,         Labraol*       -       30 °C       48 -72 h       0.30 mg/mL       Weecks         PBS <td< th=""><th>1% (w/v)</th><th></th><th>3/°C</th><th>48 h</th><th>68./1 mg/mL</th><th>(Kim et al.,</th></td<>	1% (w/v)		3/°C	48 h	68./1 mg/mL	(Kim et al.,	
acid 1% (w/v)       37 °C       48 h       00.35 mg/mL         Ethanol/water (70:30) + HPE-101       -       37 °C       48 h       64.11 mg/mL         Ethanol/water (70:30) +       -       37 °C       48 h       69.05 mg/mL         PBS       -       3 °C       24 h       0.06 mg/mL       2006)         PBS/ FG (50:50; w/w)       -       3 °C       24 h       1.80 mg/mL       2006)         PBS/ FG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       2006)         PBS/ FG (50:50; w/w)       -       30 °C       24 h       1.48 mg/mL       2006)         PBS/ FG (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       48.18 mg/mL         PG       -       30 °C       48.72 h       0.43 mg/mL       1993)       14brafif*         Ldpatid petrolatum       -       30 °C       48.72 h       0.46.10 mg/mL       1993)       14brafif*         Labrafif*       -       30 °C       48.72 h       0.03 mg/mL       1993)       14brafif*         DPPG       -       30 °C       48.72 h       0.03 mg/mL       1993)       12brafif*         Labrafif*       -       30 °C       48.72 h       0.03 mg/mL	Ethanol/water (70:30) + lauric	-	27.00	40.1	(( )5	2001)	
Ethanol/water (70:30) + HPE-101       -       37 °C       48 h       64.11 mg/mL         1% (w/v)       -       37 °C       48 h       69.05 mg/mL         Ethanol/water (70:30) +       -       3 °C       24 h       0.06 mg/mL         PBS       -       3 °C       24 h       1.85 mg/mL       (Mills et al.,         PBS/F6thanol (50:50; w/w)       -       3 °C       24 h       1.82 mg/mL       2006)         PBS       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/F6thanol (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48-72 h       0.43 mg/mL       (Mills, 2007)         PBS       -       30 °C       48-72 h       0.43 mg/mL       (Hills, 2007)         PBS       -       30 °C       48-72 h       0.40 00 mg/mL       (Bonina et al.,         Labrasol®       -       30 °C       48-72 h       0.40 00 mg/mL       (Bonina et al.,         Labrasol®       -       30 °C       48-72 h       0.30 mg/mL       1993)         Labrasol®       -       30 °C       48-72 h       0.30 mg/mL       2012)         Wate	acid 1% (w/v)		3/ °C	48 n	66.35 mg/mL		
1% (w/v)       37 °C       48 h       69.11 mg/mL         Ethanol/water (70:30) +       -       37 °C       48 h       69.05 mg/mL         PBS       -       3 °C       24 h       0.06 mg/mL       (Mills et al., PBS/EG (50:50; w/w)         PBS/FG (50:50; w/w)       -       3 °C       24 h       1.50 mg/mL       2006)         PBS       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       24 h       1.48 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       (Mills, 2007)         PBS/FG (50:50; w/w)       -       30 °C       48.72 h       0.03 mg/mL       1993)         Laptasit <sup>®</sup> -       30 °C       48.72 h       40.00 mg/mL       1993)         Labrasit <sup>®</sup> -       30 °C       48.72 h       40.00 mg/mL       1993)         Labrasit <sup>®</sup> -       30 °C       48.72 h       0.30 mg/mL       1993)         Labrasit <sup>®</sup> -       32 °C       1 or 2       0.03 mg/mL       1993) <t< th=""><th>Ethanol/water (70:30) + HPE-101</th><th>-</th><th>27.90</th><th>40 1</th><th>64.11 m ~/m T</th><th></th></t<>	Ethanol/water (70:30) + HPE-101	-	27.90	40 1	64.11 m ~/m T		
Ethanol/water (70:30) + transcutol* 1% (wv)       -       37 °C       48 h       69.05 mg/mL         PBS       -       3 °C       24 h       0.06 mg/mL       2006)         PBS/PG (50:50; w/w)       -       3 °C       24 h       1.85 mg/mL       (Mills et al., 2006)         PBS/PG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       24 h       1.48 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48.72 h       0.40 mg/mL       (Bonia et al., 1.40 mg/mL         Vater       -       30 °C       48.72 h       0.40 mg/mL       1993)         Labra61*       -       30 °C       48.72 h       0.00 mg/mL       1993)         Labra61*       -       30 °C       48.72 h       10.00 mg/mL       1993)         Labra61*       -       30 °C       48.72 h       10.90 mg/mL       1993)         Labra61*       -       30 °C       48.72 h       13.20 mg/mL       20.20 mg/mL       20.20 mg/mL       20.20 mg/mL       20.20 mg/mL       20.20 mg/mL       2	1% (w/v)		3/ 0	48 n	64.11 mg/mL		
transcutol® 1% (w/v)         3/°C         48 n         0.96.0 mg/mL           PBS         -         3 °C         24 h         0.06 mg/mL           PBS/PC (50:50; w/w)         -         3 °C         24 h         1.85 mg/mL         (Mills et al., PBS/PC (50:50; w/w)           PBS         -         30 °C         24 h         0.06 mg/mL         2006)           PBS         -         30 °C         24 h         1.82 mg/mL         (Mills, 2007)           PBS/PC (50:50; w/w)         -         30 °C         24 h         1.48 mg/mL         (Mills, 2007)           PBS/PC (50:50; w/w)         -         30 °C         48.72 h         0.43 mg/mL         (Mills, 2007)           PG         -         30 °C         48.72 h         0.40 mg/mL         (Bonia et al., Labrasit*         -           Vater         -         30 °C         48.72 h         104.00 mg/mL         (Bonia et al., Labrasit*         -           DPPG         -         30 °C         48.72 h         10.40 mg/mL         (Bonia et al., Labrasit*           Vater         -         32 °C         1 or 2         0.03 mg/mL         -           Vater         -         32 °C         1 or 2         0.04 mg/mL         - <td< th=""><th>Ethanol/water (70:30) +</th><th>-</th><th>27 °C</th><th>18 h</th><th>60.05  mg/mI</th><th></th></td<>	Ethanol/water (70:30) +	-	27 °C	18 h	60.05  mg/mI		
PBS       -       3 °C       24 h       0.06 mg/mL         PBS/Ethanol (50:50; w/w)       -       3 °C       24 h       1.85 mg/mL       (Mills et al., 2006)         PBS       -       3 °C       24 h       1.85 mg/mL       2006)         PBS       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/EG (50:50; w/w)       -       30 °C       24 h       1.88 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       (Mills, 2007)         PG       -       30 °C       48.72 h       0.43 mg/mL       (Bonina et al., 1.45 mg/mL       1.48 mg/mL       (Bonina et al., 1.45 mg/mL       1.49 mg/mL       1.993)         Labrafil*       -       30 °C       48.72 h       0.40 mg/mL       1.993)       1.45 mg/mL       1.993)         Labrafil*       -       30 °C       48.72 h       13.20 mg/mL       1.993)	transcutol <sup>®</sup> 1% (w/v)		37 C	40 11	09.05 mg/mL		
PBS/Ethanol (50:50; w/w)       -       3 °C       24 h       1.85 mg/nL       (Mills et al., 2006)         PBS       -       3 °C       24 h       1.50 mg/nL       2006)         PBS       -       30 °C       24 h       1.50 mg/nL       2006)         PBS/Ethanol (50:50; w/w)       -       30 °C       24 h       1.48 mg/nL       (Mills, 2007)         PBS/PC (50:50; w/w)       -       30 °C       24 h       1.48 mg/nL       (Mills, 2007)         PBS/PC (50:50; w/w)       -       30 °C       48-72 h       0.43 mg/nL       (Mills, 2007)         PBS/PC (50:50; w/w)       -       30 °C       48-72 h       0.43 mg/nL       (Bonina et al., Labrasol*       140.00 mg/nL       (Bonina et al., 1993)         Labrasol*       -       30 °C       48-72 h       10.40 00 mg/nL       (Bonina et al., 2012)         Labrasol*       -       30 °C       48-72 h       13.20 mg/mL       (Bonina et al., 2012)         Labrasol*       -       32 °C       1 or 2       0.03 mg/mL       (Bonina et al., 2012)         Water       -       32 °C       1 or 2       0.04 mg/mL       2012)         Water       -       32 °C       1 or 2       8.08 mg/mL       2012)	PBS	-	3 °C	24 h	0.06 mg/mL		
PBS/PG (50:50; w/w)       -       3 °C       24 h       1.50 mg/mL       2006)         PBS       -       30 °C       24 h       0.06 mg/mL       (Mills, 2007)         PBS/Eftanol (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/Eftanol (50:50; w/w)       -       30 °C       48.72 h       0.43 mg/mL       (Mills, 2007)         PG       -       30 °C       48.72 h       0.43 mg/mL       (Bonina et al., 1.48 mg/mL       1993)         Labrasol*       -       30 °C       48.72 h       0.44.10 mg/mL       1993)         Labrasol*       -       30 °C       48.72 h       44.10 mg/mL       1993)         Labrasol*       -       30 °C       48.72 h       10.40.00 mg/mL       1993)         Labrasol*       -       30 °C       48.72 h       13.20 mg/mL       1993)         Labrasol*       -       30 °C       48.72 h       13.00 mg/mL       1993)         Labrasol*       -       32 °C       1 or 2       0.03 mg/mL       2012)         Weter       -       32 °C       1 or 2       8.08 mg/mL       2012)         Weeks       -       20 °C       48 h       0.34 mg/mL <td< th=""><th>PBS/Ethanol (50:50; w/w)</th><th>-</th><th>3 °C</th><th>24 h</th><th>1.85 mg/mL</th><th>(Mills et al.,</th></td<>	PBS/Ethanol (50:50; w/w)	-	3 °C	24 h	1.85 mg/mL	(Mills et al.,	
PBS       -       30 °C       24 h       0.06 mg/mL         PBS/Ethanol (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PG       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PG       -       30 °C       48-72 h       0.43 mg/mL       (Mills, 2007)         Transcutol*       -       30 °C       48-72 h       0.03 mg/mL       (Bonina et al.,         Labrafil*       -       30 °C       48-72 h       104.00 mg/mL       (Bonina et al.,         Labrafil*       -       30 °C       48-72 h       46.10 mg/mL       1993)         Labrafil*       -       30 °C       48-72 h       20.90 mg/mL       1993)         DPPG       -       30 °C       48-72 h       20.00 mg/mL       1993)         Labrafil*       -       32 °C       1 or 2       0.03 mg/mL       2012)         Water       -       32 °C       1 or 2       0.04 mg/mL       2012)         Wecks       -       -       32 °C       1 or 2       20.90 mg/mL       2012)         20% thydroethanolic solution       -       37 °C       48 h       0.34 mg/mL       2012)         20% t	PBS/ PG (50:50; w/w)	-	3 °C	24 h	1.50 mg/mL	2006)	
PBS/PG (50:50; w/w)       -       30 °C       24 h       1.82 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       24 h       1.48 mg/mL       (Mills, 2007)         PBS/PG (50:50; w/w)       -       30 °C       48-72 h       0.43 mg/mL       (Bonina et al., 1.43 mg/mL         Liquid petrolatum       -       30 °C       48-72 h       0.03 mg/mL       (Bonina et al., 1.43 mg/mL         Transcutol*       -       30 °C       48-72 h       104.00 mg/mL       (Bonina et al., 1.93 mg/mL         Labrasol*       -       30 °C       48-72 h       20.90 mg/mL       1993)         Labrasol*       -       30 °C       48-72 h       20.90 mg/mL       1993)         Labrasol*       -       30 °C       48-72 h       20.90 mg/mL       1993)         Labrasol*       -       30 °C       48-72 h       10.03 mg/mL       1993)         Labrasol*       -       32 °C       1 or 2       0.04 mg/mL       20.90 mg/mL         Vater       -       32 °C       1 or 2       8.08 mg/mL       2012)         weeks       -       20 °C       48 h       0.34 mg/mL       2012)         20% hydroethanolic solution       -       37 °C <t< th=""><th>PBS</th><th>-</th><th>30 °C</th><th>24 h</th><th>0.06 mg/mL</th><th></th></t<>	PBS	-	30 °C	24 h	0.06 mg/mL		
PBS/ PG (50:50; w/w)       -       30 °C       24 h       1.48 mg/mL         PG       -       30 °C       48-72 h       75.90 mg/mL         Liquid petrolatum       -       30 °C       48-72 h       0.43 mg/mL         Water       -       30 °C       48-72 h       0.43 mg/mL         Instruction       -       30 °C       48-72 h       104.00 mg/mL       (Bonina et al., 1993)         Labrafil®       -       30 °C       48-72 h       13.20 mg/mL       1993)         Labrafil®       -       30 °C       48-72 h       13.20 mg/mL       1993)         Water       -       32 °C       1 or 2       0.04 mg/mL       20.90 mg/mL         Water       -       32 °C       1 or 2       0.04 mg/mL       2012)         weeks       (Binks et al., weeks       (Binks et al., 2012)       2012)       weeks         20% hydroethanolic solution       -       20 °C       48 h       0.25 mg/mL       2012)         20% ethanol in PBS       -       20 °C       48 h       0.30 mg/mL       50 mg/mL       50 mg/mL         20% othanolic solution       -       20 °C       48 h       0.30 mg/mL       50 mg/mL       50 mg/mL       50 mg/mL       50 mg/m	PBS/Ethanol (50:50; w/w)	-	30 °C	24 h	1.82 mg/mL	(Mills, 2007)	
PG       -       30 °C       48.72 h       75.90 mg/mL         Liquid petrolatum       -       30 °C       48.72 h       0.43 mg/mL         Transcutol <sup>®</sup> -       30 °C       48.72 h       0.03 mg/mL       (Bonina et al., Labrasol <sup>®</sup> Transcutol <sup>®</sup> -       30 °C       48.72 h       46.10 mg/mL       1993)         Labrasol <sup>®</sup> -       30 °C       48.72 h       13.20 mg/mL       1993)         Labrafil <sup>®</sup> -       30 °C       48.72 h       13.20 mg/mL       1993)         DPPG       -       30 °C       48.72 h       13.20 mg/mL       1993)         Water       -       32 °C       1 or 2       0.03 mg/mL       2012)         Weeks       Weeks       (Binks et al., 2012)       weeks       (Binks et al., 2012)         20% hydroethanolic solution       -       20 °C       48 h       0.32 mg/mL       2012)         20% hydroethanolic solution       -       37 °C       48 h       0.34 mg/mL       2012)         20% bydroethanolic solution       -       37 °C       48 h       0.30 mg/mL       50% hydroethanolic solution       -       77 °C       48 h       0.06 mg/mL       50% hydroethanolic solution       -       77 °C <th>PBS/ PG (50:50; w/w)</th> <th>-</th> <th>30 °C</th> <th>24 h</th> <th>1.48 mg/mL</th> <th></th>	PBS/ PG (50:50; w/w)	-	30 °C	24 h	1.48 mg/mL		
Liquid petrolatum       -       30 °C       48-72 h       0.43 mg/mL         Water       -       30 °C       48-72 h       104.00 mg/mL       (Bonina et al., Labrasol*)         Labrasol*       -       30 °C       48-72 h       46.10 mg/mL       1993)         Labrasol*       -       30 °C       48-72 h       46.10 mg/mL       1993)         Labrasol*       -       30 °C       48-72 h       13.20 mg/mL         PPG       -       30 °C       48-72 h       13.20 mg/mL         Water       -       32 °C       1 or 2       0.03 mg/mL         weeks       weeks       (Binks et al., 2012)       2012)         weeks       -       32 °C       1 or 2       0.04 mg/mL       2012)         20% hydroethanolic solution       -       37 °C       48 h       0.34 mg/mL       2012)         20% ethanol in PBS       -       20 °C       48 h       0.34 mg/mL       48/mg/mL       48/mg/mL         20% othanolic solution       -       20 °C       48 h       0.30 mg/mL       48/mg/mL       48/mg/mL <th>PG</th> <th>-</th> <th>30 °C</th> <th>48-72 h</th> <th>75.90 mg/mL</th> <th></th>	PG	-	30 °C	48-72 h	75.90 mg/mL		
Water       -       30 °C       48.72 h       0.03 mg/mL         Transcutol®       -       30 °C       48.72 h       104.00 mg/mL       (Bonina et al., Labrasol®         Labrasol®       -       30 °C       48.72 h       104.00 mg/mL       1993)         Labrafil®       -       30 °C       48.72 h       13.20 mg/mL       1993)         DPPG       -       30 °C       48.72 h       13.20 mg/mL       1993)         Water       -       32 °C       1 or 2       0.03 mg/mL       weeks         PBS       -       32 °C       1 or 2       0.04 mg/mL       (Binks et al., 2012)         weeks       -       32 °C       1 or 2       8.08 mg/mL       2012)         weeks       -       20 °C       48 h       0.34 mg/mL       2012)         weeks       -       20 °C       48 h       0.34 mg/mL       2012)         weeks       -       20 °C       48 h       0.30 mg/mL       50% hydroethanolic solution       -       37 °C       48 h       0.30 mg/mL       50% hydroethanolic solution       -       20 °C       48 h       0.06 mg/mL       60.06 mg/mL       60.06 mg/mL       Touitou, 2005)       60.5% Brij® 98 aqueous solution       -       27 °C<	Liquid petrolatum	-	30 °C	48-72 h	0.43 mg/mL		
Transcutol <sup>®</sup> -       30 °C       48-72 h       104.00 mg/mL       (Bonina et al., 1993)         Labrafil <sup>®</sup> -       30 °C       48-72 h       46.10 mg/mL       1993)         Labrafil <sup>®</sup> -       30 °C       48-72 h       20.90 mg/mL       (Bonina et al., 1993)         Labrafil <sup>®</sup> -       30 °C       48-72 h       13.20 mg/mL       (Bonina et al., 1993)         Water       -       32 °C       1 or 2       0.03 mg/mL       (Binks et al., 2012)         Water       -       32 °C       1 or 2       0.04 mg/mL       (Binks et al., 2012)         Weeks       -       32 °C       1 or 2       8.08 mg/mL       2012)         Weeks       -       (Binks et al., 2012)       (Binks et al., 2012)       (Binks et al., 2012)         Weeks       -       -       32 °C       1 or 2       8.08 mg/mL       2012)         20% hydroethanolic solution       -       20 °C       48 h       0.34 mg/mL       Control (Alther et mg/mL)       (Ainbinder et motina)       -         20% hydroethanolic solution       -       20 °C       48 h       0.107 mg/mL       (Ainbinder et motina)       -         50% hydroethanolic solution       -       20 °C       48 h	Water	-	30 °C	48-72 h	0.03 mg/mL		
Labrasol <sup>®</sup> -       30 °C       48-72 h       46.10 mg/mL       1993)         Labrafil <sup>®</sup> -       30 °C       48-72 h       20.90 mg/mL         DPPG       -       30 °C       48-72 h       13.20 mg/mL         Water       -       32 °C       1 or 2       0.03 mg/mL         Weeks       -       32 °C       1 or 2       0.04 mg/mL         PBS       -       32 °C       1 or 2       0.04 mg/mL         Isopropyl myristate       -       32 °C       1 or 2       0.04 mg/mL         20% hydroethanolic solution       -       20 °C       48 h       0.25 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.30 mg/mL         20% ethanol in PBS       -       37 °C       48 h       0.30 mg/mL         20% ethanol in PBS       -       37 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       37 °C       48 h       0.17 mg/mL       Fourbould, 205)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       27 °C       -       6.00 mg/g       Edun	Transcutol <sup>®</sup>	-	30 °C	48-72 h	104.00 mg/mL	(Bonina et al.,	
Labrafil*       -       30 °C       48-72 h       20.90 mg/mL         DPPG       -       30 °C       48-72 h       13.20 mg/mL         Water       -       32 °C       1 or 2       0.03 mg/mL         weeks       -       32 °C       1 or 2       0.04 mg/mL         Water       -       32 °C       1 or 2       0.04 mg/mL         weeks       -       32 °C       1 or 2       0.04 mg/mL         weeks       -       32 °C       1 or 2       0.04 mg/mL         20% hydroethanolic solution       -       20 °C       48 h       0.25 mg/mL         20% hydroethanolic solution       -       20 °C       48 h       0.34 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.32 mg/mL         20% ethanol in PBS       -       37 °C       48 h       0.37 mg/mL         50% hydroethanolic solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij* 98 aqueous solution       -       37 °C       48 h       0.07 mg/mL       Fourier, 20 °C         6.0 % Brij* 98 aqueous solution       -       27 °C       -       6.00 mg/g       Fourier, 20 °C         6.0 % Brij* 98 aqueous soluti	Labrasol <sup>®</sup>	-	30 °C	48-72 h	46.10 mg/mL	1993)	
DPPG         -         30 °C         48-72 h         13.20 mg/mL           Water         -         32 °C         1 or 2         0.03 mg/mL           Weeks         -         32 °C         1 or 2         0.04 mg/mL           Weeks         -         32 °C         1 or 2         8.08 mg/mL         2012)           Weeks         -         32 °C         1 or 2         8.08 mg/mL         2012)           20% hydroethanolic solution         -         20 °C         48 h         0.25 mg/mL         2012)           20% ethanol in PBS         -         20 °C         48 h         0.30 mg/mL         2012)           20% ethanol in PBS         -         20 °C         48 h         0.30 mg/mL         2012)           20% ethanol in PBS         -         37 °C         48 h         0.30 mg/mL         20%           20% ethanol in PBS         -         37 °C         48 h         0.06 mg/mL         6.06 mg/mL         C           50% hydroethanolic solution         -         20 °C         48 h         0.08 mg/mL         6.00 mg/mL         1000000000000000000000000000000000000	Labrafil <sup>®</sup>	-	30 °C	48-72 h	20.90 mg/mL		
Water       -       32 °C       1 or 2       0.03 mg/mL weeks         PBS       -       32 °C       1 or 2       0.04 mg/mL weeks         Isopropyl myristate       -       32 °C       1 or 2       8.08 mg/mL 2012)         20% hydroethanolic solution       -       32 °C       48 h       0.25 mg/mL 2012)         20% hydroethanolic solution       -       37 °C       48 h       0.34 mg/mL 0.34 mg/mL 0.34 mg/mL 0.34 mg/mL 0.34 mg/mL 0.30 °C         20% ethanol in PBS       -       20 °C       48 h       0.30 mg/mL 0.30 mg/mL 0.30 °C 0.48 h       0.30 mg/mL 0.30 mg/mL 0.30 °C 0.48 h         50% hydroethanolic solution       -       20 °C 0.48 h       0.25 mg/mL 0.30 mg/mL 0.30 °C 0.48 h       0.06 mg/mL 0.30 °C 0.3	DPPG	-	30 °C	48-72 h	13.20 mg/mL		
PBS         -         32 °C         1 or 2 weeks         0.04 mg/mL weeks         (Binks et al., 2012)           1sopropyl myristate         -         32 °C         1 or 2 lor 2         8.08 mg/mL         2012)           20% hydroethanolic solution         -         20 °C         48 h         0.25 mg/mL         2012)           20% hydroethanolic solution         -         37 °C         48 h         0.34 mg/mL         2012)           20% ethanol in PBS         -         20 °C         48 h         0.30 mg/mL         50% hydroethanolic solution         -         20 °C         48 h         0.30 mg/mL         50% hydroethanolic solution         -         20 °C         48 h         0.30 mg/mL         -         Touitou, 2005)           50% hydroethanolic solution         -         20 °C         48 h         0.67 6 mg/mL         Touitou, 2005)           0.5% Brij <sup>®</sup> 98 aqueous solution         -         20 °C         48 h         0.06 mg/mL         Touitou, 2005)           0.5% Brij <sup>®</sup> 98 aqueous solution         -         27 °C         -         6.00 mg/g           Posiccated soybean oil         -         27 °C         -         8.60 mg/g         2005)           Hydrated olive oil         -         27 °C         -         8.00 mg/g	Water	-	32 °C	1 or 2	0.03 mg/mL		
PBS       -       32 °C       1 or 2       0.04 mg/mL weeks       (Binks et al., 2012)         Isopropyl myristate       -       32 °C       1 or 2       8.08 mg/mL       2012)         20% hydroethanolic solution       -       20 °C       48 h       0.25 mg/mL       2012)         20% hydroethanolic solution       -       20 °C       48 h       0.34 mg/mL       2012)         20% ethanol in PBS       -       20 °C       48 h       0.34 mg/mL       48 h       0.34 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.30 mg/mL       48 h       0.30 mg/mL         50% hydroethanolic solution       -       20 °C       48 h       0.30 mg/mL       48 h       0.30 mg/mL         50% hydroethanolic solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.19 mg/mL       Hydrated solbean oil         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       27 °C       -       6.00 mg/g       2005)         Hydrated soybean oil       -       27 °C       -				weeks			
weeks       (Binks et al., 2012)         weeks       (Binks et al., 2012)         weeks       20% hydroethanolic solution       -       20 °C       48 h       0.25 mg/mL       20% of C       48 h       0.25 mg/mL       20% of C       48 h       0.37 °C       48 h       0.30 mg/mL         20 °C       48 h       0.30 mg/mL         50% hydroethanolic solution       -       37 °C       48 h       0.06 mg/mL         50% brij® 98 aqueous solution       -       20 °C       48 h       0.01 mg/mL       Colicut colspan="2">Colspan="2">(Ainbinder e         50% brij® 98 aqueous solution       -       20 °C       48 h       0.06 mg/g         Hydrated soybean oil       -       27 °C <td colsp<="" th=""><th>PBS</th><th>-</th><th>32 °C</th><th>1 or 2</th><th>0.04 mg/mL</th><th></th></td>	<th>PBS</th> <th>-</th> <th>32 °C</th> <th>1 or 2</th> <th>0.04 mg/mL</th> <th></th>	PBS	-	32 °C	1 or 2	0.04 mg/mL	
Isopropyl myristate       - $32  ^{\circ}$ C       1 or 2 $8.08 \text{ mg/mL}$ $2012$ )         20% hydroethanolic solution       - $20  ^{\circ}$ C       48 h $0.25 \text{ mg/mL}$ 20% ethanol in PBS       - $20  ^{\circ}$ C       48 h $0.34 \text{ mg/mL}$ 20% ethanol in PBS       - $20  ^{\circ}$ C       48 h $0.32 \text{ mg/mL}$ 20% ethanol in PBS       - $37  ^{\circ}$ C       48 h $0.30 \text{ mg/mL}$ 50% hydroethanolic solution       - $20  ^{\circ}$ C       48 h $0.30 \text{ mg/mL}$ 50% hydroethanolic solution       - $20  ^{\circ}$ C       48 h $0.7  \text{mg/mL}$ (Ainbinder e $50  ^{\circ}$ hydroethanolic solution       - $37  ^{\circ}$ C       48 h $0.06  \text{mg/mL}$ Touitou, 2005) $0.5  ^{\circ}$ Brij $^{\circ}$ 98 aqueous solution       - $20  ^{\circ}$ C       48 h $0.17  \text{mg/mL}$ Touitou, 2005) $6.0  ^{\circ}$ Brij $^{\circ}$ 98 aqueous solution       - $27  ^{\circ}$ C       - $6.00  \text{mg/g}$ Desiccated soybean oil       - $27  ^{\circ}$ C       - $8.00  \text{mg/g}$ $2005$ )         Hydrated olive oil       - $27  ^{\circ}$ C       - $8.00  \text{mg/g}$ $2005$				weeks		(Binks et al.,	
weeks         20% hydroethanolic solution       -       20 °C       48 h       0.25 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.34 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.32 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.30 mg/mL         20% ethanol in PBS       -       20 °C       48 h       0.30 mg/mL         50% hydroethanolic solution       -       37 °C       48 h       0.30 mg/mL         50% hydroethanolic solution       -       37 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       27 °C       48 h       0.17 mg/mL       (Ainbinder e         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       27 °C       48 h       0.19 mg/mL       (Land et al.,         Hydrated soybean oil       -       27 °C       -       6.00 mg/g       2005)         Hydrated olive oil       -       27 °C       -       8.00 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -	Isopropyl myristate	-	32 °C	1 or 2	8.08 mg/mL	2012)	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$				weeks			
20% hydroethanolic solution       - $37  ^{\circ}$ C       48 h       0.34 mg/mL         20% ethanol in PBS       - $20  ^{\circ}$ C       48 h       0.30 mg/mL         20% ethanol in PBS       - $37  ^{\circ}$ C       48 h       0.30 mg/mL         50% hydroethanolic solution       - $20  ^{\circ}$ C       48 h       0.30 mg/mL         50% hydroethanolic solution       - $20  ^{\circ}$ C       48 h       0.77 mg/mL       (Ainbinder e         50% hydroethanolic solution       - $37  ^{\circ}$ C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij® 98 aqueous solution       - $20  ^{\circ}$ C       48 h       0.08 mg/mL       Touitou, 2005)         0.5% Brij® 98 aqueous solution       - $37  ^{\circ}$ C       48 h       0.17 mg/mL       Touitou, 2005)         0.5% Brij® 98 aqueous solution       - $27  ^{\circ}$ C       - $6.00  \text{mg/mL}$ -         6.0 % Brij® 98 aqueous solution       - $27  ^{\circ}$ C       - $6.00  \text{mg/mL}$ -         Hydrated soybean oil       - $27  ^{\circ}$ C       - $8.60  \text{mg/g}$ 2005)         Hydrated olive oil       - $27  ^{\circ}$ C       - $8.00  \text{mg/g}$ 2005)	20% hydroethanolic solution	-	20 °C	48 h	0.25 mg/mL		
20% ethanol in PBS       -       20 °C       48 h       0.22 mg/mL         20% ethanol in PBS       -       37 °C       48 h       0.30 mg/mL         50% hydroethanolic solution       -       20 °C       48 h       0.30 mg/mL         50% hydroethanolic solution       -       20 °C       48 h       6.76 mg/mL       (Ainbinder e         50% hydroethanolic solution       -       37 °C       48 h       0.06 mg/mL       Touitou, 2005)         50% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.17 mg/mL       (Ainbinder e         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.17 mg/mL       -         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       27 °C       -       6.00 mg/g       -         Posiccated soybean oil       -       27 °C       -       8.60 mg/g       2005)         Hydrated olive oil       -       27 °C       -       8.90 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -       8.90 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -	20% hydroethanolic solution	-	37 °C	48 h	0.34 mg/mL		
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50% hydroethanolic solution       -       20 °C       48 h       6.76 mg/mL         50% hydroethanolic solution       -       37 °C       48 h       10.77 mg/mL       (Ainbinder e Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.06 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       37 °C       48 h       0.08 mg/mL       Touitou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.17 mg/mL       Touitou, 2005)         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.19 mg/mL       -         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.19 mg/mL       -         Hydrated soybean oil       -       27 °C       -       6.00 mg/g       -         Desiccated soybean oil       -       27 °C       -       8.60 mg/g       2005)         Hydrated olive oil       -       27 °C       -       8.00 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -       8.90 mg/g       2005)         Braiceated miglyol 812 oil       -       27 °C       -       14.30 mg/g       2005)         Kater	20% ethanol in PBS	-	37 °C	48 h	0.30 mg/mL		
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0.5% Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.06 mg/mL       1 outrou, 2005)         0.5% Brij <sup>®</sup> 98 aqueous solution       -       37 °C       48 h       0.08 mg/mL         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       20 °C       48 h       0.17 mg/mL         6.0 % Brij <sup>®</sup> 98 aqueous solution       -       37 °C       48 h       0.19 mg/mL         Hydrated soybean oil       -       27 °C       -       6.00 mg/g         Desiccated soybean oil       -       27 °C       -       8.60 mg/g         Hydrated olive oil       -       27 °C       -       8.00 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -       8.90 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -       8.90 mg/g       2005)         Hydrated miglyol 812 oil       -       27 °C       -       14.30 mg/g       -         Desiccated miglyol 812 oil       -       27 °C       -       14.30 mg/g       -         Lethanol       -       25 °C       One whole       209.99 mg/mL       -         day and       -       25 °C       One whole       0.02 mg/mL       2005)         day and	50% hydroethanolic solution	-	37 °C	48 h	10.77 mg/mL	(Ainbinder e	
0.5% Brij <sup>®</sup> 98 aqueous solution       - $37  ^{\circ}$ C       48 h $0.08 \text{ mg/mL}$ 6.0 % Brij <sup>®</sup> 98 aqueous solution       - $20  ^{\circ}$ C       48 h $0.17 \text{ mg/mL}$ 6.0 % Brij <sup>®</sup> 98 aqueous solution       - $37  ^{\circ}$ C       48 h $0.19 \text{ mg/mL}$ Hydrated soybean oil       - $37  ^{\circ}$ C       - $6.00 \text{ mg/g}$ Desiccated soybean oil       - $27  ^{\circ}$ C       - $8.60 \text{ mg/g}$ Hydrated olive oil       - $27  ^{\circ}$ C       - $8.60 \text{ mg/g}$ Desiccated olive oil       - $27  ^{\circ}$ C       - $8.00 \text{ mg/g}$ $2005$ )         Hydrated miglyol 812 oil       - $27  ^{\circ}$ C       - $8.90 \text{ mg/g}$ $2005$ )         Desiccated miglyol 812 oil       - $27  ^{\circ}$ C       - $8.90 \text{ mg/g}$ $2005$ )         Ethanol       - $27  ^{\circ}$ C       - $14.30 \text{ mg/g}$ $43 \text{ and}$ night       (Araya et al., 2005)         Water       - $25  ^{\circ}$ C       One whole $0.02 \text{ mg/mL}$ $2005$ )	0.5% Brij <sup>®</sup> 98 aqueous solution	-	20 °C	48 h	0.06 mg/mL	Touitou, 2005)	
6.0 % Brij <sup>®</sup> 98 aqueous solution       - $20  ^{\circ}\text{C}$ $48  \text{h}$ $0.17  \text{mg/mL}$ 6.0 % Brij <sup>®</sup> 98 aqueous solution       - $37  ^{\circ}\text{C}$ $48  \text{h}$ $0.19  \text{mg/mL}$ Hydrated soybean oil       - $27  ^{\circ}\text{C}$ - $6.00  \text{mg/g}$ Desiccated soybean oil       - $27  ^{\circ}\text{C}$ - $8.60  \text{mg/g}$ Hydrated olive oil       - $27  ^{\circ}\text{C}$ - $8.60  \text{mg/g}$ Desiccated olive oil       - $27  ^{\circ}\text{C}$ - $8.00  \text{mg/g}$ Desiccated olive oil       - $27  ^{\circ}\text{C}$ - $8.00  \text{mg/g}$ $2005$ )         Hydrated miglyol 812 oil       - $27  ^{\circ}\text{C}$ - $8.90  \text{mg/g}$ $2005$ )         Desiccated miglyol 812 oil       - $27  ^{\circ}\text{C}$ - $14.30  \text{mg/g}$ $-$ Ethanol       - $25  ^{\circ}\text{C}$ One whole $209.99  \text{mg/mL}$ $2005$ )         Water       - $25  ^{\circ}\text{C}$ One whole $0.02  \text{mg/mL}$ $2005$ )	0.5% Brij <sup>®</sup> 98 aqueous solution	-	37 °C	48 h	0.08 mg/mL		
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Desiccated soybean oil- $27 \ ^{\circ}C$ - $8.60 \ mg/g$ Hydrated olive oil- $27 \ ^{\circ}C$ - $5.20 \ mg/g$ (Land et al.,Desiccated olive oil- $27 \ ^{\circ}C$ - $8.00 \ mg/g$ $2005$ )Hydrated miglyol 812 oil- $27 \ ^{\circ}C$ - $8.90 \ mg/g$ Desiccated miglyol 812 oil- $27 \ ^{\circ}C$ - $8.90 \ mg/g$ Desiccated miglyol 812 oil- $27 \ ^{\circ}C$ - $14.30 \ mg/g$ Ethanol- $25 \ ^{\circ}C$ One whole $209.99 \ mg/mL$ Water- $25 \ ^{\circ}C$ One whole $0.02 \ mg/mL$ $2005$ )	Hydrated soybean oil	-	27 °C	-	6.00 mg/g		
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Desiccated olive oil- $27 \ ^{\circ}C$ - $8.00 \ mg/g$ $2005$ )Hydrated miglyol 812 oil- $27 \ ^{\circ}C$ - $8.90 \ mg/g$ Desiccated miglyol 812 oil- $27 \ ^{\circ}C$ - $14.30 \ mg/g$ - $25 \ ^{\circ}C$ One whole $209.99 \ mg/mL$ Ethanolinght(Araya et al., 2005)Water- $25 \ ^{\circ}C$ One whole $0.02 \ mg/mL$ $2005$ )	Hydrated olive oil	-	27 °C	-	5.20 mg/g	(Land et al.,	
Hydrated miglyol 812 oil       -       27 °C       -       8.90 mg/g         Desiccated miglyol 812 oil       -       27 °C       -       14.30 mg/g         -       25 °C       One whole       209.99 mg/mL         day and       -       ight       (Araya et al., 2005)         Water       -       25 °C       One whole       0.02 mg/mL	Desiccated olive oil	-	27 °C	-	8.00 mg/g	2005)	
Desiccated miglyol 812 oil         -         27 °C         -         14.30 mg/g           -         25 °C         One whole         209.99 mg/mL           day and         night         (Araya et al., 2005)           Water         -         25 °C         One whole         0.02 mg/mL	Hydrated miglyol 812 oil	-	27 °C	-	8.90 mg/g		
-       25 °C       One whole 209.99 mg/mL day and night         Water       -       25 °C       One whole 0.02 mg/mL 2005) day and	Desiccated miglyol 812 oil	-	27 °C	-	14.30 mg/g		
Ethanol       day and night       (Araya et al., 25 °C         Water       -       25 °C       One whole 0.02 mg/mL 2005) day and       2005)		-	25 °C	One whole	209.99 mg/mL		
waternight(Araya et al., 25 °COne whole0.02 mg/mL2005) day and	Ethanol			day and		<i>.</i>	
Water-25 °COne whole0.02 mg/mL2005)day and				night		(Araya et al.,	
day and	Water	-	25 °C	One whole	0.02 mg/mL	2005)	
				day and			

			night		
PBS	-	30 °C	24 h	0.07 mg/mL	
Ethanol/PBS (50:50; w/w)	-	30 °C	24 h	6.31 mg/mL	(Mills, 2007)
PG/PBS (50:50; w/w)	-	30 °C	24 h	5.14 mg/mL	

HSA = human serum albumin ; β-CD = β-cyclodextrin; HP-β-CD = hydroxypropyl-β-cyclodextrin; SBE7-β-CD = sulfobutyl-βcyclodextrin; HSES = heptakis-6-sulfoethylsulfanyl6-deoxy- β-cyclodextrin; HTMT = eptakis-6-methylsulfanyl-6-deoxy-2-(2-(2-(2-methoxy)ethoxy)ethoxy)ethyl)]-β-cyclodextrin; HTG = heptakis-6-thioglyceryl-6-deoxy-β-cyclodextrin; LBS = glycolysed ethoxylated C<sub>8</sub>/C<sub>10</sub> glycerides; LBF = glycolysed ethoxylated glycerides; TSC = diethylene glycol monoethyl ether; DPPG = propylene glycol dipelargonate; PBS = phosphate-buffered saline; PG = propylene glycol.

## **3. RECEPTOR PHASE SELECTION**

The diffusion cell apparatus has been widely used for percutaneous permeation studies of drugs and it may be categorized into static or flow-through cells. In both classes, formulations are applied to the surface of a membrane, which is sandwiched between a donor and a receiver compartment of the diffusion cell (Moser et al., 2001). In static diffusion cells, samples are collected and the same volume of fresh perfusate is added at each time point (Bronaugh e Stewart, 1985). In these systems, the membrane, donor and receiver chambers may be placed either "vertically" as in the popular Franz diffusion cell or "horizontally" (Zsikó et al., 2019). In flow-through diffusion cells, a pump continuously supplies the receiver compartment with perfusate, simulating the blood flow from the dermis (Bronaugh e Stewart, 1985). When more physiologically relevant assessments of percutaneous permeation for lipophilic compounds are needed, flow-through diffusion cells should be prioritized (Clowes et al., 1994).

In static diffusion cells (most used in skin permeation studies), the volume of diffusion cell receiver chamber should be selected to guarantee detectable concentrations of the permeant in the receiver medium. In this context, the receptor phase should present a reduced volume for the evaluation of drugs with low permeation rate as TST. Likewise, the smaller the amount of TST added in the donor phase, the smaller volumes of receiver medium are recommended. The system agitation should be enough to obtain a homogenous distribution of the permeant and temperature equilibrium in this medium.

The composition of receiver phase should ensure that "sink conditions" may be achieved and barrier properties of biological tissues are preserved (Cilurzo et al., 2018; Ng et al., 2010). In an acceptable sink condition, the maximum concentration of drug in the receptor fluid in the permeation assay should not exceed 10% of its maximum solubility (EMA, 2012). It is calculated by the relationship between CS and CD ("CS"

is the saturated solubility of the compound in the medium whereas "CD" represents the concentration of compound in the bulk medium).

Isotonic saline or buffered isotonic saline (pH=7.4) are often used for highly soluble drugs to mimic the physiological environment; however, solubilizer agents need to be included when poorly water-soluble drugs are considered (Finnin et al., 2012). Azone<sup>®</sup> and *N*-methyl-2-pyrrolidone, for example, increased the hormone solubility from 0.02 mg/mL (in water) to 92.29 mg/mL and 518.89 mg/mL, respectively (Table 1). Although the solubilization of TST by these agents allows sink conditions to be achieved, the impact on the skin structure discourages their application (Fang et al., 2003). Thus, aqueous solutions containing less aggressive agents such as ethanolic solutions, bovine serum albumin or propylene glycol should be alternatively considered for the composition of receptor fluid (Table 2). Sodium azide was also added to the receptor solution in some studies with an experimental time of 24 h (Table 2). This compound acts as a preservative and is therefore recommended for long-term permeation assays (Bartosova e Bajgar, 2012).

### 4. EFFECT OF DRUG CONCENTRATION FROM DONOR PHASE

According to Fick's 1st law (Equation 1), the permeation flux of a drug is proportional to its concentration in the vehicle (or thermodynamic activity,  $A_v$ ), occurring in favor of a concentration gradient. High drug solubilization leads to a high thermodynamic activity in the donor phase, improving its permeation flux through the SC (Ceschel et al., 2005).

$$J = \frac{K.C_0.D}{L} = \frac{\gamma_v.C_0.D}{\gamma_s.L} = \frac{A_v.D}{\gamma_s.L}$$
(Equation 1)

Where: *K*, *C*<sub>0</sub>, *D*,  $\gamma_{\nu}$  and  $\gamma_{s}$  represent the partition coefficient, initial concentration in the formulation, diffusion coefficient, activity coefficient in the formulation and that from the skin barrier, respectively. *L* is the tissue thickness (Ishii et al., 2010).

The maximum permeation flux  $(J_{max})$ , in turn, is achieved at the maximum solubility  $(S_s)$  of a solute in the SC (Couto et al., 2014), which also corresponds when solute solubility in vehicle achieves its saturation condition.

$$J_{max} = \frac{D.S_s}{L}$$
 (Equation 2)

In general, the permeation profile of TST through the skin follows a Fickian diffusion given the absorbed drug fraction or rate in specific period is constant. When the diffusion of the TST in the skin is analyzed, its transfer rate to the SC is faster whereas a slower rate from the dermis to the systemic route is showed. As already discussed, this fact is attributed to reservoir effect of TST in tissue. The greater the interaction or partitioning of the drug with the tissue, the slower the diffusion rate.

The fluid composition of the donor phase appears to impact on the permeation rate more significantly than the TST concentration (Table 2). In a study that considered 0.22, 6.31 and 5.14 mM of TST in only PBS, ethanol/PBS and propylene glycol/PBS in the donor phase of diffusion cells, permeation coefficient  $(K_p)$  values from equine skin were 6.82, 1.59 and 2.04 x 10<sup>-3</sup> cm/h, respectively (Mills, 2007). Although the donor phase containing only PBS had a lower drug concentration, it provided a higher K<sub>p</sub> value. These findings may be associated with solubility differences of TST in the donor phase, which affect the supersaturation degree (greater in aqueous solutions). In another study considering a similar donor phase composition (solubilizer type and ratio as well as TST concentration) and canine skin as membrane model, donor solutions without solubilizer agent and with lower TST concentration also provided higher K<sub>p</sub> values (Mills et al., 2006). In the same way, in a permeation study carried out with different concentrations of ethanol in water (20, 40, 50, 60, 70, 80 and 100 %) and TST (0.45, 8.53, 18.52, 38.95, 68.32, 140.01 and 33.04 mg mL<sup>-1</sup>), lower K<sub>p</sub> values through rat skin were found for increasing amounts of ethanol and TST (Kim et al., 2001). In these analyzed studies, the inclusion of propylene glycol seems to be more advantageous than ethanol in increasing the cutaneous transport of TST.

In permeation studies, authors commonly mention that an increased thermodynamic activity provided by a state of supersaturation results in higher permeation rate (Schwarb et al., 1999), however, it is necessary to be careful with this assumption. In a study comparing the permeation rate of TST from saturated solution and suspensions presenting different supersaturation degrees (1.4, 2.1 and 2.6), only the suspension with a supersaturation degree of 1.4 provided a higher flux permeation than control. For the other conditions, a permeation flux similar to the saturated solution was found because drug precipitation phenomena would be observed, reducing the free drug amount available to be absorbed (M.-L. Leichtnam et al., 2006) Considering these findings, it is important to consider the inclusion of anti-precipitating agents in tested

medium so that the advantages resulting from the increase in the degree of supersaturation can be observed. This may also explain why membrane-based systems (e.g., polymeric films or transdermal patches) can provide higher rates of cutaneous permeation compared to the solution or other liquid systems in specific situations.

## 5. IMPACT OF TISSUE PREPARATION ON DRUG PERMEATION

Permeation assays may be performed by using epidermal membranes, dermatomed skin, or full-thickness skin. The use of epidermis provides a greater correlation with *in vivo* situation (in humans). The presence of dermis in dermatomed skin and full thickness skin acts as an unreal barrier since continuous blood flow occurs within watery dermis only in *in vivo* environment. Although the use of the epidermis provides more realistic information on drug transport (Barbero e Frasch, 2016), the tissue preparation is more difficult and susceptible to structural damages. Currently available methods to separate the epidermis from the dermis use heating or chemical treatment (e.g. treatment with salts, detergents, enzymes) (Abd et al., 2016), which can affect both tissue integrity and skin metabolic activity. The tissue treatment with trypsin, for example, may lead to a SC more heterogeneous, contributing to the partitioning of more polar solutes (Magnusson et al., 2006). Another disadvantage is that hair follicles may be damaged during the tissue separation process, leading to drug leakage (Barbero e Frasch, 2016).

The transappendageal route is preferentially used to transport hydrophilic drug, high molecular weight compounds and drug delivery systems as nanoparticles (Knorr et al., 2009); however, Hueber et al. (1992) has also demonstrated its contribution in transport of TST (via sebaceous glands). In another study performed by the same research team (Hueber et al., 1994), the absorption of TST from the human normal skin was approximately 2.4-fold higher than in human scar skin (without skin appendages) at the end of 8 h. These findings showed the importance of preserving the appendages of the skin in permeation studies.

Transfollicular drug delivery may be evaluated by comparing different body regions, use of specific animal models, artificial introduction of hair follicles in skin equivalents and/or blockage of hair follicles compared to untreated skin (Knor et al., 2009). Porcine skin is regarded as the most representative animal model to determine the contribution of the follicular route in drug transport due to the similarity to human

skin. In fact, porcine and human skin present about 20-30 follicles per cm<sup>2</sup> of skin area, and a hair density of 11-25 hairs/cm<sup>2</sup> with a diameter of 58-97  $\mu$ m. Animal models such as rat and mouse skin exhibit much more follicles, with smaller diameters when compared to human skin (Lauterbach e Müller-Goymann, 2015). Follicle-free skin models as EpidermFT<sup>TM</sup> (for comparison with human skin presenting follicles) and *in vitro* models of human fibroblasts or keratinocytes with hair follicles may represent alternatives to animal use in these studies (Krugluger et al., 2005; Michel et al., 1999).

"Full-thickness skin" is prepared by removal of connective tissue and subcutaneous fat and its thickness may be reduced with a dermatome. This procedure reduces experimental variability (Abd et al., 2016). Although dermatomed and full-thickness skin present an additional barrier that is not found *in vivo*, particularly for lipophilic drugs, the tissue integrity is preserved (Barbero e Frasch, 2016). As a result, the absorption rate of drugs could be underestimated. The presence of this additional layer represents a situation analogous to vasoconstriction (Abd et al., 2016).

Interestingly, a study testing only the dermis layer as membrane model and TST was also performed (Kretsos et al., 2008). The permeation coefficient value found from this study showed to be higher than other studies performed with full-thickness or dermatomed skin as well as only with human epidermis (Baert et al., 2012; Guth et al., 2015; Hewitt et al., 2020; J R Heylings et al., 2018; Kretsos et al., 2008; Netzlaff et al., 2006; Qvist et al., 2000; Schreiber et al., 2005; Veryser et al., 2015), confirming that both SC and viable epidermis play key barrier properties.

A significant difference in permeation flux of TST through human dermatomed skin (most of the dermis removed) and full-thickness skin was observed, with a TST flux of approximately 10-fold higher in the thinner skin (Wilkinson et al., 2006). Thus, membrane thickness should be a critical variable for TST permeation across human skin. The intra- and inter-laboratory study between 10 laboratories found higher flux rates of TST using skin samples presenting thickness between 300 and 500  $\mu$ m (2.82 to 5.39  $\mu$ g.cm<sup>-2</sup> .h<sup>-1</sup>) compared to skin samples with thickness between 700 and 900  $\mu$ m (0.40 to 0.80  $\mu$ g.cm<sup>-2</sup> .h<sup>-1</sup>) (van de Sandt et al., 2004). This finding is consistent with the lipophilic nature of TST, which difficult its transport through the dermis. Therefore, permeation results obtained for lipophilic agents after the transport in full-thickness or dermatomed tissue preparations should be carefully analyzed (Magnusson et al., 2006).

The use of epidermal membranes, in turn, may overestimate the drug absorption in humans because of insufficient barrier function. The use of cultured and reconstructed human skin models (e.g. constructed from keratinocytes) is not recommended for the determination of dermal penetration as these models have not been validated for dermal absorption studies and there are reports that their barrier properties are not comparable with those of 'natural origin' skin (SCCS, 2010; WHO, 2006).

The OECD recommends the use of dermatomed skin with a thickness between 200 and 400 µm for harmonization of ex vivo studies with human skin. In these range of thickness, the membranes tend to have significantly lower levels of residual material than full-thickness preparations. For permeation assays with TST (Table 1), dermatomed and full-thickness skin models appears in most of these studies. The use of membranes with 400 µm of thickness was prevalent among the studies with human skin, even though, different values of permeated drug and rate flux were showed. Higher permeability coefficient values (K<sub>p</sub>) have been reported with the use of only epidermis compared to dermatomed skin as already mentioned. Correlations can be established by regarding studies performed in similar experimental conditions (donor and receptor fluid). Netzlaff et al. (2006) and Schreiber et al. (2005), for instance, found  $K_p$  values of 9.4 x 10<sup>-4</sup> and 8.3 x 10<sup>-4</sup> cm.h<sup>-1</sup>, respectively, with human epidermis separated by heat. In contrast, lower  $K_p$  values, 4.5 x 10<sup>-4</sup> and 3.92 x 10<sup>-4</sup> cm.h<sup>-1</sup>, were showed by Baert et al. (2012) and Veryser et al. (2015), respectively, using skin samples with a thickness of 400 µm. In fact, the tissue preparation has a significant effect on the TST permeation. The studies showed an increase of about 2-fold in permeation rate of the hormone without dermis layer.

The tissue storage conditions also impact on drug permeation. As the skin preparation step is laborious, the tissue is commonly frozen for a specific period before the use in the permeation assays. The freezing does not affect the skin barrier integrity and it is a suitable procedure to measure the passive permeation of drugs (Barbero e Frasch, 2016). One the other hand, it is worth mentioning that frozen tissue is not appropriated to investigate the metabolic activity of drugs (Fahmy et al., 1993).

In general, the permeation assays with TST were performed using similar storage conditions. The tissues were frozen at -20 °C in all studies with human skin, which is according with regulatory agencies guidelines (OECD, EU Scientific

Committee on Consumer Products, US Environmental Protection Agency, International Programme on Chemical Safety). Tissue preparation for permeation studies was performed prior to freezing. The human skin storage time, in turn, often ranged from 3 (Baert et al., 2012; Hewitt et al., 2020) or 6 months (Schreiber et al., 2005; Veryser et al., 2015). A consensus among regulatory agencies regarding the freezing time of tissues prior the use has not yet been reached. For example, the International Programme on Chemical Safety (IPCS) states that human skin can be stored to one year, whereas the US Environmental Protection Agency (EPA) allows the storage for up to 3 months. The skin thawing should also be standardized in assays since the use of frozen skin without hydration provides different permeation rate values when compared to fresh tissue. Thus, the tissues should be appropriately rehydrated before use (Swarbrick et al., 1982).

## 6. SELECTION OF ANIMAL MODELS OR MEMBRANE TYPE

Human skin is the preferred membrane model to predict the *ex vivo* permeation of compounds. Excised skin is commonly obtained from autopsies (cadaver skin) or plastic surgery. Overall, the permeation assays of TST were performed with tissues from abdomen and breast. Several animal models have been alternatively considered to human skin, including pig, mouse, rat, guinea pig and rabbit skin. Porcine skin has been widely used given its histological similarity to human skin. It presents a comparable SC thickness, which varies from 21 to 26  $\mu$ m, and similar hair-follicle density (20 vs. 14-32/cm<sup>2</sup> for porcine ear skin and human forehead skin, respectively) (Jacobi et al., 2007). Porcine SC lipids are organized as a hexagonal lattice whereas human SC lipids are arranged in the denser orthorhombic lattice (Caussin et al., 2008).

The easy availability and relatively low cost make the group of rodents (mice, rat, and guinea pigs) widely used in *ex vivo* percutaneous permeation studies. Although rat skin is most structurally similar to human skin among the rodents, rat dermis is thicker than human and does not present a defined limit between papillary and reticular dermis. In addition, rats have no subcutaneous fatty tissue or subcutaneous muscle tissue (McFarlane et al., 1965). In terms of permeability, significant differences between rat and human skin have been found for various compounds presenting different physicochemical properties (Abd et al., 2016). *Ex vivo* permeation studies with saturated and supersaturated solutions of TST applied in rat skin have been performed (Kim et al.,

1999; Leichtnam et al., 2006). Overall, higher values of flux and permeation coefficient were found with rat skin compared with human skin (Table 1). Nevertheless, the rat is also often used for *in vivo* studies because the extensive pharmacokinetics/ pharmacodynamics data that may be obtained with this species.

Netzlaff et al. (2006) performed *ex vivo* permeation studies with TST and skin models from different species. The permeation coefficient values of TST through bovine udder, human and pig skin were 5.42, 2.31 and 1.29 (x  $10^{-7}$ ) cm.s<sup>-1</sup>, respectively. Differences in skin composition explain these findings. Human and pig skin are characterized by a much higher lipid fraction (triglycerides and free fatty acids) than udder skin, which increase the affinity of TST by the tissue. Although these tissues significantly differ in relation to the number of follicles per tissue area (6, 30-36 and 207-338 mm.cm<sup>-2</sup> for human, pig and bovine udder skin, respectively), this permeation route is not used for lipophilic molecules such as TST. The epidermal thickness also varied with the animal species (57-82, 40 and 54-92 µm for human, pig and bovine udder skin, respectively), but this parameter would present a marginal contribution on permeation rate of TST.

TST has been used as lipophilic molecule model in several percutaneous permeation studies. When the different studies are compared (Table 2), mice, amphibian and equine skin seem to be more permeable to TST than human or porcine skin ( $K_p$  values increased from 10 to 100 times). In fact, amphibian skin, for example, presents only 1-2 cell layers of stratum corneum with no intercellular lipid layers, offering less resistance to TST transport (Lillywhite, 2006). In the same way, mice also have a stratum corneum that is thinner than humans (5 vs. 10-20 µm) (Wei et al., 2017), which may explain the greater permeation rate.

As amphibian skin is much more permeable than other vertebrates and more sensitive to environmental contaminants, this skin model has not only been considered to assess the absorption of molecules but also as an indicator of the relative health of an ecosystem (Llewelyn et al., 2019). Kaufmann and Dohmen (2016) performed permeation assays with TST in the African clawed frog skin (X. laevis, wild type). The permeation coefficient values ranged from 1.3 to 3.0 x  $10^{-3}$  cm.h<sup>-1</sup>. Regarding the same composition of donor and receptor fluid, Baert et al. (2012) evaluated the TST permeation across human skin and found a permeation coefficient of 4.5 x  $10^{-4}$  cm.h<sup>-1</sup>.

Therefore, the permeation of TST through amphibian skin was 2.9 to 6.7-fold higher than human skin.

*In vitro* permeation studies in equines and canines have also been performed (Mills, 2007; Mills et al., 2006) since the skin is an administration route used for the TST in these animals. The main clinical use of TST in horses aims to increase libido and treat hypogonadism (Snow, 1993). Its application to improve the performance of animals in competitions is illegal (Houghton e Maynard, 2010). In dogs, the hormone is particularly used for treating TST-responsive urinary incontinence and for suppression of oestrus in racing greyhounds (Blythe et al., 1994; Plumb, 2002).

Synthetic membranes are structurally simpler and demonstrate superior permeation data reproducibility since *in vivo* variables as skin age, race, sex and anatomical site are eliminated. In contrast, they do not exhibit the lipid perturbation effects undergone by biological samples (Ng et al., 2010). For the TST, all synthetic silicone-based membranes demonstrated lower barrier properties and thus greater permeation compared to biological tissues. The fact that synthetic membranes do not have receptors for TST, and present greater permeability may raise the hypothesis of a probable relationship between physiological receptors and the reservoir effect observed for this hormone; however, further investigations should be performed.

Reconstructed skin models are interesting substitutes for human and animal skin because they overcome the problems of availability of human skin and the ethical issues of using animals. On the other hand, the OECD has not yet approved the use of these models for skin absorption assays (only for *in vitro* skin irritation assays). Therefore, more studies are needed to validate the models for this purpose. In addition, reconstructed skin is usually more permeable than human skin. The low barrier property of reconstructed skin was demonstrated by Schreiber et al. (2005). The authors founded higher permeation coefficient values for TST using reconstructed epidermis ( $K_p$ = 176.4 x 10<sup>-5</sup> and 777.6 x 10<sup>-5</sup> cm.h<sup>-1</sup>) compared to human epidermis ( $K_p$ = 7.6 x 10<sup>-5</sup> cm.h<sup>-1</sup>) and porcine skin ( $K_p$ = 31.7 x 10<sup>-5</sup> cm.h<sup>-1</sup>). The assays with porcine skin and human epidermis were performed in 24 h whereas the use of reconstructed skin reduced this time to 6 h. The short lag time and high flux values are associated with the reduction of experimental time. For this reason, the reconstructed skin model has been considered advantageous for routine and screening assays, particularly when several compounds or topical formulations should be analyzed at the same time.

The efforts to improve the barrier function of these models is constant. Simard et al. (2019), for instance, evaluated the effect of different lipids on skin barrier properties by considering a 3D reconstructed skin model. A slower permeation rate of TST was found after supplementation with alpha-linolenic acid. On the other hand, the supplementation with linoleic acid did not affect the drug absorption. Thus, a proper composition and proportion of fatty acids in culture media should be considered to achieve similar barrier properties to human skin.

Table 2. Experimental conditions used in skin permeatio	n studies with testosterone.
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Animal model	Thickness of tissue	Section of tissue	Tissue preparation	Composition of donor chamber	Volume of donor solution	Drug concentration in the donor chamber	Composition of receptor chamber	Volume of receptor medium	Experimental total time	Permeation parameters (lag time, permeation coefficient and flux)	Retained drug amount or information on TST retention	Protein precipitation solvent (Y or N) / freezing samples? (Y or N)	Quantification method / Injected amount (HPLC)	Area for permeation	Reference
Human cadaver skin	300 µm	Dermis	Dermatomed tissue, following by heating at 55°C for 2min	TST solubilized in ethanol	100 μL	50Ci/mL of <sup>3</sup> H-TST	PBS, PBS + 2% BSA or PBS + 2% HSA	6 mL	-	3.3 x 10 <sup>-3</sup> cm/h	NF	N N	Liquid scintillation counting	1.77 cm <sup>2</sup>	(Kretsos et al., 2008)
Neonatal piglet skin	-	Epidermis and dermis	Surgical scissors	TST (4% w/v) solubilized in ethanol 95% (V/V)	5 μL	4% of TST	Ethanol 20% (v/v) and sodium azide 0,1% (m/v) in water	NF	24 h	0.059 µg/cm².h	NF	N N	HPLC 20 μL	0.79 cm <sup>2</sup>	
Neonatal piglet skin	-	Epidermis and dermis	Surgical scissors	TST (4% w/v) solubilized in ethanol 95% (V/V) and oleic acid 5% (m/V)	5 μL	4% of TST	Ethanol 20% (v/v) and sodium azide 0,1% (m/v) in water	NF	24 h	0.186 µg/cm².h	NF	N N	HPLC 20 μL	0.79 cm <sup>2</sup>	(Nicolazzo et al., 2005)
Neonatal piglet skin	-	Epidermis and dermis	Surgical scissors	TST (4% w/v) solubilized in ethanol 95% (V/V), followed by application of 400 μL of PG 20%	5 μL	4% of TST	Ethanol 20% (v/v) and sodium azide 0,1% (m/v) in water	NF	24 h	0.515 μg/cm².h	NF	N N	HPLC 20 μL	0.79 cm <sup>2</sup>	
Nude mouse skin	-	Full-thickness skin	NF	TST in 50% aqueous ethanol solution	3.1 mL	27 mg/mL (TST was at 71% saturation)	Hydroalcoholic solution (50% ethanol)	3.1 mL	25 h	2.5 x 10 <sup>-3</sup> cm/h	NF	N N	Scintillation counting	NF	
Nude mouse skin	-	Full-thickness skin	NF	Inclusion of a chemical absorption enhancer Eutectic mixture of menthol:TST (4:1) in 50% aqueous ethanol solution	3.1 mL	76.8 mg/mL	Hydroalcoholic solution (50% ethanol)	3.1 mL	25 h	6.6 x 10 <sup>-3</sup> cm/h	NF	N N	Scintillation counting	NF	(Kaplun- Frischoff e
Silastic membrane (silicone membrane)	-	-	NF	TST in 50% aqueous ethanol solution	3.1 mL	27 mg/mL (TST was at 71% saturation)	Hydroalcoholic solution (50% ethanol)	3.1 mL	25 h	9.95 x 10 <sup>-3</sup> cm/h	NF	N N	Scintillation counting	NF	Touitou, 1997)
Silastic membrane (silicone membrane)	-	-	NF	Inclusion of a chemical absorption enhancer Eutectic mixture of menthol:TST (4:1) in 50% aqueous ethanol solution	3.1 mL	76.8 mg/mL	Hydroalcoholic solution (50% ethanol)	3.1 mL	25 h	10.46 x 10 <sup>-3</sup> cm/h	NF	N N	Scintillation counting	NF	
Rat skin	-	Epidermis+dermis	Surgical materials	Ethanol/water (70/30)	3 mL	Saturated solution of TST	Saline solution containing 40% (v/v) polyethylene glycol 400	12 mL	24 h	0.04 x 10 <sup>-3</sup> cm/h (lag time = 2.88 h)	NF	N Y(-20°C)	HPLC 20 μL	2.14 cm <sup>2</sup>	
Rat skin	-	Epidermis+dermis	Surgical materials	Ethanol/water (70/30) plus 1% (m/V) transcutol	3 mL	Saturated solution of TST	Saline solution containing 40% (v/v) polyethylene glycol 400	12 mL	24 h	0.04 x 10 <sup>-3</sup> cm/h (lag time = 1.96 h)	NF	N Y(-20°C)	HPLC 20 µL	2.14 cm <sup>2</sup>	(KIM et al., 2000)
Rat skin	-	Epidermis+dermis	Surgical materials	Ethanol/water (70/30) plus 1% (m/V) lauric acid	3 mL	Saturated solution of TST	Saline solution containing 40% (v/v) polyethylene glycol 400	12 mL	24 h	0.09 x 10 <sup>-3</sup> cm/h (lag time = 5.5 h)	NF	N Y(-20°C)	HPLC 20 µL	2.14 cm <sup>2</sup>	

Rat skin	-	Epidermis+dermis	Surgical materials	Ethanol/water (70/30) plus 1% (m/V) oleic acid	3 mL	Saturated solution of TST	Saline solution containing 40% (v/v) polyethylene glycol 400	12 mL	24 h	0.20 x 10 <sup>-3</sup> cm/h (lag time = 7.04 h)	NF	N Y(-20°C)	HPLC 20 μL	2.14 cm <sup>2</sup>	
Rat skin	-	Epidermis+dermis	Surgical materials	Ethanol/water (70/30) plus 1% (m/V) dodecylamine	3 mL	Saturated solution of TST	Saline solution containing 40% (v/v) polyethylene glycol 400	12 mL	24 h	0.72 x 10 <sup>-3</sup> cm/h (lag time = 7.09 h)	NF	N Y(-20°C)	HPLC 20 μL	2.14 cm <sup>2</sup>	
Human skin	400 µm	Epidermis+dermis	Dermatomed tissue	[ <sup>14</sup> C]-TST was dissolved in 40% ethanol in water	10 μL/cm²	10 μL/cm <sup>2</sup> (from a solution with 1 mg/mL of TST)	Physiological saline + 5% bovine serum albumine + 0.1% sodium azide	4.5 mL	24h	0.005 µg/cm².h	Stratum corneum (0.16%); remaining epidermis (0.41%); dermis (0.26%)	N N	Scintillation Tissue digestion	2.54cm <sup>2</sup>	(J R Heylings
Pig skin	400 µm	Epidermis+dermis	Dermatomed tissued	[ <sup>14</sup> C]-TST was dissolved in 40% ethanol in water	10 μL/cm²	10 μL/cm <sup>2</sup> (from a solution with 1 mg/mL of TST)	Physiological saline + 5% bovine serum albumine + 0.1% sodium azide	4.5 mL	24h	0.055 μg/cm².h	Stratum corneum (0.23%); remaining epidermis (0.89%); dermis (1.17%)	N N	Scintillation Tissue digestion	2.54cm <sup>2</sup>	et al., 2018)
Human skin	400 µm	Epidermis+dermis	Dermatomed tissue	Ethanol/water 50/50 (V/V)	500 μL	32 mM	PBS + 5% (m/V) bovine serum albumin	5 mL	24 h	2.08 to 6.79 x 10 <sup>-4</sup> cm/h (range)	NF	Y (acetonitrile) N	HPLC Limit of quantification = 0.17 µg/mL	0.64 cm <sup>2</sup>	(Baert et al., 2012)
Equine Skin	-	Epidermis+dermis	Surgical materials	PBS	1 mL	Saturated TST solution (0.22 mmol/L)	PBS + 4% (m/V) bovine serum albumin	3.5 mL	24h	6.82 x 10 <sup>-3</sup> cm/h	PBS > EtOH = PG	N N	Liquid Scintillation Analyzer	-	
Equine Skin	-	Epidermis+dermis	Surgical materials	Ethanol/PBS 50/50 (m/m)	1 mL	Saturated TST solution (6.31 mmol/L)	PBS + 4% (m/V) bovine serum albumin	3.5 mL	24h	1.59 x 10 <sup>-3</sup> cm/h	PBS > EtOH = PG	N N	Liquid Scintillation Analyzer	-	(Mills, 2007)
Equine Skin	-	Epidermis+dermis	Surgical materials	Propylene glycol (PG)/PBS 50/50 (m/m)	1 mL	Saturated TST solution (5.14 mmol/L)	PBS + 4% (m/V) bovine serum albumin	3.5 mL	24h	2.04 x 10 <sup>-3</sup> cm/h	n PBS > EtOH = PG	N N	Liquid Scintillation Analyzer	-	
Human skin	-	Stratum corneum + epidermis	Forceps	TST solubilized in 2% of Igepal®	0.5 mL	40 μg/mL	Phosphate buffered saline (PBS)	12.1 mL	30 h	2.31 x 10 <sup>-7</sup> cm/s	-	NF NF	HPLC 50 μL	1.767 cm <sup>2</sup>	
Porcine skin	1000 μm	Epidermis+dermis	Dermatomed tissue	TST solubilized in 2% of Igepal®	0.5 mL	40 μg/mL	Phosphate buffered saline (PBS)	12.1 mL	30 h	1.29 x 10 <sup>-7</sup> cm/s	-	NF NF	HPLC 50 μL	1.767 cm <sup>2</sup>	(Netzlaff et al., 2006)
Bovine skin	1000 μm	Epidermis+dermis	Dermatomed tissue	TST solubilized in 2% of Igepal®	0.5 mL	40 μg/mL	Phosphate buffered saline (PBS)	12.1 mL	30 h	5.42 x 10 <sup>-7</sup> cm/s	-	NF NF	HPLC 50 μL	1.767 cm <sup>2</sup>	
Canine skin (thorax)	-	Epidermis+dermis	Forceps and scissors	Phosphate buffered saline (PBS)	1 mL	Saturated solution (0.21 mmol/L)	PBS pH 7.4 + 4% bovine serum albumin (BSA)	3.5 mL	24h	5.53 x 10 <sup>-4</sup> cm/h	No de comin o	N N	Scintillation Radiolabeled (14C) TST	-	
Canine skin (neck)	-	Epidermis+dermis	Forceps and scissors	Phosphate buffered saline (PBS)	1 mL	Saturated solution (0.21 mmol/L)	PBS pH 7.4 +4% BSA	3.5 mL	24h	4.32 x 10 <sup>-4</sup> cm/h	thorax	N N	Scintillation Radiolabeled (14C) TST	-	(84:11
Canine skin (groin)	-	Epidermis+dermis	Forceps and scissors	Phosphate buffered saline (PBS)	1 mL	Saturated solution (0.21 mmol/L)	PBS pH 7.4 + 4% BSA	3.5 mL	24h	3.85 x 10 <sup>-4</sup> cm/h		N N	Scintillation Radiolabeled (14C) TST	-	(1911); et al., 2006)
Canine skin (thorax)	-	Epidermis+dermis	Forceps and scissors	Ethanol/PBS 50/50 (w/w)	1 mL	Saturated solution (6.26 mmol/L)	PBS pH 7.4 + 4% BSA	3.5 mL	24h	1.12 x 10 <sup>-4</sup> cm/h	Neck = thorax > groin	N N	Scintillation Radiolabeled (14C) TST	-	
Canine skin	-	Epidermis+dermis	Forceps and	Ethanol/PBS 50/50 (w/w)	1 mL	Saturated	PBS pH 7.4 + 4%	3.5 mL	24h	1.21 x 10 <sup>-4</sup> cm/h	1 DOPT OPELOH	N	Scintillation	-	

(neck)			scissors			solution (6.26 mmol/L)	BSA					N	Radiolabeled (14C) TST		
Canine skin (groin)	-	Epidermis+dermis	Forceps and scissors	Ethanol/PBS 50/50 (w/w)	1 mL	Saturated solution (6.26 mmol/L)	PBS pH 7.4 + 4% BSA	3.5 mL	24h	0.73 x 10 <sup>-4</sup> cm/h		N N	Scintillation Radiolabeled (14C) TST	-	
Canine skin (thorax)	-	Epidermis+dermis	Forceps and scissors	Propylene glycol/PBS 50/50 (w/w)	1 mL	Saturated solution (5.21 mmol/L)	PBS pH 7.4 + 4% BSA	3.5 mL	24h	1.63 x 10 <sup>-4</sup> cm/h	Nock = thoray >	N N	Scintillation Radiolabeled (14C) TST	-	
Canine skin (neck)	-	Epidermis+dermis	Forceps and scissors	Propylene glycol/PBS 50/50 (w/w)	1 mL	Saturated solution (5.21 mmol/L)	PBS pH 7.4 + 4% BSA	3.5 mL	24h	3.14 x 10 <sup>-4</sup> cm/h	groin	N N	Scintillation Radiolabeled (14C) TST	-	
Canine skin (groin)	-	Epidermis+dermis	Forceps and scissors	Propylene glycol/PBS 50/50 (w/w)	1 mL	Saturated solution (5.21 mmol/L)	PBS pH 7.4 + 4% BSA	3.5 mL	24h	0.81 x 10 <sup>-4</sup> cm/h	PB3/FG/Ltoff	N N	Scintillation Radiolabeled (14C) TST	-	
Human skin	< 1000 µm	Epidermis Full-thickness skin	Forceps and scissors	PBS + 2% lgepal®	-	0.004%	PBS	12 mL	24h	9.4 x 10 <sup>-4</sup> cm/h (lag time = 0.03 h)	NF	N N	HPLC	1.768 cm <sup>2</sup>	
Reconstructed epiderm EpiDerm <sup>™</sup>	-	Epidermis	Purchased by Laboratoire MatTek Corp.	PBS + 2% Igepal®	-	0.004%	PBS	12 mL	24h	122.4 x 10 <sup>-4</sup> cm/h (lag time = 0 h)	NF	N N	HPLC	1.768 cm <sup>2</sup>	(Schreiber et al., 2005)
Reconstructed epiderm SkinEthic®	-	Epidermis	Purchased by Laboratoire SkinEthic	PBS + 2% Igepal®	-	0.004%	PBS	12 mL	24h	212.4 x 10 <sup>-4</sup> cm/h (lag time = 0.01 h)	NF	N N	HPLC	1.768 cm <sup>2</sup>	
Porcine Skin	< 1000 μm	Epidermis	Forceps and scissors	PBS + 2% lgepal®	-	0.004%	PBS	12 mL	24h	11.5 x 10 <sup>-4</sup> cm/h (lag time = 7.63 h)	NF	N N	HPLC	1.768 cm <sup>2</sup>	
Rat skin	-	Full-thickness skin	Dermatomed skin	EtOH/propylene glycol /water (4:1:1)	250 μL	Saturated solution	PBS	Perfusion (0.6 mL/min)	6 h	2.1 µg/cm².h	NF	N N	HPLC	0.785 cm <sup>2</sup>	
Rat skin	-	Full-thickness skin	Dermatomed skin	EtOH/propylene glycol /water (4:1:1)	250 μL	Supersaturated solution (supersaturation degree = 1.4)	PBS	Perfusion (0.6 mL/min)	6 h	3.1 µg/cm².h	NF	N N	HPLC	0.785 cm <sup>2</sup>	(ML.
Rat skin	-	Full-thickness skin	Dermatomed skin	EtOH/propylene glycol /water (4:1:1)	250 μL	Saturated solution (supersaturation degree = 2.1)	PBS	Perfusion (0.6 mL/min)	6 h	2.2 μg/cm <sup>2</sup> .h (drug crystallization)	NF	N N	HPLC	0.785 cm <sup>2</sup>	et al., 2006)
Rat skin		Full-thickness skin	Dermatomed skin	EtOH/propylene glycol /water (4:1:1)	250 μL	Saturated solution (supersaturation degree = 2.6)	PBS	Perfusion (0.6 mL/min)	6 h	2.2 μg/cm <sup>2</sup> .h (drug crystallization)	NF	N N	HPLC	0.785 cm <sup>2</sup>	
Mice skin	-	Full-thickness skin	NF	TST solubilized in isopropyl myristate (1/99)	1 g	1%	PBS	7-7.3 mL	10 h	2.0 μg/cm <sup>2</sup> .h (lag time<0.5h)	NF	N	HPLC	1.74 cm <sup>2</sup>	
Mice skin	-	Full-thickness skin	NF	TST in isopropyl myristate/lecithin (1/69/30)	1 g	1%	PBS	7-7.3 mL	10 h	2.6 μg/cm <sup>2</sup> .h (lag time<2h)	NF	N	HPLC	1.74 cm <sup>2</sup>	(Imai et al., 2016)
Mice skin	-	Full-thickness skin	NF	TST in isopropyl myristate/lecithin/water (1/65.6/30/3.4)	1 g	1%	PBS	7-7.3 mL	10 h	3.98 µg/cm <sup>2</sup> .h (lag time=1.06h)	NF	Ν	HPLC	1.74 cm <sup>2</sup>	

Mice skin	-	Full-thickness skin	NF	TST in water/lecithin/D- ribose (1/62.8/30/6.2)	1 g	1%	PBS	7-7.3 mL	10 h	2.69 μg/cm <sup>2</sup> .h (lag time=0.77h)	NF	N	HPLC	1.74 cm <sup>2</sup>	
Mice skin	-	Full-thickness skin	NF	TST in lecithin/D-ribose/ tetraglycerol (1/30/55.6/13.4)	1 g	1%	PBS	7-7.3 mL	10 h	2.35 μg/cm <sup>2</sup> .h (lag time=1.02h)	NF	N	HPLC	1.74 cm <sup>2</sup>	
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol/water (20:80)	3 mL	Saturated solution of TST (0.45 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	1.8 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 μL	2.01 cm <sup>2</sup>	
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol/water (40:60)	3 mL	Saturated solution of TST (8.53 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	0.13 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 µL	2.01 cm <sup>2</sup>	
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol/water (50:50)	3 mL	Saturated solution of TST (18.52 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	0.07 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 µL	2.01 cm <sup>2</sup>	
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol/water (60:40)	3 mL	Saturated solution of TST (38.95 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	0.05 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 µL	2.01 cm <sup>2</sup>	(Kim et al., 2001)
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol/water (70:30)	3 mL	Saturated solution of TST (68.32 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	0.04 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 µL	2.01 cm <sup>2</sup>	
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol/water (80:20)	3 mL	Saturated solution of TST (140.01 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	0.02 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 µL	2.01 cm <sup>2</sup>	
Rat Skin	-	Full-thickness skin	Surgical materials	Ethanol 100%	3 mL	Saturated solution of TST (334.03 mg/mL)	Saline solution containing 40% (V/V) polyethylene glycol 400	12 mL	24 h	0.00 x 10 <sup>-3</sup> cm/h	NF	N Y (-20°C)	HPLC 20 µL	2.01 cm <sup>2</sup>	
Mice skin	-	Epidermis+dermis	Surgical materials	TST solubilized in ethanol with and without chemical absorption enhancer (Azone <sup>®</sup> , isopropyl myristate, <i>N</i> -methyl-2- pyrrolidone and propylene glycol)	100 µL	5% (m/V)	40% PEG 200	7 mL	24 h	Azone>IPM>PG>NMP	NF	N N	HPLC	3.14 cm <sup>2</sup>	(Lu et al., 2013)
Amphibian s (fresh dorsa	kin - al) -	Full-thickness skin	Surgical materials	TST dissolved in ethanol/water 1/1 (v/v)	10μL/cm²	40 μg/cm² (dose solution 4 mg/mL)	5% BSA in amphibian Ringer's solution	12.5 mL	8h	J = 5.3 μg/cm².h P = 1.3 x 10 <sup>-3</sup> cm/h	NF	N N	Liquid scintillation counting	1.85 cm²	(Kaufmann e Dohmen, 2016)
Amphibian skin (fresh ventral tissue)	-	Full-thickness skin	Surgical materials	TST dissolved in ethanol/water 1/1 (v/v)	10µL/cm²	40 μg/cm² (dose solution 4 mg/mL)	5% BSA in amphibian Ringer's solution	12.5 mL	8h	J = 12.0 µg/cm².h P = 3.0 x 10 <sup>-3</sup> cm/h	NF	N N	Liquid scintillation counting	1.85 cm²	
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Amphibian skin (frozen dorsal tissue)	-	Full-thickness skin	Surgical materials	TST dissolved in ethanol/water 1/1 (v/v)	10µL/cm²	40 μg/cm² (dose solution 4 mg/mL)	5% BSA in amphibian Ringer's solution	12.5 mL	8h	J = 7.4 μg/cm².h P = 1.9 x 10 <sup>-3</sup> cm/h	NF	N N	Liquid scintillation counting	1.85 cm²	
Amphibian skin (frozen ventral tissue)	-	Full-thickness skin	Surgical materials	TST dissolved in ethanol/water 1/1 (v/v)	10µL/cm²	40 μg/cm² (dose solution 4 mg/mL)	5% BSA in amphibian Ringer's solution	12.5 mL	8h	J = 11.4 µg/cm².h P = 2.9 x 10 <sup>-3</sup> cm/h	NF	N N	Liquid scintillation counting	1.85 cm²	
Gottingen Minipigs (1.5 months)	765 μm	Epidermis+dermis	Dermatomed skin	20% ethanol in 0.05 M phosphate buffer (pH=7.4)	NF	0.1 mg/ml	0.05 M phosphate buffer	12.1 mL	28 h	J = 0.223 μg/cm².h Lag time: 3 to 5 h	NF	N N	Liquid scintillation counting	1.77 cm²	
Gottingen Minipigs (3 months)	765 µm	Epidermis+dermis	Dermatomed skin	20% ethanol in 0.05 M phosphate buffer (pH=7.4)	NF	0.1 mg/ml	0.05 M phosphate buffer	12.1 mL	28 h	J = 0.361 µg/cm².h Lag time: 3 to 5 h	NF	N N	Liquid scintillation counting	1.77 cm <sup>2</sup>	
Gottingen Minipigs (6 months)	765 μm	Epidermis+dermis	Dermatomed skin	20% ethanol in 0.05 M phosphate buffer (pH=7.4)	NF	0.1 mg/ml	0.05 M phosphate buffer	12.1 mL	28 h	J = 0.538 μg/cm².h Lag time: 3 to 5 h	NF	N N	Liquid scintillation counting	1.77 cm²	(Qvist et al., 2000)
Domestic pigs	765 µm	Epidermis+dermis	Dermatomed skin	20% ethanol in 0.05 M phosphate buffer (pH=7.4)	NF	0.1 mg/ml	0.05 M phosphate buffer	12.1 mL	28 h	J = 0.792 μg/cm².h Lag time: 3 to 5 h	NF	N N	Liquid scintillation counting	1.77 cm <sup>2</sup>	
Human skin (abdominal and breast skin)	408 µm	Epidermis+dermis	Dermatomed skin	20% ethanol in 0.05 M phosphate buffer (pH=7.4)	NF	0.1 mg/ml	0.05 M phosphate buffer	12.1 mL	28 h	J = 0.501 μg/cm².h Lag time: 3 to 5 h	NF	N N	Liquid scintillation counting	1.77 cm²	

Mice skin	-	Full-thickness skin	Surgical materials	PEG 400 (5%, w/v) in saline	5 mL (containing 5 mg TST)	1 mg/mL	PBS/PEG 400 (95/5, V/V)	17.5mL	12 h	J = 0.25 µg/cm <sup>2</sup> .h P = 2.49 x 10 <sup>-4</sup> cm/h	NF	N N	HPLC 10µL	1.33 cm²	(Zhang et
Mice skin	-	Full-thickness skin	Surgical materials	Imidazolium ionic liquids and PEG 400 (5%, w/v) in water *Assay were performed with 20 different types of imidazolium ionic liquids	5 mL (containing 5 mg TST)	1 mg/mL	PBS/PEG 400 (95/5, V/V)	17.5mL	12 h	J = 0.42 to 0.9 µg/cm <sup>2</sup> .h (range) P= 4.18 to 8.95 x 10 <sup>-4</sup> cm/h	NF	N N	HPLC 10µL	1.33 cm²	al., 2017)
Human skin (abdominal)	400 ± 50 μm	Edpiermis+dermis	Dermatomed	Pure ethanol	10 µL	1.64 µg/cm²	0.9% NaCl in water + 1% (m/V) BSA + 0.05% (V/V) gentamycin sulfate	NF	24 h	Q <sub>permeated</sub> = 4.7% of applied dose	NF	N N	Scintillation counting	1.00 cm²	(Hewitt et al., 2020)
Reconstructed skin model ( <i>in</i> <i>vitro</i> )	-	Epidermis+dermis	Cell cultivation without alpha- linolenic acid and linoleic acid	TST solubilized in Ethanol/water (1/1, V/V)	100 μL (400 μg TST)	4 mg/mL	PBS + 5% bovine serum albumin	5 mL	24 h	J = 78.3 and 52.9 μg/cm <sup>2</sup> .h (up to 2 h – two different experimental groups)	NF	N Y (4°C)	UPLC 5 μL	0.63 cm²	
Reconstructed skin model ( <i>in</i> <i>vitro</i> )	-	Epidermis+dermis	Cell cultivation with alpha- linolenic acid (ALA)	TST solubilized in Ethanol/water (1/1, V/V)	100 μL (400 μg TST)	4 mg/mL	PBS + 5% bovine serum albumin	5 mL	24 h	J = 46.8 µg/cm <sup>2</sup> .h (up to 2 h)	NF	N Y (4°C)	UPLC 5 μL	0.63 cm²	(Simard et al., 2019)
Reconstructed skin model ( <i>in</i> <i>vitro</i> )	-	Epidermis+dermis	Cell cultivation with linolenic acid (ALA)	TST solubilized in Ethanol/water (1/1, V/V)	100 μL (400 μg TST)	4 mg/mL	PBS + 5% bovine serum albumin	5 mL	24 h	J = 48.8 µg/cm <sup>2</sup> .h (up to 2 h)	NF	N Y (4°C)	UPLC 5 μL	0.63 cm²	
Silicone membrane	0.05 cm	HT-6240 BISCO <sup>®</sup> transparent membrane	-	TST solubilized in EtOH:PBS (50: 50)	900 μL	7.07 x 10 <sup>3</sup> μg/mL	EtOH: PBS (50: 50)	5 mL	8 h	J = 64 μg/cm <sup>2</sup> P = 9.2 x 10 <sup>-3</sup> cm/h Lag time = 0.22 h	NF	N N	UV spectroscopy (245 nm)	0.64 cm²	(Alberti et al., 2017)
Human skin (abdominal)	1000 μm	Full-thickness skin	Dermatomed skin	TST solubilized in ethanol/water 1/1 (v/v)	25 μL/cm²	4 mg/mL	5% BSA in water	4 mL	24 h	P = 69.3 x10 <sup>-5</sup> cm/h Lag time = 4.4	1.7%	Y (ethanol) N	Liquid scintillation counting	1 cm²	(Guth et al., 2015)

Human skin	400 μm	Epidermis+dermis	Dermatomed skin	80% of the maximal solubility in a 50/50 ethanol/H2O (% V/V)	500 μL	9.71 mg/mL	PBS + 5% (m/V) bovine serum albumin	5 mL	24 h	3.92 x 10 <sup>-4</sup> cm/h	NF	Y (acetonitrile) N	HPLC 25 μL	0.64 cm²	
Human skin	400 μm	Epidermis+dermis	Dermatomed skin	80% of the maximal solubility in a 50/50 ethanol/H2O (% V/V) + Azone (1%, W/V)	500 μL	9.71 mg/mL	PBS + 5% (m/V) bovine serum albumin	5 mL	24 h	25.5 x 10 <sup>-4</sup> cm/h	NF	Y (acetonitrile) N	HPLC 25 µL	0.64 cm²	(Veryser et al., 2015)
Human skin	400 μm	Epidermis+dermis	Dermatomed skin	80% of the maximal solubility in a 50/50 ethanol/H2O (% V/V) + phytoceramides (1%, W/V)	500 μL	9.71 mg/mL	PBS + 5% (m/V) bovine serum albumin	5 mL	24 h	2.05 to 7.02 x 10 <sup>-4</sup> cm/h (range) 10 phytoceramides were tested as chemical absorption enhancers	NF	Y (acetonitrile) N	HPLC 25 µL	0.64 cm²	
Human skin	1) 500 μm 2) 900- 1100 μm	Epidermis+dermis	1) Dermatomed skin 2) Full-thickness skin	TST solubilized in ethanol 50% (V/V)	16 μL	4 mg/mL	Saline + 5% (w/v) bovine serum albumin, pH 7.4	0.4 mL	24	1) 1.82 μg/cm <sup>2</sup> .h (lag time = 0.6 h) 2) 0.18 μg/cm <sup>2</sup> .h (lag time = 2.2 h)	1) 136 μg /mL 2) 112 μg/mL	N	Liquid scintillation counting	0.64 cm <sup>2</sup>	(Wilkinson et al., 2006)

## 7. ASSAY TOTAL DURATION & SAMPLING INTERVALS

Considering that the TST presents a low diffusion rate through different skin layers, it is rational to consider a long experimental time in permeation studies. Moreover, the more effective barrier effect of the membrane, the longer the time of experiment required to obtain the permeation data for a certain compound. In fact, most of the permeation assays with TST were performed over a time equal or greater than 24 h (Table 2). One the other hand, a time exposure of 12 h or less was enough for permeation assays performed with tissues characterized by greater permeation as mice, rat and amphibian skin. Therefore, the assay total duration depends mainly on the selected membrane model. The distribution of the sampling intervals should be as homogeneous as possible and should not consider too many points. Permeation studies with a sampling time at 12 h and the next at 24 h can be found, however, if any operator error occurs during sampling, all permeation parameters (flux, permeation coefficient and lag time) may be erroneously calculated. Therefore, a homoscedastic distribution of sampling intervals should be designed. Overall, the concentration of TST in receptor fluids was evaluated at time intervals of 0, 1, 2, 4, 8, 12, 16 18, 20, 21 and 24 h in studies with human skin (Baert et al., 2012; Hewitt et al., 2020; Veryser et al., 2015) Another problem for drugs as TST is the inclusion of many sampling intervals. For each replacement of the medium of the receptor phase, the drug is diluted. In these situations, if the quantification method is not sensitive enough, TST may not be detected. This problem occurs particularly when various initial sampling intervals are defined (drug is found in the least amount in the medium in these periods).

# 8. SKIN INTEGRITY ASSAYS

The evaluation of tissue integrity is strongly recommended for permeation tests involving long incubation periods, such as those considered for the TST. Permeation assay itself can provide indications of the impairment of barrier properties. A change in tissue structure can result in permeability changes (Bennion et al., 2017). If the permeation rate is modified over the course of the assay or an extremely large amount of drug permeates through membrane in specific intervals, this fact may indicate a compromise in the barrier properties and further investigations should be carried out.

OECD, WHO and FDA guidelines recommend the evaluation of skin integrity prior to the permeation study to identify samples that affect the tissue barrier function. In addition to a visual inspection of the skin, the OECD proposes the measure of transepidermal electrical resistance (TEER), transepidermal water loss (TEWL) and tritiated water. No guidance on how to perform and interpret these integrity tests or then reference values for each assay is available in official guidelines.

While physical methods evaluating TEER and TEWL are useful for the selection of tissues for assays (Bartosova e Bajgar, 2012), histological studies allows to identify the region of tissue most affected (e.g. epidermal desquamation, dermal vacuolization).

TEER represents a simple, rapid, and cost-effective method of cellular barrier integrity. TEER measurements based on impedance spectroscopy have proven to be more reliable and provide more information about the cells when compared to the Ohm's law method (Srinivasan et al., 2015). The measured resistance is dependent on the device, applied frequency, resulting current, ionic strength of the solution, temperature and surface area of the tissue (Guth et al., 2015). Skin samples from different animal models may also affect TEER values. In a study with human, mouse, guinea pig, pig, rat and rabbit skin, TEER values were  $\geq 10, 5, 5, 4, 3$  and 0.8 k $\Omega$ , respectively (Davies et al., 2004). Kaufmann and Dohmen (2016), for instance, used TEER to evaluate the integrity of fresh excised and frozen amphibian skin. The higher impedance measurements for fresh skin suggested that the freezing step affected the epidermal integrity. In fact, the ice crystal formation in the frozen skin can lead to the disruption of epidermal cells, impairment the tight junctions, and hence impedance decreases. The  $K_p$  value of TST through the fresh dorsal skin was lower (1.3 x  $10^{-3}$ cm.h<sup>-1</sup>) than frozen skin (1.9 x 10<sup>-3</sup> cm.h<sup>-1</sup>). On the other hand, a minor transition between intact and damaged skin can be found due to the low impedance of amphibian skin and thus methods such as TEWL would be recommended for analysis of amphibian skin integrity.

TEWL methods measure water vapor flux in the air above the SC, which is an indicator of water diffusion through SC and its barrier properties. A temperature and moisture stabilization time around the probe are required before measurements (Neupane et al., 2020). Another caution refers to the complete removal of moisture on the skin surface generated from rehydration of the frozen skin samples or TEER measurements given that the assay objective is to measure exclusively the water loss from the skin sample (Guth et al., 2015). For human skin, TEWL values ranges from 4

to  $10 \text{ g/m}^2/\text{h}$  depending on the tissue area. When the epidermis is damaged, these values may increase up to 30-times (Boer et al., 2016).

The transepidermal water flux (TWF), which involves the measurement of tritiated water (a radioactive form of water), is another assay able to evaluate the skin barrier function. This test can be carried out before, during or after the permeation assay. When it is performed only before permeation study, eventual tissue damages caused by treatments will not be identified. In this same way, limitations may be found when TWF is performed only at the end of permeation assays. Once the tissue is exposed to infinite amounts of water or hydration for many hours during the permeation and it is often washed after its removal of diffusion cells, which may lead to tissue deterioration, TWF measurements at this stage may reject previously intact tissues (Fabian et al., 2017).

Although these assays detect only changes in the polar transport pathway (Flynn et al., 1974), Guth et al. (2015) considered them during the evaluation of tissue integrity prior to permeation studies with TST. The limit to distinguish damaged and intact tissues was 10 g m<sup>-2</sup> h<sup>-1</sup>, 4.5 x 10<sup>-3</sup> cm h<sup>-1</sup> and 2 k $\Omega$  for TEWL, TWF and TEER, respectively.

The inclusion of chemical markers (e.g., mannitol) whose permeation is already well known have also been considered. In general, two reference compounds presenting different physicochemical characteristics are selected. The results of these tests should be carefully analyzed given that the presence of additional compounds in the donor phase may affect the absorption characteristic of the test compound due to changes in solubility or saturation levels (Guth et al., 2015).

Absorption chemical enhancers are the main agents responsible for compromising the skin barrier properties; however, it is important to highlight that the drug itself may also affect them. For TST, studies have demonstrated that fluctuations in its level modulate barrier function and that hormone replacement with TST can exert a negative consequence for permeability barrier homeostasis (Kao et al., 2001).

# 9. SAMPLE STORAGE BEFORE QUANTIFICATION

Samples should be quantified immediately after the permeation test has been completed; however, some studies have stored the samples under refrigeration until quantification step (Table 2). It is well known that the storage of drugs presenting low aqueous solubility under refrigeration can lead to the precipitation and that analytical errors can increase because of a non-soluble drug fraction. On the other hand, storage at room temperature increases the chances of drug degradation by hydrolysis, oxidation and/or enzymatic reactions. Thus, a prior analysis of the drug's stability in different aqueous media and conditions (presence and absence of enzymes, temperature, pH, etc.) is recommended.

## **10. DRUG QUANTIFICATION STEP**

The quantification of steroids from biological matrices (e.g. blood, plasma, urine and skin homogenate) usually involves extraction as the first step and such procedures are carried out by selecting specific solvents. The liquid-liquid extraction may lead to emulsion formation, which would be avoided by a centrifugation step; however, this latter is time-consuming. During the selection of extraction solvent, both polarity of the steroid and its interaction with binding proteins should be considered. Solvents such as acetonitrile and methanol act by providing the disruption of steroid-protein binding through a mechanism of protein denaturation. After solvent addition, the mixture is vortexed, and the protein plug removed by centrifugation (Makin et al., 2010). The protein precipitation efficiency is associated with the solvent type and solvent/biological matrix ratio used. Overall, a ratio 2:1 of solvent (acetonitrile or trichloroacetic acid) to biological matrix is able to precipitate more than 90% of proteins (Polson et al., 2003). In human skin matrices containing TST, Baert et al. (2012) removed proteins by adding a ratio 1:1 of acetonitrile/matrix. No analytical problems were reported in this study.

The absence of proteins in the supernatant may be confirmed by adding a dye capable of binding proteins. When proteins are found, Coomassie blue dye under acidic conditions changes the color from brown to blue (Rodger e Sanders, 2017). The dye binds to basic amino acids by a combination of hydrophobic interactions and heteropolar bonding (Drabik et al., 2016).

In addition to liquid-liquid extraction, solid-phase extraction (SPE) has been often applied to separate the steroids from biological interferences. It is composed by a sorbent such as microparticulate silica coated with octadecasilane, which is packed into syringes or cartridges (Makin et al., 2010). In the case of TST, which has an extremely low permeation rate and the volumes collected cannot be large, the application of this method is limited. The large number of samples also makes this method expensive.

HPLC has been the most important technique to quantify steroids in different matrices; however, not enough sensitivity could be achieved when a UV detector is used (Makin et al., 2010). Baert et al. (2012), for example, found a quantification limit of 0.17  $\mu$ g/mL for TST by using this technique. Once the TST has an extremely low initial permeation rate (in the nanogram range) and several authors have considered HPLC as a quantification method, initial time intervals are more susceptible to analytical errors.

If reversed-phase columns are selected, guard column (30-70 mm in length) should be included to avoid the accumulation of non-polar material on the columns or time-consuming clean-up procedures of sample or column. LC-MS-MS (i.e. tandem mass spectrometry) may represent an alternative to overcome this extensive need of clean-up of samples and also is able to provide a lower quantification limit than HPLC-UV (Makin et al., 2010).

Another alternative to improve the sensitivity of the quantification method is to increase the amount of sample injected, however, this must be compatible with the size of column used. In our database for TST (Table 2), most of the authors used an injection volume of 20  $\mu$ L and a non-processing of the samples before injection (without solvent extraction or SPE). When the injection volume is increased, a greater need for sample clean-up/purification steps should be considered.

## **11. DRUG RETENTION**

Androgen receptors are expressed in sebocytes, hair follicle, dermal papilla and keratinocytes. Hence, TST plays an important role in the sebum production, control hair growth as well as hair loss, epidermal growth and differentiation (Choudhry et al., 1992). The mechanism of action of TST occurs by its local conversion to dihydrotestosterone (DHT) by  $5\alpha$ -reductase (type 1). DHT then binds to the androgen receptor. This means that the increase of TST lead to an increase of DHT concentrations, which is associated with hair loss (alopecia) (Riggs et al., 2002; Shapiro e Price, 1998). The action of this hormone is regulated by cell-type-specific activation or deactivation in human skin. In physiological conditions, fibroblasts present in the

dermis activate mainly TST to DHT. In contrast, keratinocytes present in the epidermis deactivate TST, forming androstenedione (Münster et al., 2003).

The higher concentration of TST in men is associated with a thicker epidermis, and greater amount of collagen and moisture compared to women skin (Markova et al., 2004). On the other hand, the higher levels of TST in male stimulate a large production of sebum, leading to fatty glow and coarser pores of the skin (Baumann, 2002). TST can also perturbs the epidermal barrier homeostasis considering that studies showed a retard in barrier development of skin fetus and slowed barrier recovery in adult skin (Kao et al., 2001). Other negative effects include the inhibition of skin wound healing in males and the TST association with an enhanced inflammatory response (Ashcroft e Mills, 2002). In contrast, a reduction in physiological levels of the hormone has also a negative impact on epidermal skin moisture, elasticity, and thickness (Bernard et al., 2012). In these cases, the local hormone replacement with TST can be advantageous to improve the aspect of skin aging in men.

Transdermal products containing 1% TST are commonly used to reproduce diurnal physiological variations of the hormone in the treatment of male hypogonadism (Mazer, 2000). Once application of TST in skin is focused on the systemic treatment, many *in vitro* studies do not evaluate the presence of the hormone in skin homogenates. Magnusson et al. (2006), for instance, evaluated the distribution of TST in human skin using excised epidermis, dermis, and full-thickness skin. TST retention was higher in the uppermost skin layers (SC > epidermis> dermis), which is attributed to its lipophilicity. In hormone retention assays performed with viable epidermis *versus* full-thickness skin, the dermis layer seemed to restrict TST permeation. The TST retention in viable epidermis (isolated tissue). Heylings et al. (2018) also reported a lower affinity of the TST for dermis. The authors found 0.41 and 0.32% of dose applied of TST (1.0 mg/mL) in epidermis and dermis, respectively.

TST has a great affinity for the lipophilic layers of the skin, contributing for its accumulation in the tissue. This could generate important adverse effects from a topical administration. However, studies have not thoroughly investigated the possible toxic effects of this hormone in skin.

#### **12. FINAL CONSIDERATIONS**

The interplay/tradeoff solubility-permeation should not ignored and researchers need to achieve an optimal solubility-permeation balance to maximize the permeation of poorly soluble drugs such as TST. The inclusion of ethanol in aqueous medium increased the solubility or free fraction of TST in the solution, which solves problems of solubility in receptor fluids and maintains sink conditions; however, a reduction in its permeation rate was found when compared to aqueous solutions without solubilizing agents. In this same way, an optimal supersaturation degree should be found to maximize permeation and avoid drug precipitation phenomena. Infinite doses have been frequently used in commercial preparations of hormone due to its low cutaneous absorption; however, an increase in permeation is not always achieved with this approach as reported here.

The type of tissue selected as the membrane model as well as its preparation has also been shown to strongly affect the TST permeation rate. When full-thickness skin is considered, the dermis acts as a barrier that restricts the diffusion of TST, significantly reducing the partitioning into the receptor chamber. Therefore, the amount of absorbed TST is underestimated. The drug retention in epidermis is also increased in an unrealistic way. This fact represents the main reason for prioritizing dermatomized skin. Substantial differences in the rate of transport of the hormone were found depending on the animal model and between biological and artificial membranes/tissues. Despite its widespread use, mouse/mice skin is much more permeable than human skin and should not be prioritized in initial studies (when the permeation of a certain molecule is not yet known). Human skin and pig skin are more suitable in these situations. Mouse/mice skin can be useful, for example, for comparing the performance of different formulations and *in vivo* distribution studies. Likewise, the TST permeation results obtained with artificial membranes are quite different those with human skin.

Although various cutaneous permeation studies with TST were presented in this review, few studies evaluating its retention in the dermis and epidermis are reported. As already mentioned, the skin presents specific receptors for TST, which makes this type of study even more important. This fact could be justified by the complexity of separating the epidermis from the dermis. As already mentioned, the currently available methods that use heating or chemical treatment can affect tissue integrity. For this reason, manual tissue separation has been proposed, but poorly reproducible results are usually found.

Although traditional methods in the pharmaceutical field have been considered for the quantification of TST in biological matrices, in-depth studies of sample preparation are still scarce. Improvements in terms of the method sensitivity could be achieved with an adequate processing of the samples, reducing the experimental time and the data generated would also be more reliable.

In summary, this review contributes to define more properly the experimental variables of cutaneous permeation assays with poorly soluble molecules such as the TST, creating new discussions for future revisions of official guidelines.

# **CONFLICT OF INTEREST**

The authors report no conflicts of interest.

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# 4. CAPÍTULO III – EFEITO DOS TERPENOS COMO PROMOTORES DE ABSORÇÃO PARA A TESTOSTERONA

Este capítulo trata da avaliação da permeação percutânea da testosterona em que diferentes terpenos foram selecionados como promotores químicos de absorção. Análises espectroscópicas (FTIR) e térmicas (DSC) foram consideradas para uma investigação de interações específicas entre estes terpenos e componentes da pele (domínios lipídicos e estruturas proteicas) a fim de entender como as propriedades de barreira do estrato córneo são modificadas. O produto deste capítulo é um artigo científico, que está sob avaliação na revista *Journal of Drug Delivery Science and Technology*.

# ENHANCING THE PERMEATION OF TESTOSTERONE THROUGH HUMAN SKIN WITH NATURAL TERPENES

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#### ABSTRACT

Chemical absorption enhancers (CAEs) may be demanded in formulations with drugs having low diffusion rate or high retention in skin. Although several CAEs are available for commercial use, incompatibility or local irritation issues are common in daily applied formulations. Terpenes has been shown to be promising candidates in this context. In addition to high efficacy as CAE, they are classified as safe by FDA and can provide secondary biological properties (e.g., antioxidant, anti-inflammatory and antimicrobial action). Therefore, the effects of three different terpenes on the cutaneous permeation of testosterone (TST) was investigated by the Franz-type diffusion model. An enhancement ratio of 7.61, 5.10 and 2.52 was obtained after treatment with carvacrol,  $\alpha$ -bisabolol and menthol, respectively. Carvacrol provided higher permeation flux and cutaneous retention of TST than other terpenes. Mechanistic studies were performed to evaluate the effect of CAEs on biophysical properties of the stratum corneum. Spectroscopic and thermal analyses showed structural changes in the tissue after all treatments and suggested that carvacrol causes a greater disorder in lipid domains. As TST is preferentially transported through the skin by passive diffusion due to its lipophilic nature, this terpene would be more promising as CAE. This terpene could also play a synergistic effect with TST in hypogonadism, but further studies are needed to prove the advantages of this association.

Keywords: terpenes; skin permeation; testosterone; human skin.

#### **1. INTRODUCTION**

The transdermal route has been widely used in recent years due to numerous advantages over the oral once it avoids the first-pass metabolism, protects drugs from the gastrointestinal tract, allows a sustained release of drugs and provides high patient compliance to treatment [1]. This alternative administration route is even more attractive for the testosterone (TST) once it is extensively metabolized after an oral administration. In this situation, large oral doses are required to achieve normal plasma levels, increasing the chances of side effects such as liver toxicity and lowering of HDL cholesterol levels [2].

Numerous derivatives of TST have also been synthesized aiming to extend its *in vivo* biological activity following an oral application. Although these agents are characterized by a high efficacy, behavioral changes (e.g., aggression), hepatotoxicity, modification of blood lipid levels and coagulation factors and some degree of masculinization in women and children are often observed [3].

In this context, the transdermal route has been shown to be a promising alternative to overcome these limitations, which explain the large number of transdermal preparations of TST available on the market in recent years. Transdermal patches and gels represent the most common dosage forms [4]. Given that the TST presents a high value of log P, this compound binds so strongly to the tissues, particularly in the more superficial layers (*stratum corneum* and viable epidermis), resulting in a reservoir effect and low transport rate to the dermis [5]. Moreover, the skin has receptors for this hormone due to the physiological action it plays in that region. The TST interacts directly with androgen receptors, which are localized in most epidermis keratinocytes [6].

In this context, the inclusion of chemical absorption enhancers (CAEs) in transdermal formulation can be considered as an approach to increase the delivery of TST in plasma. Alcohols have been often included in gel formulations to play a role as a CAE and solubilizing agent, but its highly dehydrating nature causes a quick dryness of the epidermis. Consequently, the permeation of TST through the skin may be hampered. In addition, local irritation may be observed [7]. Indeed, in a study with different

concentrations of ethanol in water, lower transport rate of TST through rat skin was observed for increasing amounts of ethanol [8].

Natural terpenes, in turn, would avoid these events and have been pointed out as promising CAE candidates due to their low toxicity compared to other synthetic representatives [9]. As dosage forms usually contain high doses of TST, which also can irritate the skin, terpenes with anti-inflammatory or anti-irritant activity would minimize or even eliminate this problem. In addition, these compounds may play a synergistic action with TST in different organic tissues or provide additional properties in formulations (e.g., antimicrobial, antioxidant and solubilizing action). Therefore, the role of three terpenes (carvacrol,  $\alpha$ -bisabolol, and menthol; Figure 1) on the permeation rate of TST through human skin was studied. Spectroscopic and thermal studies were also performed to understand the terpene/skin interactions.



Figure 1. Chemical structures of terpenes and testosterone.

#### 2. MATERIAL AND METHODS

## 2.1 Materials

TST base (98% purity) and propylene glycol (PG) were obtained from Infinity<sup>®</sup> Pharma (Campinas, SP, Brazil) and CRQ (Diadema, SP, Brazil), respectively. Carvacrol (98%), α-bisabolol (93%) and menthol (99%) were purchased from Sigma-Aldrich (St. Louis, MO, USA). Phosphate buffered saline was prepared with 10.1 mM Na<sub>2</sub>HPO<sub>4</sub>, 1.8 mM KH<sub>2</sub>PO<sub>4</sub>, 136.9 mM NaCl, 2.7 mM KCl and adjusted to pH 7.4 with 1 M HCl. Krebs-Ringer bicarbonate buffer was prepared with 119 mM NaCl, 4.7 mM KCl, 1.2 mM KH<sub>2</sub>PO<sub>4</sub>, 25 mM NaHCO<sub>3</sub>, 2.5 mM CaCl<sub>2.2</sub>H<sub>2</sub>O, 1.0 mM MgCl<sub>2.6</sub>H<sub>2</sub>O, 5.5 mM glucose and adjusted to pH 7.4 with 1 M HCl. Methanol and acetonitrile were

of HPLC grade. All other chemicals and reagents were of analytical grade. Ultrapure water was obtained from a Milli-Q<sup>®</sup> purification system (Millipore, Milford, MA, USA).

#### 2.2 Methods

## 2.2.1 Ex vivo permeation studies

### 2.2.1.1 Preparation of human skin samples

Human skin was obtained from abdominal surgery performed at a local hospital. The study was approved by the University Research Ethics Committee (CAE/protocol number: 87349418.7.0000.0121). Full-thickness skin was stored in cold Krebs buffer immediately after the surgery. The skin samples were prepared by removing the adipose tissue, placed on aluminum foil sheets and stored at 20 °C until use (no more than one month). Tissues presenting a thickness between 500 and 700  $\mu$ m were used in the assays. Prior to use, the tissue was thawed in PBS pH 7.4 for, at least, 30 min.

### 2.2.1.2 TST permeation/retention through the human skin

Prehydrated human skin membranes were fixed between the donor and receptor chambers of the Franz-type diffusion cell, which present a diffusional area of 1.77 cm<sup>2</sup>. The receptor chamber was filled with 10 mL of a PBS/propylene glycol solution (50/50, v/v) and kept at 37°C under constant magnetic stirring at 750 rpm. Propylene glycol was selected for the receptor phase aiming to maintain a skin condition. The donor chamber was filled with 1% (w/w) of TST in propylene glycol (control solution) or this solution in presence of 10% (w/w/w) of each terpene (carvacrol,  $\alpha$ -bisabolol and menthol). Different sampling time intervals were considered to define the experimental time (1, 2, 2)4, 6, 8, 12, 18 e 24 h or 8, 10, 12, 16, 18, 24, 30 and 36 h). After each aliquot withdrawal, fresh solution was added to keep constant the same volume in receptor phase. After 24 or 36 h, tissues were removed from diffusion chambers, cleaned using a cotton swab embedded with Milli-Q water and the epidermis was separated from the dermis with a cutting blade. The TST was extracted from the tissues by using an acetonitrile/PBS buffer mixture (70/30 v/v), followed of an ultrasound treatment for 30 min. Finally, the mixture solvent/tissue was centrifuged at 14,000 rpm for 5 min for protein precipitation. The content of TST was determined by HPLC for samples obtained from the receptor fluid and skin layers membranes.

The steady-state flux ( $J_{ss}$ ) was calculated from the linear slope of the cumulative amount of TST permeated *versus* time. The permeability coefficient (P) was determined by the relation between Js and the initial donor concentration ( $P = J_{ss}/C_d$ ) and lag time by the intercept of the regression line with the abscissa.

The expected testosterone plasma concentration ( $C_{ss}$ ) was calculated by the following equation [10]:

$$C_{ss} = J_{ss}^* A / CL$$
 (Equation 1)

Where:  $J_{ss}$  is the permeation flux; A is the application area (40 cm<sup>2</sup>), and CL represents the total testosterone body clearance, which is 54.33 L/h [11].

#### 2.2.2 HPLC analysis

All the chromatographic separations were performed in an Agilent 1200 series HPLC system (Waldbronn, Germany) equipped with online degasser, quaternary pump, autosampler and diode array detector. The separation was performed in a Zorbax Eclipse Plus C18 column (100 x 4.6 mm ID, 3.5  $\mu$ m particle size; Agilent, USA) at 30 °C. A isocratic elution of methanol and water (85/15, v/v) was considered as mobile phase. The flow rate and injection volume were 0.75 mL min<sup>-1</sup> and 10  $\mu$ L, respectively. The detector wavelength was set at 240 nm. The Open Lab software was used for data acquisition and treatment. A calibration curve in the range of 0.01 to 100  $\mu$ g mL<sup>-1</sup> was prepared in acetonitrile and water (70/30, v/v) from appropriate dilutions of stock solution. The calibration curve was constructed using peak area *versus* TST concentration (y = 42.6x + 12.1; R<sup>2</sup> = 0.999).

#### 2.2.3 Mechanistic studies

#### 2.2.3.1 DSC analysis

Thermal analyses were performed in a DSC-60 equipment (Shimadzu, Kyoto, Japan). Firstly, the stratum corneum (SC) was carefully separated from other skin layers and divided into sections of approximately 2 mg with a scalpel. SC samples were immersed in propylene glycol (control solution) containing 1% (w/w) of TST or in this solution in presence of 10% (w/w/w) of each terpene (carvacrol,  $\alpha$ -bisabolol and menthol). Non-treated tissue (control) was immersed in a PBS solution. The period of exposure was similar to that of the permeation assay. The treated samples and control (untreated tissues) were sealed in aluminum hermetic pans and heated from 30 to

200 °C at a 5 °C/min rate in a nitrogen atmosphere. Changes in the enthalpy ( $\Delta H$ ) and corresponding mean transition temperature (*Tm*) values were monitored to understand lipid fluidization behavior or extraction processes as well as protein denaturation on the SC.

## 2.2.3.2 FT-IR spectral analysis

The SC was carefully separated from other skin layers and divided into sections of approximately 20 mg with a scalpel. SC samples were immersed in propylene glycol (control solution) containing 1% (w/w) of TST or in this solution in presence of 10% (w/w/w) of each terpene (carvacrol,  $\alpha$ -bisabolol and menthol). Non-treated tissue (control) was immersed in a PBS solution. The period of exposure was similar to that of the permeation assay. FT-IR spectra of each sample were recorded in a frequency range from 600 to 4,000 cm<sup>-1</sup> on a Perkin-Elmer Frontier equipment (Waltham, USA), at room temperature, with the collection of 20 scans at a resolution of 4 cm<sup>-1</sup>. Data were shown as the mean spectrum of three different samples for each treatment.

# 2.2.4 Statistical analysis

The OriginPro 9 software was used for data analysis. One-way ANOVA followed by Tukey's post-hoc test was selected for analysis of the differences between groups/treatments.

## **3. RESULTS**

### 3.1. Ex vivo permeation studies

Once the TST has low aqueous solubility (24  $\mu$ g mL<sup>-1</sup>), a theoretical analysis of its solubility in different media was performed to define the composition of the receptor and donor fluids. The TST exhibits a solubility of 14.86 and 133.00 mg mL<sup>-1</sup> in propylene glycol (PG) and ethanol, respectively (typical co-solvent often used in permeation assays) [12]. Although the hormone is more soluble in ethanol, PG was selected as solubilizing agents of TST due to the lower skin toxicity [13,14]. Moreover, studies have shown that a greatly increased solubilization of the TST in the donor phase of diffusion cells can negatively impact its transport rate. In a study testing 0.22, 6.31 and 5.14 mM of TST in only PBS, ethanol/PBS and propylene glycol/PBS, permeation coefficient values through equine skin were 6.82, 1.59 and 2.04 x 10<sup>-3</sup> cm/h, respectively [15]. In another study with different concentrations of ethanol in water and TST, lower transport rate through rat skin was found for increasing amounts of ethanol and TST [8].

The first part of the study was focused on determining the experimental time for the permeation assay. No terpene was considered in this step, only the control solution (TST in propylene glycol). This analysis is remarkable considering that the TST presents low transport rate and high affinity for the skin as already mentioned [16]. The ex vivo permeation assays were performed up to 24 and 36 h and different sampling time intervals were selected as described in method section. Interestingly, TST was quantified with precision only from 18 and 16 h for assays performed within the 24 and 36 h, respectively (Figure 2A). Once various samplings are carried out at the initial time intervals in the 24-hour assay, the receptor medium is diluted after the addition of fresh buffer. For this reason, TST was quantified appropriately later in the shorter duration assay (24 h). Based on this result, the first sample from 36 h assay was taken only 12 h after the start of the test. In this new time interval, TST was quantified with precision in different times and thus a more reliable permeation flux and profile were achieved. In cutaneous retention analysis, a two to threefold increase in amount of TST retained was found in both tissue layers (epidermis and dermis) when the assay time ranged from 24 to 36 h (Figure 2B). Therefore, 36 h was selected as experimental time to evaluate the effect of terpenes on TST permeation.





**Figure 2.** Cutaneous permeation (A) and retention (B) of testosterone through the human skin after 24 and 36 h (n=6).

The permeation of TST across human skin was enhanced in the presence of the terpenes (Fig. 3) compared to the control solution (1% w/w of TST in PG). In addition, the time required to TST achieve a steady-state flux (lag time) was approximately 1.6-fold shorter for carvacrol-treated skin compared to control-treated tissue (Table 1), suggesting a faster onset therapeutic activity in the presence of this terpene. The treatment with carvacrol increased the permeation flux and permeation coefficient of the TST in 7.6-fold compared to the control (Table 1). This terpene also significantly increased the amount of TST accumulated in the receptor chamber at 36 h (47.08 ± 16.79 µg.cm<sup>-2</sup>) in relation to menthol (14.68 ± 9.39 µg.cm<sup>-2</sup>). On the other hand, no significant difference (ANOVA/Tukey test, p>0.05) was observed between carvacrol and  $\alpha$ -bisabolol (29.20 ± 11.85 µg.cm<sup>-2</sup>).



**Figure 3.** Skin permeation profile of testosterone in the presence of carvacrol ( $\bullet$ ),  $\alpha$ -bisabolol ( $\blacktriangle$ ) and menthol ( $\nabla$ ). TST in propylene glycol was used as control ( $\blacksquare$ ) (*n*=6).

**Table 1.** Permeation parameters of testosterone (TST) alone (control solution) and in the presence of terpenes (n=6).

Treatment	$J_{ss}$ (µg.cm <sup>-2</sup> .h)	<i>P</i> (cm.h <sup>-1</sup> x 10 <sup>-3</sup> )	Lag time (h)	Enhancement ratio (ER)
Control	$0.215\pm0.03$	$0.022\pm0.003$	$12.99\pm2.46$	-
Carvacrol	$1.638\pm0.57*$	$0.164 \pm 0.060 *$	$7.95\pm0.24\texttt{*}$	7.61
α-bisabolol	$1.072\pm0.43$	$0.107\pm0.043$	$11.65\pm1.97$	5.10
Menthol	$0.542\pm0.27$	$0.0542 \pm 0.027$	$11.42\pm3.35$	2.52

\*Statistically different compared to control (ANOVA/Tukey test, p < 0.05).

For the obtained maximum permeation flux ( $J_{ss} = 1.638 \ \mu g/cm^2.h$ ), the expected TST plasma concentration is 120.59 ng/dL. Although this concentration is outside the therapeutic range, which is 300-1000 ng/dL [17,18], it is important to mention that patients with hypogonadism already have a certain physiological concentration of the hormone (even if reduced). In addition, the absence of blood flow in *ex vivo* assays

hinders the permeation flux of active compounds through dermis layer. Therefore, the real permeation rates of TST should be higher in relation to the values found in *ex vivo* experiments. Changes in TST and terpene concentration as well as combination of CAEs could be alternatively considered.

Retention results of TST in the epidermis layer are in accordance with permeation data (Fig. 4). The treatment with carvacrol provided a significantly higher TST retention in the epidermis ( $305.20 \pm 45.00 \ \mu g.g^{-1}$ ) than control ( $120.60 \pm 48.20 \ \mu g.g^{-1}$ ) and menthol ( $119.20 \pm 56.80 \ \mu g.g^{-1}$ ); however, no significant difference (Tukey test, p < 0.05) was found compared to  $\alpha$ -bisabolol ( $249.50 \pm 48.90 \ \mu g.g^{-1}$ ). Once TST have more affinity for epidermis (more lipophilic skin layer), less amount of this hormone was found in the dermis. The amount of TST accumulated in dermis (Figure 4) after treatment with terpenes (carvacrol,  $77.70 \pm 21.20$ ;  $\alpha$ -bisabolol,  $89.70 \pm 14.20$ ; menthol 29.90  $\pm 6.80$ ) was not statistically different from the control ( $64.00 \pm 4.60 \ \mu g.g^{-1}$ ).



**Figure 4.** Retention of testosterone (TST) in the epidermis and dermis layers after treatment with terpenes. Control: 1% TST (w/w) in propylene glycol (n=6).

Spectroscopic studies were considered to evaluate the effect of absorption enhancers on the biophysical properties of the stratum corneum (SC), particularly interactions with lipid or protein domains. The FT-IR spectrum of untreated SC (control) showed various bands associated to molecular vibrations of lipids (nonpolar) and proteins (polar) (Fig. 5). C-H stretching vibrations primarily associated with the lipid alkyl chains are represented by bands between 2,800 and 3,000 cm<sup>-1</sup>. The bands at 2,920 and 2,850 cm<sup>-1</sup>, in turn, are associated to the asymmetric  $CH_2$  and symmetric  $CH_2$ vibrations of long chain hydrocarbons of lipids, respectively. Asymmetric and symmetric  $CH_3$  vibrations appear at 2,955 and 2,870 cm<sup>-1</sup>, respectively [19]. Reduction in the intensity or significant shifts in these bands suggest disorder of the hydrocarbon chains or SC lipid domains [20]. Conversely, strong bands at 1,650 and 1,550 cm<sup>-1</sup> are related to the amide I and amide II stretching vibrations of SC proteins, respectively [21]. The amide I bands arises from C=O stretching vibration and the amide II bands from C-N bending vibration. Both absorption bands are sensitive to hydrogen bonding interactions [22]. The amide I band represents various secondary structures of keratin.

The absorption band between 1,100 and 900 cm<sup>-1</sup> is attributed to stretching vibrations of alcoholic C-O in C-O-H bonds as well as asymmetric and symmetric C-O-C bonds of propylene glycol [23]. A strong absorption band was identified close to 1,000 cm<sup>-1</sup>. The presence of this agent in SC samples containing terpenes led to a reduction in the intensity of the bands in region between 3,000 and 3,600 cm<sup>-1</sup>, which may be associated to its dehydration effect [24]. PG reduces the bound water content due to changes in the protein domains [25]. It probably works by solvating  $\alpha$ -keratin and occupying hydrogen-bonding sites [26]. In fact, SC samples treated only with PG showed broader and lower band intensity at 1,650 and 1,550 cm<sup>-1</sup> (amide region) compared to the control (untreated tissue), which suggest more disorder in protein domains. Menthol, in turn, resulted in less disorder in polar domains than carvacrol and  $\alpha$ -bisabolol (carvacrol appears to have been the most effective). Unlike the SC sample containing menthol, the treatment with carvacrol and  $\alpha$ -bisabolol led to broader and lower-intensity absorption bands at 2,920 and 2,850 cm<sup>-1</sup>. This event indicates an increased translational movement or mobility of lipid acyl chains [24]. In summary, more changes in skin were showed after treatment with carvacrol.



**Figure 5.** FT-IR spectra of untreated stratum corneum (i) and tissue treated only with propylene glycol (ii), TST in propylene glycol (iii); TST in propylene glycol and carvacrol (iv); TST in propylene glycol and  $\alpha$ -bisabolol (v) and TST in propylene glycol and menthol (vi). (n = 2).

# 3.3 DSC analysis

DSC thermograms of SC treated with the tested absorption enhancers were evaluated by comparing  $T_m$  and  $\Delta H$  of endotherms. The shift in  $T_m$  to a lower temperature indicates a disruption of the lipid bilayer whereas the reduction in  $\Delta H$ values suggest fluidization of lipid bilayers [27].

Lipids extracted from the SC intercellular region show a single transition close to 65 °C [28]. As the SC treated with different solutions (PG alone or in presence of absorption enhancers) was dried with paper towel at room conditions before DSC analysis, extracted intercellular lipids were not identified in endotherms. In other words, the extracted lipids present in liquid medium are removed after the drying step and the amount of lipid available on the tissue would be not sufficient to generate an endotherm event (or significant signal in DSC analysis). This procedure was considered to minimize the PG effect in thermal analysis studies. A slight perturbation in the baseline was found for SC treated with carvacrol at this temperature, but even more analyses should be performed to confirm this assumption.

The thermal transition close to 80 °C is found when lipids passing from the gel to the liquid-crystalline state. In this situation, lipids are associated with proteins (not more in a free form), probably as a lipid-protein complex [29]. Protein-lipid combinations often result in greater lipid ordering, which shift  $T_m$  to a higher temperature value compared to alone lipids [28]. SC treated with menthol and carvacrol showed a slight reduction in  $T_m$  value when compared to the control, suggesting a greater disruption of the lipid bilayer. Carvacrol and  $\alpha$ -bisabolol-treated SC presented lower  $\Delta$ H values than menthol-treated SC.

Transitions at temperatures above 90 °C are often related to irreversible protein denaturation and have been detected in the delipidized SC matrices [30,31]. Therefore, this event could be detected in tissue samples where lipid extraction is found. SC samples treated with propylene glycol presented an event of protein denaturation close to 150 °C, except the tissue that included menthol as absorption enhancer. Endothermic events in this temperature were more evident for the SC sample prepared with carvacrol and  $\alpha$ -bisabolol, which may be associated with an increased delipidization provided by these absorption enhancers. As suggested in the FT-IR analyzes, both carvacrol and  $\alpha$ -bisabolol interact more strongly with lipid domains than menthol. In addition, changes in hydration degree may affect relative peak heights. High temperature transitions (T>90 °C) are much more emphasized in the less hydrated lipids [31]. PG would reduce the SC hydration, making these endotherm events more evident. For the control, lipid-protein interactions are preserved and thus endotherm events associated with proteins (e.g., keratin) are not observed.



**Figure 6.** Thermograms of untreated stratum corneum (i), and stratum corneum treated with TST in propylene glycol (ii); TST in propylene glycol and carvacrol (iii); TST in propylene glycol and  $\alpha$ -bisabolol (iv) and TST in propylene glycol and menthol (v). (n = 2).

#### 4. DISCUSSION

Natural terpenes have attracted attention during the development of new pharmaceutical formulations due to permeation enhancement effect and low skin irritation. These agents have shown to be effective for both lipophilic and hydrophilic drug groups. They can fluidize and/or extract the SC lipids, weakening the skin barrier properties. This compound class is also included in the list of Generally Recognized As Safe (GRAS) agents issued by the US Food and Drug Administration [9].

Although the use of the menthol as the chemical absorption enhancer (CAE) in TST-based transdermal systems is already reported in the literature [32], the other two terpenes selected in this study had not yet been considered in transdermal systems with this drug. In a previous study carried out by our research team, menthol showed to be less active than the limonene monoterpene in enhancing drug permeation. Differences in degree of disorder in lipid bilayers would explain these findings [27]. For this reason, in this new study, a comparison among three terpenes was considered taking in account

the action differences. In this same way, menthol showed to be the least effective terpene as CAE. An enhancement ratio (ER) of 2.52 was obtained in our study, which is in agreement with the study of Kaplun-Frischoff and coworkers (1997). The authors found a 2.6-fold increase in the flux value of TST through skin after a treatment with menthol. The formation of a eutectic mixture between TST and menthol as well as changes in barrier properties could explain these results [32]. Although carvacrol and menthol are structurally similar and present quite similar physicochemical properties (boiling point, log P and molecular weight; Table 2), different absorption rates for the TST were found after treatment with these CAEs. Spectroscopic and thermal analyses suggest that carvacrol showed greater affinity for lipid domains than menthol. As TST is preferentialy transported by the transcelullar/intercellular passive route due to its lipophilic nature [33,34], CAEs that modify these domains are more effective in increasing its permeation flux. The presence of the aromatic ring and phenolic OH in carvacrol also contribute to a greater chemical reactivity than menthol, which present a hydroxyl group linked to the cycloalkane (a less reactive chemical group). Therefore, a greater interaction with organic molecules (skin constituents) was expected for the carvacrol.

A similar interaction/specificity for skin lipid and protein domains was observed for carvacrol and  $\alpha$ -bisabolol in spectroscopic and thermal analyses. On the other hand, the permeation flux of TST was higher in the presence of the carvacrol. A detailed analysis of the physicochemical properties of terpenes can contribute to explain these differences.  $\alpha$ -bisabolol presents a higher molar mass and log P. The high log P value leads to a preferential interaction with the most superficial layers of the skin and its higher molar mass compared to carvacrol reduces the diffusion through the skin. In fact, TST presented a slower diffusion (lower lag time) through skin in the presence of  $\alpha$ bisabolol. Furthermore,  $\alpha$ -bisabolol has a higher melting point and boiling point than carvacrol. Higher melting and boiling points indicate stronger noncovalent intermolecular forces. Consequently, lower degree of interaction with other molecules (e.g., skin constituents) is expected. The boiling point of a terpene have been inversely related to its skin penetration enhancement effects [9]. Therefore, interactions among competitive hydrogen bonding of terpenes and skin ceramides are facilitated when the terpene presents a low boiling point.
Terpene	Classification	Molar mass	Melting	Boiling	Log P
		(g mol <sup>-1</sup> )	point (°C)	point (°C)	
α-bisabolol	Sesquiterpene	222.37	25	314.5	5.07
Carvacrol	Monoterpene	150.22	3.5	237.7	3.28
Menthol	Monoterpene	156.27	43	215.4	3.20

Table 2. Classification and properties of terpenes.

Chen *et al.* (2016) and Kang *et al.* (2013)

Concerning the TST retention data, the greater affinity of the hormone for skin lipophilic structures can explain its higher retention in the epidermis compared to the dermis layer for all treatments. The presence of androgen receptors in the skin structures such as sebocytes, hair follicle, dermal papilla and keratinocytes [36] may contribute to an increased concentration of TST in the superficial layers.

Terpenes have also been considered in permeation studies with other hormones. Songkro et al. (2009), for example, evaluated the impact of carvone, carvacrol and menthol on the permeation rate of luteinizing hormone releasing hormone (LHRH) through newborn pig. Unlike the TST, LHRH is a hydrophilic molecule with a large molecular weight (1,182 g mol<sup>-1</sup>), which limits its transport via the passive route. Terpenes were able to improve the cutaneous absorption of LHRH. Similar to our study, carvacrol was also the most effective CAE [37].

In addition to the effect of enhancing percutaneous absorption, the selection of carvacrol for transdermal formulations with TST leads to several other advantages various. In assays with rats, carvacrol demonstrated a significant antioxidant effect and improved sperm quality parameters such as motility, concentration, abnormal spermatozoa and live-dead sperm ratio [38]. This effect is particularly relevant in the case of hypogonadism. Due to the antibacterial and antifungal properties [39], carvacrol also could play a role as a formulation preservative. In this context, the irritating synthetic preservatives could be avoided or their use concentration could be reduced. Another interesting aspect is that this terpene does not affect the metabolization of TST via CYP3A4 [40], a common problem when natural materials are co-administered with synthetic drugs.

## **5. CONCLUSION**

The terpenes carvacrol,  $\alpha$ -bisabolol and menthol showed to be effective in enhancing the permeation of TST through human skin, resulting in ER values of 7.61, 5.10 and 2.52, respectively. Carvacrol interacts more strongly with lipid domains than other terpenes, which result in a greater disorder at these sites. As the TST is preferentially transported through the skin by the passive diffusion, a greater lipid disorder has direct impact on the permeation flux. In addition to a permeationenhancing effect, carvacrol is non-irritating, presents antioxidant, antifungal and antibacterial effect and may improve sperm quality parameters. Taken together, these aspects suggest that this terpene is an ideal candidate for transdermal delivery systems containing TST.

### **DECLARATION OF COMPETING INTEREST**

The authors declare no conflict of interest.

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# 5. CAPÍTULO IV – UMA ANÁLISE COMPARATIVA DE MODELOS DE PELES BIOLÓGICAS E SINTÉTICAS PARA ESTUDOS DE TRANSPORTE DE FÁRMACOS

Este capítulo trata da utilização de modelos alternativos de peles em estudos de permeação de fármacos. Maior ênfase foi dada aos modelos de pele reconstruída e os de pele sintética, mas a pele de animais também foi incluída na análise. Uma maneira racional de comparar o desempenho de diferentes modelos biológicos e não biológicos de pele é selecionar estudos de permeação realizados com moléculas similares. Por este motivo, testosterona e cafeína foram selecionados como modelo de fármaco lipofílico e hidrofílico, respectivamente. O produto deste capítulo é um artigo científico, que será submetido para apreciação no periódico *International Journal of Pharmaceutics Sciences*.

# 5.1. ARTIGO CIENTÍFICO III

# A COMPARATIVE ANALYSIS OF BIOLOGICAL AND SYNTHETIC SKIN MODELS FOR DRUG TRANSPORT STUDIES

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#### ABSTRACT

Ethical restrictions as well as practical or economic issues related to use of animal and human skin has been the main reason for the increase in the number of investigations with alternative models. Reconstructed skin models, for example, have been useful to evaluate the in vitro toxicity of compounds; however, these models usually overestimate the amount of drug permeated due to impaired barrier properties. In this review, two compounds presenting different physicochemical properties were selected to compare the performance of synthetic and biological skin models in transport studies. Advantages and limitations of each skin model are discussed in detail. Although synthetic and reconstructed skin models have shown to be useful in formulation optimization step, they present many limitations: (1) more permeation to drugs; (2) lack of follicular transport; (3) no metabolism in synthetic membranes; (4) differences in terms of lipid organization; (5) more affected by formulation constituents. Therefore, animal and human tissues should still be prioritized in drug transport studies until new advanced alternative models are purposed. Investigations of the impact of culture conditions on skin formation, in turn, bring perspectives related to development of unhealthy/injured skin models (an aspect that still deserves attention).

Keywords: synthetic skin; skin equivalents; permeation; testosterone; caffeine.

#### 1. INTRODUCTION

The skin is a heterogeneous organ that present several drug administration sites and thus it may be targeted to treat different diseases (Abd et al., 2016). It presents various advantages over conventional administration routes (e.g., oral and intravenous). It avoids the first-pass metabolism, protects compounds from the degradation in gastrointestinal tract, allows a sustained release of drugs for extended periods due to a "reservoir" effect, may reduce toxicity and local irritation and still provides high patient compliance to treatment as it is a non-invasive administration route (Haq et al., 2018; Roberts et al., 2021; Yu et al., 2021). For this reason, pharmaceutical companies have often performed investigations aiming to reposition oral drugs to the transdermal route.

Cutaneous permeation assays represent a key point during the development of new cosmetic or transdermal products for human use (Planz et al., 2016; Yun et al., 2018). Appropriate quality control should also be performed to ensure batch-tobatch uniformity in the pharmaceutical industry. Clinical bioequivalence studies have been particularly considered in performance evaluation of new topical or transdermal products to ensure that an enough amount of drug achieves the target; however, these assays are highly costly, associated with high intra-subject variability and little sensitive to discriminate formulations presenting different composition (Abd et al., 2016).

In vitro/ex vivo permeation studies emerge as an alternative to evaluate the bioequivalence of transdermal dosage forms and for mechanistic investigations (Raney et al., 2015). Franz-type diffusion cells has been the most used *ex vivo* model for skin permeation studies, in which different membrane types (human, animal and synthetic) are sandwiched between a donor compartment and a receptor compartment (Haq et al., 2018; Küchler et al., 2013). Assays with human skin represent the gold standard; however, ethical, practical or economic reasons as well as tissue supply have limited its use (Esposto Biondo et al., 2021). Regulatory agencies and consumers have also pressured the industry to avoid experimentation with animal tissues in last decade, particularly in the cosmetic field (Abd et al., 2016). In this context, cell-culture skin models, synthetic or natural biomaterial-

based scaffolds have attracted the attention of researchers (Flaten et al., 2015). Although different reconstructed skin models have been already approved by OECD for toxicity assays (OECD, 2004a, 2004b), the use of these membrane models in drug absorption studies is still challenging as a barrier function similar to *in vivo* skin should be found (Van Gele et al., 2011). The advantages and disadvantages/limitations of each skin model are summarized in Table 1.

Model skin	Advantages	Limitations
Human	-Gold standard for permeation studies	-Ethical considerations; -Inter-individual variability; -Limited tissue availability.
Animal	<ul> <li>-Animal models such as porcine skin present barrier properties similar to human skin;</li> <li>-Some animals (rat/mouse) present hairless species available to investigate the role of hair follicles in drug permeation.</li> </ul>	<ul> <li>-Rat/mouse model require ethical permission;</li> <li>-Large variability interspecies (e.g., Skin thickness and hair follicle density vary depending on the animal model).</li> </ul>
Artificial/ synthetic membranes	<ul> <li>-High correlation with <i>in-silico</i> models as mimic transcellular transport of drugs;</li> <li>-Application for screening purposes in early phases of drug development;</li> <li>-Reproducible;</li> <li>-Assays may be performed in a short time (short lag time);</li> <li>-Low cost.</li> </ul>	<ul> <li>Poor barrier properties which result in increased absorption rate;</li> <li>No effective for drugs that suffer skin metabolism;</li> <li>No appendageal transport;</li> <li>Model is highly sensitive to the incorporation of chemical absorption enhancers and surfactants.</li> </ul>
Reconstructed skin model (cell cultures)	<ul> <li>-Are effective for mechanistic investigations (e.g., skin metabolism, transport type);</li> <li>-Models with and without stratum corneum can be prepared;</li> <li>-More sensitive to the incorporation of chemical absorption enhancers and surfactants than human and animal tissues;</li> <li>-Reproducible;</li> <li>-Assays may be performed in a short time (short lag time).</li> </ul>	-More permeable than human skin; -Time-consuming (long culture time); -No appendageal transport; -High cost.

**Table 1.** Advantages and limitations of different skin models.

A rational way to compare the performance of different biological and nonbiological skin models is to select permeation studies performed with the same drug model. Therefore, *in vitro-ex vivo-in vivo* correlations can be more easily established. Caffeine (CF) is often used in skin permeation assays as a highly watersoluble compound whereas the testosterone (TST) as a lipophilic drug (Abd et al., 2016; Neupane et al., 2020; Schreiber et al., 2005; van de Sandt et al., 2004). The log P values of CF (MM=194.19 g mol<sup>-1</sup>) and TST (MM=288.4 g mol<sup>-1</sup>) are -0.07 and 3.32, respectively. At 25°C, the solubility of CF and TST in water is  $2.16 \times 10^4$  and 23.4 mg L<sup>-1</sup>, respectively (Im et al., 2021). These different physicochemical properties impact on transport mechanisms through the skin. Hydrophobic drugs such as TST are preferentially transported through the "intracellular" pathway as they need to cross the lipid bilayer whereas "intercellular" pathway is preferentially used for hydrophilic compounds as CF. In this last situation, molecule diffusion around the cells occurs in a tortuous manner (Supe e Takudage, 2021) and a lipid disorder or fluidization is crucial for an efficient transport (Neupane et al., 2020). The "transappendageal" route (e.g. hair follicle, sebaceous glands), in turn, is more used for polar molecules that do not freely cross the stratum corneum (Dominguez et al., 2017).

CF and TST are also recommended as reference compounds by the OECD to validate the *in vitro/ex vivo* model used in investigation of cutaneous permeation of new products or drugs (OECD, 2000a, 2000b, 2000c). Therefore, the main objective of this literature review was comparing the performance of different biological and non-biologic membranes in cutaneous permeations studies considering these two drugs; understanding the advantages and limitations of each skin model. Studies performed with formulations of these compounds or free drug in presence of chemical absorption enhancers were not considered for this review.

#### 2. HUMAN SKIN STRUCTURE

The human skin structure is composed of three main layers: epidermis, dermis and subcutaneous tissues (Figure 1). The thickness and composition of the skin could vary with age, anatomical site, race and gender (Roberts et al., 2021).



Figure 1. Skin structure (left) and main layers of the epidermis (right).

The major drug diffusion-limiting barrier in skin is represented by *stratum corneum* (SC), which is the outermost layer of epidermis. It is a highly hydrophobic layer of 10-20  $\mu$ m, composed of 10-25 layers of stacked corneocytes, which are anucleated and devitalized cells. Corneocytes are embedded in a structurally well-organized matrix of intercellular lipids, which is referred as a "bricks-in-mortar" structure (Sjövall et al., 2018). The SC lipid matrix is dominated by three lipid classes (cholesterol, free fatty acids and ceramides), which are synthesized by lamellar bodies of keratinocytes during the differentiation process to corneocytes (Évora et al., 2021).

The viable epidermis (50-100  $\mu$ m thick), underlying the SC, is a stratified epidermis composed of basal, spinous and granular cell layers. Each viable epidermis layer is defined by position, shape, morphology and state of differentiation of keratinocytes (Bouwstra, 2003). Keratinocytes proliferate in the basal layer and then suffer a differentiation process while migrating to the skin outermost layers (Évora et al., 2021). Many proteins and lipids are synthesized and expressed during keratinocyte maturation. Expressive changes in keratinocyte structure are found in the last differentiation stage, which lead to a programmed cell death process known as "cornification". In other words, viable cells are converted into chemically and physically resistant cornified squames called "corneocytes" (Bouwstra, 2003).

The dermis is the skin deeper layer with 1-2 mm thick, which allows the drug access to circulation system due to presence of blood vessels (Prausnitz e Langer, 2008). Unlike the epidermis, which is made up of dense keratinocytes, the dermis is mostly composed of an acellular component, the extracellular matrix (ECM). This structural feature explains why highly lipophilic drugs difficulty crossing the dermis. Collagen fibers, particularly collagen type I and III, represent the major constituent of the ECM (75% of the dry weight of skin). These proteins provide tensile strength and elasticity to the skin. Elastic fiber network, in turn, allows the skin return to its original shape after stretching or deformation. Although proteoglycans and glycosaminoglycans represent only 0.2% of the dry weight of the dermis, they are able to absorb water up to 1,000 times their weight. In this way, they play an important physiological role in maintaining skin hydration. The main

resident cell type of the dermis are the fibroblasts, which are involved in the production of ECM constituents (Shin et al., 2019).

The hypodermis or subcutaneous tissue, the innermost layer of the skin, is rich in adipocytes and play an important role in thermoregulation and maintenance of the mechanical properties. In addition, it has supported keratinocyte and fibroblast proliferation and regulate cycling of hair follicles (Zimoch et al., 2021). During drug transport assays using human or animal tissues, hypodermis is often removed to prevent drug absorption data from being underestimated.

Pilosebaceous units have been the main appendages associated with drug transport through the skin. In these structures, each hair follicle is coupled with one or more flask-shaped sebaceous glands. Once the density and size of hair follicles vary depending on the region of the body, this aspect should be carefully monitored during drug transport studies. The size of the follicular orifice and presence of sebum in follicular space also affect the absorbed drug amount (Verma et al., 2016).

#### 3. TRANSPORT STUDIES IN HUMAN SKIN

Although *ex vivo* assays cannot fully reproduce *in vivo* conditions, particularly with respect to metabolism, distribution, and blood supply, an important advantage of this model is that experimental conditions can be controlled more precisely (Brain et al., 2002). *Ex vivo-in vivo* correlations have been successfully achieved under rigorous or controlled experimental conditions, which is of particular interest to regulatory agencies seeking standardized or robust methods (Abd et al., 2016).

Diffusion cells, particularly the Franz-type cells, represent the most traditional method to evaluate the *ex vivo* permeation of drugs, which consists of donor and acceptor chamber sandwiched by a membrane/tissue (Moser et al., 2001). Excised human skin is often obtained from plastic surgery or cadavers. In both situations, appropriate ethical approval is required to use the tissue. Abdominal, breast, or back skin are routinely used because the larger tissue area compared to other body regions (Abd et al., 2016).

"Full thickness skin" is prepared by removal of subcutaneous fat, separating dermis and epidermis (Cross et al., 2003). The use of membranes with a thickness of more than 1.0 mm (epidermis and dermis) is not recommended once the absorption of lipophilic drugs may be reduced or even avoided (van de Sandt et al., 2004). For this reason, dermatomes have been routinely considered to reduce the thickness tissue to approximately 500-750  $\mu$ m (Abd et al., 2016). Manual tissue preparation can also be performed, but their thickness should be carefully monitored to reduce experimental variability. Once the skin is obtained, its integrity may be evaluated through transepithelial electrical resistance, transepidermal water loss and permeation of tritiated water measurements or histological evaluation (Esposto Biondo et al., 2021).

The *in vitro* absorption of CF and TST through human skin was evaluated in a multicenter comparison study that considered data from the 9 laboratories. A common protocol was defined as well as three independent assays for each tested compound. The absorption of CF and TST through human skin was similar, presenting mean maximum absorption rates of  $2.24 \pm 1.43$  and  $1.63 \pm 1.94$  $\mu$ g/cm<sup>2</sup>/h, respectively. In 7 out of 9 laboratories, the maximum absorption rates of CF were ranked higher than TST. Although laboratories used different diffusion cell apparatus, this variable did not affect the transport rate of these compounds significantly. On the other hand, skin thickness showed to affect the absorption of TST. Thinner and dermatomed tissues resulted in a maximum absorption rate. This parameter did not affect the permeation rate of CF significantly (van de Sandt et al., 2004). Atrux-Tallau et al. (2007) also found a similar permeation flux for CF through the dermatomed human skin and heat-separated epidermal tissues (Atrux-Tallau et al., 2007).

In another study performed with CF and TST in a flow-through cells and human skin, similar findings related to impact of skin thickness on permeation of TST were found. No significant differences in maximum permeation flux values for CF were observed with thickness variations. TST, in turn, showed a permeation flux approximately 10-fold higher in the thinner skin (Wilkinson et al., 2006). In summary, these literature studies sustain the hypothesis that skin thickness needs to be reduced prior to permeation assays with lipophilic drugs to obtain results closer to physiological situation.

#### 4. ANIMAL SKIN MODELS

Animal skin models differs from the human skin with respect to number of hair follices, lipid content and composition as well as anatomical or morphological aspects (Van Gele et al., 2011) (Table 2). The porcine skin has been considered the main substitute for human skin due to anatomical, physiological and biochemical similarity (Barbero e Frasch, 2009; Simon e Maibach, 2000). Both tissues are characterized by a thick epidermis, a dermis presenting a well-differentiated papillary body and a large content of elastic tissue. Similar hair follicle density and protein fractions in SC are also found (Jung e Maibach, 2015). The porcine skin should not be scalded or flamed after animal sacrificing because this procedure compromises the tissue integrity, affecting drug transport rate (Flaten et al., 2015).

Rodent skin has also been widely used in ex vivo permeation studies due to the availability, simple handling and relatively low cost. On the other hand, studies have suggested that this tissue is more permeable than human skin (Barber et al., 1992; Schmook et al., 2001; Van Ravenzwaay e Leibold, 2004), wich may be atributted to anatomical or morphology differences. In rat skin, the most used rodent model, SC and epidermis are thinner, the hair follicle density is higher, intercellular lipid composition of the SC is different and number of corneocyte layers is lower (Capt et al., 2007).

Animal model/structural or biochemical chemistry	Human skin	Porcine skin	Rat skin	Mouse skin	Reference
Stratum corneum thickness (µm)	18.2	17.5	15.4	8.8	(Sato et al., 1991)
Epidermis thickness (µm)	51.2	50.7	23.8	18.0	(Sato et al., 1991)
Hair density (per cm <sup>2</sup> )	29	22	1,598	5,045	(Mangelsdorf et al., 2014)
Total lipid content in epidermis (% of dry weight)	±10	$\pm 8$	±20	ND	(Gray e Yardley, 1975)
Distribution of non-polar lipids in epidermis	35.5	30.2	64.2	ND	(Gray e Yardley, 1975)

**Table 2.** Comparison between human and animal skin structure.

(% of total lipids)					
Cholesterol content in epidermis (% of total nonpolar lipids)	25	30	18	ND	(Gray e Yardley, 1975)
Ceramide content in epidermis (% of total nonpolar lipids)	3	3	Not detected	ND	(Gray e Yardley, 1975)
SC lipid arrangement	Denser orthorhombic lattice	Hexagonal lattice	ND	ND	(Caussin et al., 2008)

\*All parameters were represented by mean values. ND = not determined.

In a study performed by van de Sandt et al. (2004), the permeation of TST and CF in both human and rat tissues were compared. No differences were observed between the mean absorption through human and rat skin for TST. On the other hand, the maximum absorption rate and the total penetration of CF through rat skin were clearly higher than the mean value obtained in human skin (van de Sandt et al., 2004).

A disturbed skin barrier model was also purposed by Schlupp et al. (2014) aiming to obtain more realistic information in situations in that stratum corneum is injured. Cutaneous permeation of CF and TST was evaluated in tape-stripped and abraded skin compared to untreated skin. Porcine skin was selected for the assays. Permeation or retention enhancement was higher for CF than for the TST in the modified skin. TST permeation, for example, increased by 1.9- and 2.1-fold through tape-stripped and abraded skin, respectively. No significant differences in absorption rate was found between the new abrasion method with an aluminumand the tape-stripping traditional method. coated sponge The *stratum* corneum showed to be less restrictive for lipophilic compounds than for hydrophilic compounds (Schlupp et al., 2014).

Given that rat skin present impaired barrier properties, Netzlaff et al. (2006) included bovine udder skin in a new transport study comparing different animal models. The K<sub>p</sub> values of CF through the human, porcine and bovine skin were 0.42, 1.54 and 8.27 ( $x10^{-7}$ ) cm/s, respectively. For TST, these values were 2.31, 1.29 and 5.42 ( $x10^{-7}$ ) cm/s in human, porcine and bovine skin, respectively. Bovine udder skin provided a higher permeation for the tested molecules than human and porcine skin. The tissue constitution was analyzed to make clearer the result understanding. A negative correlation between fatty acid/triglyceride contents in tissues and

permeation rate was found. Bovine udder skin is characterized by a lower content of fatty acids and triglycerides than porcine and human skin, which resulted in higher drug transport. This hypothesis is also supported by the fact that a greater permeation rate was observed for CF than TST. These tissues presented a similar cholesterol composition and epidermal thickness. On the other hand, bovine udder skin presented a much higher density of hair follicles than human skin (about  $6/\text{cm}^2$ ) and porcine skin (30–36 follicles/cm<sup>2</sup>). This characteristic could also contribute to explain the higher permeation rate of CF in bovine udder skin (Netzlaff et al., 2006).

Although human skin represents the gold standard model for permeation studies, this tissue may present limitations during investigation of transfollicular transport of drugs. Interfollicular fibers (collagen and elastin) undergo contraction after skin removal. When the skin is stretched for use in the transport assays, the fibers are also stretched, but the network of fibers close to the hair follicles is not affected by this stretching process. Therefore, hair follicles remain in the contracted state and are only available to a limited level for drug permeation. In contrast, porcine ear skin has the advantage of having an intact skin barrier on the upper face while it is still attached to the cartilage. No tissue contraction occurs during skin removal as in the case of excised human skin. On the other hand, the holes of pig ear hair follicles are approximately three times larger in diameter than those of human hair follicles (Lademann et al., 2010).

In addition to comparing drug permeation in different animal skin models with variation in the follicle density, hair follicle-blocking agents are also useful in mechanistic studies. Porcine skin treated with nail polish was used to investigate the contribution of transfollicular pathway on the transport of CF and TST. Although many studies suggest that transfollicular permeation is preferentially used for highly hydrophilic and high molecular weight molecules (Mitragotri, 2003; Ogiso et al., 2002), the presence of the hair follicle-blocking agent also impacted on the transport rate of TST. In untreated porcine skin, TST presented a K<sub>p</sub> value of 4.60 x 10<sup>-8</sup> cm/s. When hair follicles were individually closed or skin was completely covered, this parameter was reduced to 4.40 and  $0.32 \times 10^{-8}$  cm/s, respectively (Stahl et al., 2012). The presence of androgen receptors in follicles could explain the higher affinity of TST for these sites compared to other lipophilic drugs, but further investigations are still needed.

#### 5. SYNTHETIC/ARTIFICIAL MEMBRANES

Although human excised skin is regarded as the gold standard for permeation studies, the relative difficulty of obtaining human skin samples, ethical aspects and the high variability of sources have contributed to the use of artificial membranes (Hadgraft, 2004). In this way, artificial membranes composed of cellulose acetate, dimethylpolysiloxane, hydrocarbons, long-chain alcohols, isopropyl myristate or skin lipids may be considered (Ottaviani et al., 2006) due to the easy availability and result reproducibility (Abd et al., 2016). They are useful to understand the organization of the SC and the impact of each component on the barrier properties (Janůšová et al., 2011; Küchler et al., 2013). Unlike the cellulose acetate membranes, hydrophobic membranes may mimic the drug partitioning into lipids (Jetzer et al., 1986). Hydrophilic membranes, in turn, may be useful as a damaged skin model, where the relatively aqueous dermis is the only remaining barrier (Houk and Guy, 1988).

These membranes may be placed in the interface of the diffusion chambers or prepared directly on the PAMPA (Parallel Artificial Membrane Permeability Assay) apparatus (Figure 2). This latter assay may be considered for rapid screening of passively transported drugs through skin. An artificial liquid membrane separates the donor and receptor compartments in 96-well plates. The receptor compartment consists of a buffer solution, with or without a solubilizing agent, and the donor compartment contains the drug solution (Neupane et al., 2020).



Figure 2. Schematic representation of PAMPA model.

Ottaviani et al. (2006) evaluated the transport rate of CF, TST and other drugs by considering three different membrane compositions (100% silicone; 100% isopropyl myristate-IPM and 70% silicone/30% isopropyl myristate) in PAMPA assay. The liquid silicone membrane allowed a sufficient permeation of TST whereas no permeation was observed for the CF. The high hydrophobicity of this membrane was responsible for these findings. As the SC consists of two alternating lipophilic and hydrophilic layers, IMP was immobilized on hydrophobic PVDF filters to evaluate the transport rate of drugs. Although this membrane type has improved permeation rate of CF and TST, low correlation with data in human skin was found (conclusion based on all tested molecules). Moreover, the differences in K<sub>p</sub> values of molecules with different log P values were reduced, i.e., assay showed to be less discriminatory. Finally, the authors combined both silicone and IMP in a same membrane, achieving higher correlation with data in human skin. IMP would play a role as a H-bond acceptor in the membrane, similarly to proteins of stratum corneum (Ottaviani et al., 2006).

Skin constituents such as cholesterol, free fatty acids, and the ceramides analogues (namely synthetic certramides, long-chain tartaric acid diamides) were also selected to prepare artificial membranes for PAMPA assay. These new certramides are structurally similar to the natural ceramides with comparable molar mass, H-bond donor/acceptors and high lipophilicity. CF, TST and other 20 molecules were considered as drug models for transport studies. A higher correlation between permeation rate of TST in PAMPA assay and human skin was found when compared to CF. The effect of PEG 400, a known solubilizing agent and chemical penetration enhancer, on the permeation rate of these molecules was also evaluated. Interestingly, an increased transport rate in presence of this agent was found. PEG 400 would be able to dissolve membrane constituents, reducing the barrier properties (Sinkó et al., 2012). This finding limits the application of PAMPA-skin assay for mechanistic studies with new chemical penetration enhancer or formulations.

The permeation of drugs through artificial membranes (isopropyl myristate, certramides and Strat-M<sup>TM</sup>) was also compared to that of porcine skin. Strat-M<sup>TM</sup> are commercial membranes composed of multiple layers of polyether sulfone that result in a morphology similar to human skin with a very tight surface layer. In

addition to CF and TST, other 4 compounds were selected for permeation studies. Saturated and unsaturated concentrations of these compounds as well as different vehicles were tested. A stronger correlation between membrane and diffusion cell results was obtained when only saturated concentrations were considered ( $r^2 = 0.60$ , 0.63 and 0.66 for the Strat-M<sup>TM</sup>, certramide and isopropyl myristate membranes, respectively). The stronger *in vivo-ex vivo* correlation found for isopropyl myristate membranes may be explained for their amphiphilic nature (SC presents both polar and nonpolar sites, which are represented by proteins and lipids, respectively). Therefore, the authors suggest that artificial membranes in PAMPA apparatus could be useful in an initial screening of compounds (Karadzovska e Riviere, 2013).

In summary, synthetic membranes could be considered as skin models when (1) the stratum corneum represents the major barrier resistance to transport; (2) the active compound is metabolically inert and not specifically bound in viable skin (this is not the case of TST, which is able to interact with androgen receptors in skin); (3) formulations without chemical penetration enhancers able to interact with the membrane are tested; and (4) *in vivo* assays of similar design can be easily performed for result validation.

#### 6. RECONSTRUCTED SKIN MODELS

Reconstructed human skin models were developed in the 1980s to replace animal experimentation (Nakamura et al., 2018). In these skin models, layers of human cells are usually grown on a polymer matrix (e.g., collagen scaffolds) (Figure 3). Different cell types may be considered to achieve a structure of the desired composition and complexity. Reconstructed skin models may be designed to simulate only the epidermis or full-thickness skin (Flaten et al., 2015; Küchler et al., 2013).





**Figure 3.** Reconstructed human skin models for drug transport studies. In the model A (stratified epidermal equivalent), keratinocytes are seeded onto a collagen scaffold. In the model B (full-thickness skin equivalent), the collagen scaffold is previously populated with fibroblasts. The air-liquid interface is crucial for differentiation of keratinocyte in both models.

The commercial systems SkinEthic<sup>®</sup> (SkinEthics, France), epiCS<sup>®</sup> (CellSystems, Germany), and EpiDerm<sup>®</sup> (MatTek Corporation, USA) consist of epidermal keratinocytes cultured on polycarbonate membrane. EpiSkin<sup>®</sup> (L'Oréal, France), in turn, is composed of stratified differentiated human keratinocytes cultured on a dermal matrix (type I bovine collagen matrix with type IV human collagen film). In advanced skin models such as Phenion<sup>®</sup> (Henkel, Germany) and NeoDerm<sup>®</sup> (Tegoscience, Korea), human fibroblasts are initially cultivated on a collagen matrix, followed by deposition of an epidermal layer composed of human keratinocytes (Yun et al., 2018).

These skin models exhibit metabolic activity, an interesting advantage compared to the synthetic membranes (Küchler et al., 2013). Therefore, drug transport and metabolism may be evaluated with the same model. Interestingly, the esterase activity in reconstructed skin models has been correlated to that of native human skin (Gysler et al., 1999). Similar polar and nonpolar metabolites of TST were also found in a comparative study between a reconstructed skin model and native skin (Slivka, 1992). Ernesti et al. (1992) also found a similar metabolic profile when skin constructs composed of fibroblasts, keratinocytes and collagen were compared human skin. The percentages of TST metabolized to  $5\alpha$ -DHT and

androsterone was similar among the reconstructed skin model and leg skin (approximately 15%).

Although these models have already been approved by OECD for skin corrosivity (OECD, 2004a) and phototoxicity assays (OECD, 2004b), their application in drug absorption studies require previous validation with reference skin models (OECD, 2004c). A series of comparative studies has been performed in the last two decades considering this aspect (Table 3). Schreiber et al. (2005), for example, compared the cutaneous permeation of CF and TST through reconstructed epidermis models (SkinEthic<sup>®</sup> and EpiDerm<sup>®</sup>) with porcine and human skin, which serve as a reference. For both molecules, SkinEthic® was more permeable than EpiDerm<sup>®</sup>, followed by porcine skin and then human epidermis. K<sub>p</sub> values of CF through SkinEthic<sup>®</sup>, EpiDerm<sup>®</sup>, porcine skin and human epidermis were 777.6, 176.4, 31.7 and 7.6 x  $(10^{-5})$  cm/h, respectively. For TST, the same rank through different membranes was obtained, which was 212.4, 122.4, 11.5 and 9.4 x (10<sup>-4</sup>) cm/h, respectively. Less differences in the permeation rate between reconstructed epidermis and reference skin models were found for TST than CF. The lag time obtained from the reconstructed epidermis models was similar to that of human epidermis. The authors explained this finding based on thermal treatment of human tissues. Taken together, these results showed than reconstructed epidermis models were more permeable than reference models (Schreiber et al., 2005).

Schäfer-Korting et al. (2006) performed a similar comparative study, including EpiSkin<sup>®</sup> and bovine udder skin in this new analysis. A similar permeation of TST through the EpiDerm<sup>®</sup> and EpiSkin<sup>®</sup> model was found ( $K_p = 2.89$  versus 2.11 x 10<sup>-6</sup> cm/s, respectively), while the SkinEthic<sup>®</sup> model showed to be more permeable ( $K_p = 6 \times 10^{-6}$  cm/s). The  $K_p$  values of CF through the human epidermis, porcine skin and bovine udder skin were 0.42, 0.08 and 0.32 x (10<sup>-6</sup>) cm/s, respectively. For the CF, EpiSkin<sup>®</sup> model showed to be less permeable, followed by EpiSkin<sup>®</sup> and SkinEthic<sup>®</sup> ( $K_p = 0.24, 2.77$  and 3.63 x 10<sup>-6</sup> cm/s). The  $K_p$  values of TST through human epidermis, pig skin and bovine udder skin were 0.06, 0.07 and 0.63 x (10<sup>-6</sup>) cm/s, respectively. Overall, reconstructed epidermis models demonstrated a higher  $K_p$  and shorter lag time than animal and human tissues for both drugs. Lower free fatty acid and ceramide fractions or low expression of cytokines and growth factors associated to hyper-proliferation of the

epidermal cells could lead to an impaired barrier, explaining these findings (Ponec et al., 2002). Inter-laboratory variation was also evaluated (n=5). Low K<sub>p</sub> variation was found (9.3% for CF and almost zero for TST), suggesting that the standardized protocols could be efficiently transferred (Schäfer-Korting et al., 2006).

In a new study performed by the same research team involving ten different laboratories, CF, TST and seven other compounds presenting different physicochemical properties were tested in these skin models. Overall, the K<sub>p</sub> values were close to that of previous study. For TST, the SkinEthic<sup>®</sup> showed to be more permeable than the EpiDerm<sup>®</sup> and EpiSkin<sup>®</sup>. For the CF, SkinEthic<sup>®</sup> and EpiSkin<sup>®</sup> were more permeable. For both drugs, reconstructed epidermis models demonstrated higher K<sub>p</sub> values than human and animal tissues. The K<sub>p</sub> obtained for CF with human epidermis and pig skin model were higher than in the previous study. The authors attributed these differences to tissue preparation/origin. In summary, the permeation ranking of drugs was similar through all membranes; however, the reconstructed epidermis models also showed more reproducible results (Schäfer-Korting et al., 2008).

Drug transport studies have also been performed in full-thickness skin models (dermis and epidermis). In this study, the permeation of CF and TST through Phenion<sup>®</sup> full-thickness skin model was compared to that of porcine skin. As expected, Kp values of TST in Phenion<sup>®</sup> model were higher than porcine skin (1.58 vs. 0.08 x 10<sup>-6</sup> cm/s, respectively). For CF, this difference was even greater (6.7/6.73 vs. 0.03 x 10<sup>-6</sup> cm/s, respectively). When compared to EpiDerm, SkinEthic and EpiSkin, this model had a slightly weaker barrier for CF transport whereas it would retard the permeation of lipophilic compounds such as TST. In the lag time analysis, lower values were obtained with Phenion<sup>®</sup> model, which had already been observed with other reconstructed skin models. For TST, Phenion<sup>®</sup> model (0.56 h) resulted in lag time values even shorter than EpiSkin<sup>®</sup>, which was associated with structural differences. In Phenion<sup>®</sup> model, fibroblasts are embedded into loosely packed newly synthesized extracellular matrix fibers whereas keratinocytes are grown on a densely packed collagen layer in EpiSkin<sup>®</sup> (Ackermann et al., 2010).

All permeation studies mentioned in this review demonstrated that reconstructed skin models result in higher drug permeation rate and shorter lag time of CF and TST than reference models (human skin or porcine skin). This impaired barrier property could be justified by differences in lipid composition compared to native skin (Figure 4). The major skin barrier lipids are also produced in vitro; however, in different proportions than those found in native skin (Ponec et al., 2002). Incomplete ceramide profiles and reduced content of fatty acids are found in the EpidDerm<sup>®</sup>, SkinEthic<sup>®</sup> and EpiSkin models<sup>®</sup>. In fact, the content of free fatty acids is very low in EpiDerm<sup>®</sup> and Episkin<sup>®</sup> models (approximately 2%) and moderate in SkinEthic cultures (approximately 6%) as compared with native epidermis (10%). Reconstructed skin models present low levels of ceramides 5 and 6 and the absence of ceramide 7 (more polar ceramides). The SkinEthic<sup>®</sup> model, which presented higher transport rate for both drugs, is characterized by a lower total lipid content than other reconstructed skin models (0.29 vs. 0.438 and 0.526 mg/cm<sup>2</sup> in EpiSkin<sup>®</sup> and EpiDerm<sup>®</sup>, respectively) (Ponec et al., 2000). Consequently, this skin model would present weaker barrier properties. Another hypothesis for explaining these differences between reconstructed skin models and native skin is the lipid organization in stratum corneum. In native SC lipids, the lateral packing is predominantly orthorhombic while a hexagonal packing is found in reconstructed SC (Ponec et al., 2001).



**Figure 4.** Main classes of lipids present in SkinEthic<sup>®</sup>, EpiDerm<sup>®</sup>, EpiSkin<sup>®</sup> compared to native skin. Data were obtained from Ponec et al. (2002).

The role of the dermis in drug transport through reconstructed skin models is still not well defined as few studies with advanced skin models have been performed to date. In a study with the Phenion<sup>®</sup> model, the permeation of TST was more retarded. This finding was explained as it does not present subcutaneous vasculatures and its environment is more aqueous than native skin (Van Gele et al., 2011).

Another point worth to mention is that these models are not appropriate to study the appendageal transport of molecules. The absence of hair follicles or sweat glands target the drugs to inter- and transcellular pathways (Neupane et al., 2020). Skin appendages such as hair follicles and sebaceous glands have been considered an important transport route of hydrophilic molecules as CF (Otberg et al., 2008) and for steroids as TST (Hueber et al., 1994). This fact could also explain the differences in transport rate of CF and TST between reconstructed skin models and native skin.

Model skin	Permeation coefficient values (cm/s, ×10 <sup>-6</sup> )	Lag time (h)	Reference	
Testosterone				
Human epidermis (reference model)	0.26	0.03		
Porcine skin (reference model)	0.32	7.63	(Schreiber et al., 2005)	
EpiDerm®	3.40	0.00		
SkinEthic®	5.90	0.01		
Human epidermis (reference model)	0.42	1.03		
Porcine skin (reference model)	0.08	-0.13	(Sahöfan Kanting at	
Bovine udder skin (reference model)	0.32	1.19	al., 2006)	
EpiDerm®	2.89	0.00		
SkinEthic®	6.00	0.14		
EpiSkin®	2.11	0.93		
Human epidermis (reference model)	0.50	0.51		
Porcine skin (reference model)	0.25	6.66	(Schäfer-Korting et	
EpiDerm®	2.78	0.08	al., 2008)	
SkinEthic®	5.70	0.07		
EpiSkin®	2.10	0.52		
Porcine skin	0.08	5.60	(Ackermann et al.,	

**Table 3.** Permeation parameters (permeation coefficient and lag time) for caffeine and testosterone obtained from reconstructed skin models and reference tissues.

(reference model)			2010)
Phenion <sup>®</sup>	1.58	0.56	
Caffeine			
Human epidermis (reference model)	0.02	0.00	
Porcine skin (reference model)	0.08	10.7	(Schreiber et al., 2005)
EpiDerm <sup>®</sup>	0.49	0.16	
SkinEthic®	2.16	0.00	
Human epidermis (reference model)	0.06	1.73	
Porcine skin (reference model)	0.07	3.92	(Schäfer-Korting et
Bovine udder skin	0.63	1.88	al., 2006)
EpiDerm®	0.24	0.33	
SkinEthic®	3.63	0.14	
EpiSkin <sup>®</sup>	2.77	1.04	
Human epidermis (reference model)	0.28	1.75	
Porcine skin (reference model)	0.31	6.92	(Schäfer-Korting et
EpiDerm <sup>®</sup>	0.59	0.35	al., 2008)
SkinEthic®	4.32	0.17	
EpiSkin <sup>®</sup>	2.94	0.81	
Porcine skin (reference model)	0.03	6.73	(Ackermann et al.,
Phenion <sup>®</sup> $(1)$	6.70	0.55	2010)
Phenion <sup>®</sup> (2)	6.73	0.43	

Permeation coefficient values obtained by Schreiber et al. (2005) were converted in "cm/s" to make the comparison clearer.

Efforts to improve reconstructed skin models have been constant. Zimoch et al. (2021) have recently reported that the presence of hypodermis in these skin models is crucial for keratinocyte maturation (Zimoch et al., 2021). We could hypothesize that the presence of adipocytes would contribute to improve the barrier properties of the stratum corneum; however, this additional layer could retard the amount of compound permeated. Transport assays in this skin model have not yet been considered.

Intra-specie variation or origin of cells used in reconstructed skin models, in turn, also seems to impact the state of cell differentiation and barrier properties. Girardeau-Hubert et al. (2019), for example, evaluated differences in reconstructed skin models with human keratinocytes and papillary fibroblasts (fullthickness skin model) when epidermal cells from African and Caucasian subjects were obtained. The terminal differentiation of Caucasian keratinocytes was improved in relation to African keratinocytes. Caucasian epidermis presented an upregulated signaling for ceramide metabolism and an up-regulated set of genes associated with late epidermal differentiation in contrast to African epidermis, which can be mainly related to proflaggrin synthesis and processing. Therefore, differences in epidermal morphogenesis and differentiation between keratinocytes of Caucasian and African skin types could result in variable skin behaviors (e.g., permeation rate).

#### 7. FINAL CONSIDERATIONS

Although reconstructed skin models have been shown to be effective in predicting the *in vitro* cytotoxicity of compounds, they result in an increased transport rate compared to human skin. Impaired barrier properties would contribute to explain these differences. In addition, these skin models predict only part of the absorption mechanisms, not including transfollicular transport. On the other hand, they are more advantageous than synthetic membranes and animal tissues in predicting metabolic processes due to expression of skin enzymes. In synthetic membranes, no enzyme activity is present. In animal models, tissue preparation is time consuming and may involve heat treatment, which affects the level of enzymatic activity in an uncontrolled way. The relationship between freezing time or temperature and enzyme level is still not very well known, which requires further investigations.

Efforts have been performed to achieve the same lipid composition among reconstructed and physiological skin; however, differences in lipid organization are still observed. Vehicles included in transdermal formulations may also impact on skin barrier integrity. Solvents such as ethanol or surfactants, which are often used to achieve the "sink" condition in permeation assays with lipophilic drugs, have their use limited in reconstructed skin models. These agents can solubilize membrane lipids, affecting the barrier integrity. Several research groups have tried to study as culture conditions affect the cell phenotype and barrier properties, but the development of hair follicles and blood vessels in these models is still a long way off. On the other hand, this learning will allow a faster development of unhealthy/injured skin models.

Human and animal tissues still remain as the most representative skin models for drug absorption studies, particularly when new molecules are investigated or for mechanistic purposes (e.g., study the transport route of the compound through the skin, action of absorption chemical enhancers). Among the animal models, porcine skin should be prioritized due to a similar composition and permeability to human skin. Synthetic skin models would be useful for comparative tests between different formulations containing molecules where transcellular transport is predominant.

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## 6. DISCUSSÃO GERAL

Como já mencionado no Capítulo I, a TST é o hormônio responsável pelo desenvolvimento e manutenção das características masculinas. No decorrer da vida, e principalmente após os 40 anos de idade, a produção da TST começa a diminuir. Isso pode estar relacionado com uma produção inefetiva do hormônio pelos testículos ou também por uma deficiência na produção dos precursores hormonais. Essa condição é conhecida como hipogonadismo e está associada a diversos sintomas clínicos, sexuais e psicológicos. Para reduzir ou evitar efeitos indesejáveis da redução hormonal, diferentes formulações a base de TST têm sido desenvolvidas a fim de restaurar os seus níveis fisiológicos, incluindo diferentes rotas de administração (MILLS; MAGNUSSON; CROSS, 2006). Os sistemas transdérmicos têm sido priorizados devido a maior adesão do paciente ao tratamento como são não invasivos e indolores (BAERT et al., 2012).

Embora as formulações transdérmicas sejam mais convenientes para o paciente, a TST apresenta limitações biofarmacêuticas que desafiam o desenvolvimento de novos sistemas. A TST é um hormônio relativamente lipofílico, com uma massa molar e log P intermediário, o que contribui para uma alta afinidade pela pele. A sua afinidade ao tecido também ocorre devido à presença de receptores androgênicos no tecido cutâneo (CHEN; THIBOUTOT; ZOUBOULIS, 2002). Dentre as estratégias utilizadas para promover uma entrega de TST em concentrações adequadas, destaca-se o uso de promotores de absorção. Estas substâncias agem modificando a estrutura da pele, seja nos domínios lipídicos ou proteicos, reduzindo, assim, as propriedades de barreira do estrato córneo e facilitando a absorção.

Há um número apreciável de estudos de permeação cutânea com a TST na literatura. A maioria desses estudos selecionou a TST como fármaco modelo para investigar o perfil de permeação de moléculas com características lipofílicas. Nesse sentido, o capítulo II apresentou um artigo de revisão que trata da avaliação do comportamento da TST em ensaios de permeação que utilizaram células de difusão do tipo Franz diante de diferentes condições experimentais.

De forma resumida, a análise dos artigos concluiu que as diferentes etapas de um estudo de permeação com moléculas lipofílicas devem ser realizadas de forma criteriosa. Por exemplo, para o desenvolvimento de estudos de permeação com a TST, a etapa prévia de avaliação da solubilidade garante que o fármaco esteja solúvel tanto nos meios receptor quanto no doador e não haja subestimação do coeficiente de permeação. Nesse contexto, a utilização de agentes solubilizantes ou cossolventes se torna necessário sempre que o fármaco apresentar uma baixa solubilidade em água. Nos estudos analisados, os cossolventes mais utilizados foram o etanol e o propilenoglicol, os quais também podem atuar como promotores de permeação. Este estudo de revisão também ressalta que deve ser estabelecida uma concentração máxima de solvente para não alterar a integridade do tecido e, desta forma, se observar um aumento de permeação inespecífico (dados de fluxo e coeficiente de permeação superestimados).

A seleção da fase doadora mostrou impactar significativamente no perfil de permeação, uma vez que se observou uma redução da taxa de permeação da TST com o aumento da sua solubilidade nesta fase. Dessa forma, os estudos que utilizaram o propilenoglicol (no qual a TST é menos solúvel) apresentaram uma maior taxa de absorção da TST comparado ao etanol. Além disso, a condição de supersaturação nem sempre vai ser favorável nos estudos de permeação e deve ser avaliada com cautela pelos pesquisadores. Em alguns casos pode ocorrer a precipitação do fármaco no meio supersaturado, sendo necessária a inclusão de agentes que inibem a precipitação. Outra importante análise dessa revisão traz que o uso da pele inteira pode subestimar a quantidade de fármaco efetivamente permeada. Os estudos em que se utilizou uma espessura de aproximadamente 400 µm resultou em maiores taxas de permeação comparativamente aos estudos com as peles inteiras. Isso sugere que os experimentos com fármacos lipofílicos devem ser realizadas preferencialmente com pele dermatomizada para evitar retenção na camada da derme.

Outro parâmetro analisado foi o modelo de membrana. A taxa de permeação variou entre os estudos que usaram diferentes tipos de pele animal, destacando uma grande diferença entre as peles de roedores e a humana para vários tipos de moléculas lipofílicas e hidrofílicas. Ainda, a baixa taxa de difusão da TST através de pele humana e suína levou os pesquisadores a estender o tempo experimental, sendo que a maior parte dos experimentos foi realizada em 24 horas ou em um intervalo ainda maior quando esses tecidos foram usados. Assim, um maior tempo experimental deve ser considerado quando do uso destes tecidos, porém, os pesquisadores devem estar atentos aos efeitos dos solventes e outras condições que afetam a integridade dos tecidos. Por fim, observou-se que a maior parte dos estudos utilizou a cromatografia líquida de alta

eficiência para a quantificação da TST em diferentes matrizes biológicas. Acetonitrila e o metanol foram os solventes mais empregados para romper a ligação do hormônio com as proteínas e precipitação delas. Poucos estudos analisados apresentam dados de retenção cutânea da TST. Ainda assim, foi possível identificar que a TST fica retida em maior concentração no estrato córneo e na epiderme (são mais lipofílicos) do que na derme, respeitando a natureza mais lipofílica da molécula. É importante mencionar que a presença de receptores androgênicos nas glândulas sebáceas, na região dos folículos e na papila dérmica dos queratinócitos pode favorecer a sua retenção, porém, há uma carência de estudos de avaliação da retenção cutânea da TST e seu consequente efeito nas camadas da pele.

Considerando esta lacuna, o capítulo III avaliou o perfil de permeação e retenção da TST nas camadas de pele humana e investigou o efeito da presença de diferentes terpenos (α-bisabolol, carvacrol e mentol) na promoção da permeação. Estes agentes atuam extraindo ou fluidizando os lipídeos, enfraquecendo a barreira do estrato córneo. Por terem origem natural, apresentam baixa toxicidade, além de possuírem atividades secundárias interessantes para tratamentos de ação local como ação anti-inflamatória e antimicrobiana.

A primeira etapa do estudo determinou o tempo necessário para realizar o experimento, visto que a TST possui baixa taxa de permeação. A TST foi detectada em nível quantificável apenas após 16 horas. Com relação à retenção cutânea, houve um aumento de 2 a 3 vezes na quantidade de hormônio nas camadas da pele quando os ensaios passaram de 24 para 36 horas. Nesse sentido, o tempo de 36 horas foi o mais adequado, pois permitiu a construção de um perfil de permeação e assim a determinação dos parâmetros de permeação, como o fluxo, coeficiente de permeação, tempo de latência, bem como a quantificação da TST nas camadas da pele. Na segunda etapa do estudo, os dados fornecidos sugeriram que os terpenos melhoraram a permeação da TST quando comparados ao controle. Dentre os terpenos testados, o carvacrol foi o que resultou em um maior aumento de fluxo e coeficiente de permeação. Embora os outros tratamentos também tenham sido efetivos, essa diferença não foi estatisticamente significante. O carvacrol também foi o único terpeno que reduziu o tempo de latência comparado ao controle, sugerindo um início mais rápido da atividade terapêutica na presença desse agente. Além disso, o carvacrol promoveu uma maior retenção da TST

na camada mais superficial da epiderme, corroborando com os resultados de maior permeação.

Na sequência, análises de FT-IR e DSC foram realizadas para identificar os principais mecanismos de promoção da permeação dos terpenos. Como a TST possui preferência pela rota intercelular de transporte devido a sua natureza lipofílica, promotores que alterem mais os domínios lipídicos são mais efetivos no aumento do fluxo de permeação. A análise no FT-IR identificou desordens em domínios lipídicos do estrato córneo. O carvacrol foi o terpeno que promoveu alterações mais significativas no estrato córneo (maior desordem nas cadeias lipídicas), o que está de acordo com a maior permeação e retenção da TST na presença deste agente. Nos ensaios de DSC, as mudanças térmicas mais importantes também apontaram para uma maior extração de lipídios do estrato córneo pelo carvacrol, especialmente próximo a 65°C. O mentol e carvacrol apresentaram pequenas reduções em temperatura próxima a 80°C, sugerindo formação de complexos entre proteínas e lipídeos e, com isso, maior rompimento da bicamada lipídica. As amostras com carvacrol e o α -bisabolol mostraram maior redução endotérmica que o mentol nessa região, indicando maior fluidização. Ainda, cabe ressaltar que acima de 90°C inicia-se um processo de desnaturação das proteínas. Este evento pode ser detectado em amostras de tecido onde houve extração de lipídios e foi observado com maior intensidade para o α-bisabolol e o carvacrol, o que pode estar relacionado com a extração lipídica.

Por fim, o conjunto dos dados do estudo de permeação e da avaliação mecanística sugere que, embora o carvacrol e o mentol apresentem estruturas químicas e propriedades físico-químicas semelhantes, há diferenças importantes no tipo e intensidade de interações entre estes compostos e as estruturas lipídicas e proteicas da pele. De fato, isso é observado nos diferentes perfis obtidos na permeação da TST. O carvacrol se mostrou mais efetivo no aumento de ambas permeação e retenção da TST por apresentar maior afinidade por domínios lipídicos. Ainda, o carvacrol e o  $\alpha$ -bisabolol mostraram semelhanças no infravermelho, porém, o carvacrol apresentou maior fluxo de permeação. Isso pode ser explicado pelas características físico-químicas desses compostos já que o  $\alpha$ -bisabolol apresenta maior log P, o que resulta em maior interação com as camadas superficiais da pele. Além disso, o  $\alpha$ -bisabolol possui maior massa molar, o que pode reduzir a taxa de difusão do composto através da pele.

Os resultados do capítulo III apontam o carvacrol como o terpeno que gera efeitos mais expressivos na estrutura da pele e, por isso tende a promover maior permeação de compostos lipofílicos. Assim, a incorporação do carvacrol como agente promotor em formulações de uso tópico contendo TST pode ser uma estratégia para aumentar os níveis de TST na circulação sanguínea. Ainda, o carvacrol apresenta atividade antioxidante, pode melhorar a qualidade dos espermatozoides (fator relevante para os pacientes como hipogonadismo), além de suas funções bactericida e antifúngica que podem auxiliar na conservação das formulações. Outro ponto interessante é que esse terpeno não interage com a CYP3A4, não atrapalhando assim a metabolização de xenobióticos e, consequentemente, não ocasionando interações medicamentosas.

O capítulo II apresentou uma análise detalhada dos principais modelos de membranas usadas nos estudos de permeação cutânea da TST, sendo que os modelos animais foram os mais empregados. Dentro desse contexto, o capítulo III apresentou um estudo de permeação com a TST usando pele humana, a qual tem vanta gens inerentes a sua constituição. No entanto, questões éticas e muitas vezes a difícil aquisição da pele humana podem inviabilizar a realização dos experimentos. As membranas sintéticas são uma alternativa ao uso de pele. Além disso, há um esforço contínuo das agências regulatórias para que as pesquisas evitem a utilização de animais. Contudo, embora a utilização das membranas sintéticas e de cultura celular seja reconhecida pela OECD para estudos de toxicologia, ainda continua sendo um desafio a utilização dessas membranas nos ensaios de permeação. Nesse cenário, o capítulo IV apresentou uma revisão atualizada sobre a aplicação de modelos de membranas em estudos de permeação cutânea para moléculas com diferentes propriedades físico-químicas.

O capítulo IV foi baseado na comparação de dados de estudos de permeação usando membranas biológicas, sintéticas ou cultura de células a partir de dois fármacos modelos: a cafeína (CF), um composto hidrofílico e a TST, um composto lipofílico. A CF e a TST são as moléculas mais usadas em estudos de permeação cutânea com novos modelos de membrana a fim de validá-los. A análise desses estudos mostrou que a espessura das membranas biológicas (pele animal ou humana) afetou significativamente a absorção da TST, sendo dez vezes maior nos estudos que utilizaram espessuras menores de pele. No entanto, esse parâmetro não afetou a permeação da CF. Isso confirma a necessidade de um maior rigor com relação à espessura de membranas biológicas para se alcançar resultados mais similares a situação fisiológica, priorizandose tecidos dermatomizados em ensaios com moléculas lipofílicas.

As membranas sintéticas geram resultados mais reprodutíveis que as membranas biológicas, além de permitirem uma maior compreensão da organização do estrato córneo, bem como do impacto de cada componente nas propriedades de barreira. Estas membranas podem ser utilizadas alocando-as em um dispositivo de placas de 96 cavidades chamado PAMPA ou na interface dos compartimentos que compõe as câmaras de difusão. Os estudos apontaram maior correlação entre PAMPA e a pele humana para a TST que CF. A utilização de membranas sintéticas no aparato PAMPA parece ser mais interessante durante a triagem de novos compostos. De modo geral, estudos mostram que as membranas sintéticas podem ser consideradas como modelos de pele preferencialmente nas seguintes situações: quando o estrato córneo é a principal barreira; o composto não sofre metabolização enzimática nem apresenta receptores na pele; quando não há a necessidade de utilização de promotores de permeação que podem interagir com a membrana e danificá-la e quando os ensaios *in vivo* podem ser facilmente executados para validação dos dados.

Com relação às membranas de cultura celular, diferentes tipos de células humanas tem sido cultivadas em camadas sobre uma matriz polimérica. Essas podem ser desenvolvidas visando representar somente a epiderme ou a pele completa. Uma vantagem do uso de células é que os modelos podem expressar enzimas metabólicas e outras estruturas importantes, o que torna o experimento mais próximo a situação fisiológica. No entanto, os estudos mostram que esse modelo é mais permeável que as peles de humano e suína ao apresentarem maior K<sub>p</sub> e menor *lag time*. Essa diferença pode ser decorrente da diferente composição lipídica na membrana de cultura de células em comparação com a membrana biológica em proporção. Outra possível explicação é com relação às diferenças obtidas no empacotamento e organização lipídica do estrato córneo das membranas cultivadas em cultura. Além disso, esse tipo de modelo não é adequado para a avaliação de transporte pelos apêndices, principalmente para moléculas hidrofílicas como a CF e os esteroides como a TST.

O desenvolvimento de estudos de permeação para avaliação do desempenho de moléculas através da pele exige avaliação criteriosa dos diferentes parâmetros envolvidos. De modo geral, cada etapa deve ser analisada com relação ao tipo de fármaco, membrana usada e seleção de fluidos. O capítulo II desmembrou cada etapa e apresentou a importância da seleção adequada desses parâmetros para moléculas com características lipofílicas. Já o capítulo III apresentou um estudo da avaliação de terpenos, que são promotores que impactam preferencialmente em domínios lipídicos da pele humana, e os consequentes efeitos na permeação da TST. A análise mecanística gerou dados que apoiam o carvacrol como um promotor promissor para uso em produtos tópicos que necessitam aumentar a taxa de permeação de fármacos para obtenção de níveis terapêuticos, como o caso da TST. Por fim, a revisão apresentada no capítulo IV analisou as vantagens e limitações de cada modelo de membrana (biológica e não biológica) na permeação de compostos, uma vez que é crescente o incentivo ao não uso de animais. Assim, modelos sintéticos e de cultura celular vem sendo aprimorados na busca de resultados cada vez mais próximos aos estudos in vivo. De modo geral, a seleção da membrana deve considerar as características da molécula do fármaco em análise, além de manter um rigoroso controle de qualidade e especificação de procedimentos para garantir a reprodutibilidade, já que esses ensaios são fundamentais durante o desenvolvimento de novas formulações transdérmicas.

## 7. CONSIDERAÇÕES FINAIS

 Deve-se alcançar um equilíbrio entre solubilidade e permeação para maximizar a taxa de transporte de fármacos pouco solúveis como a TST através da pele;

Uso de etanol aumenta a solubilidade, porém, reduz a taxa de permeação da TST;

 Um grau de supersaturação ideal deve ser encontrado para maximizar a permeação de fármacos e evitar fenômenos de precipitação dos mesmos;

A derme atua como uma barreira reduzindo a partição da TST para o meio receptor;

 Priorizar tecidos dermatomizados em ensaios com fármacos lipofílicos para reduzir a retenção dos mesmos;

 Diferenças substanciais no transporte do hormônio foram encontradas entre as espécies animais analisadas e membranas artificiais;

- Os terpenos se mostraram efetivos no aumento da permeação da TST pela pele;

 Dentre os terpenos analisados, o carvacrol e o mentol foram o mais e menor efetivo como promotores de absorção, respectivamente;

O carvacrol e o α-bisabolol interagiram mais com os domínios lipídicos do que o mentol;

 Como a TST prefere rotas de transporte passivo, uma maior desordem dos domínios lipídicos tem impacto direto na permeação;

 Modelos de pele reconstruída (aqueles que envolvem cultivo celular) são mais permeáveis que pele *in natura*; As membranas sintéticas e de cultura celular não preveem o transporte de ativos pelos apêndices;

 Modelos de pele reconstruída apresentam a vantagem de possibilitar a avaliação metabólica nos ensaios de permeação diferente das membranas sintéticas;

 Embora haja esforços para se obter a mesma composição lipídica entre os modelos de pele reconstruída e as biológicas, ainda há grandes diferenças na organização lipídica dos tecidos;

– Deve-se ter cautela com a utilização de solventes para se alcançar as condições "sink" durante a permeação de moléculas lipofílicas devido à possibilidade de solubilização dos lipídios, afetando as propriedades de barreira das membranas artificiais ou em modelos de pele reconstruída;

 As peles humanas e suínas ainda são os modelos mais representativos para os estudos de permeação.

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